

# Phosphorus speciation by 31P NMR spectroscopy 1 in bracken 2 (Pteridium aquilinum (L.) Kuhn) and bluebell (Hyacinthoides non3 scripta (L.) Chouard ex Rothm.) dominated semi-natural upland soil

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1	Phosphorus speciation by <sup>31</sup> P NMR spectroscopy in bracken
2	(Pteridium aquilinum (L.) Kuhn) and bluebell (Hyacinthoides non-
3	scripta (L.) Chouard ex Rothm.) dominated semi-natural upland soil
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#### 25 Abstract

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Access to P species is a driver for plant community composition based on nutrient 27 acquisition. Here we investigated the distribution and accumulation of soil inorganic P (Pi) 28 29 and organic P (Po) forms in a bracken and bluebell dominated upland soil for the period between bluebell above ground dominance until biomass is formed from half bluebells and 30 half bracken. Chemical characterisation and <sup>31</sup>P Nuclear Magnetic Resonance spectroscopy 31 was used to determine the organic and inorganic P species. Total P concentration in soils was 32 0.87 g kg<sup>-1</sup>, while in plants (above- and below-ground parts) total P ranged between 0.84 - 4.0 33 g kg<sup>-1</sup> and 0.14 - 2.0 g kg<sup>-1</sup> for bluebell and bracken, respectively. The P speciation in the 34 plant samples was reflected in the surrounding soil. The main forms of inorganic P detected 35 in the NaOH-EDTA soil extracts were orthophosphate (20.0 - 31.5 %), pyrophosphate (0.6 -36 2.5 %) and polyphosphate (0.4 - 7.0 %). Phytate (myo-IP<sub>6</sub>) was the most dominant organic P 37 form (23.6 - 40.0 %). Other major peaks were scyllo-IP<sub>6</sub> and  $\alpha$ - and  $\beta$ - glycerophosphate 38 (glyP). In bluebells and bracken the main P form detected was orthophosphate ranging from 39 (21.7 – 80.4 %) and 68.5 - 81.1 %, in above-ground and below-ground biomass, respectively. 40 Other detected forms include  $\alpha$ -glyP (4.5-14.4 %) and  $\beta$ -glyP (0.9 -7.7 %) in bluebell, while 41 in bracken they were detected only in stripe and blade in ranges of 2.5 - 5.5 % and 4.4 - 9.6 42 %, respectively. Pyrophosphate, polyphosphate, scyllo-IP<sub>6</sub>, phosphonates, found in soil 43 44 samples, were not detected in any plant parts. In particular, the high abundance of phytate in the soil and in bluebell bulbs, may be related to a mechanism through which bluebells create 45 46 a recalcitrant phosphorus store which form a key part of their adaptation to nutrient poor conditions. 47

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- 50 Introduction
- 51

Plant growth depends strongly on the macronutrients NPK (Nitrogen-Phosphorus-Potassium), 52 of which P is the least well understood in relation to the distribution and abundance of its 53 different chemical species. Resource partitioning of P in soil dominated by diverse plant 54 communities is critical in order to understand P cycling (Turner, 2008; Zemunik et al., 2015), 55 particularly in view of accessing "legacy P" (Withers et al., 2014) because of predicted 56 shortfalls of fossil P (Reijnders, 2014; Rockström et al., 2009). In soil, P can be present in 57 58 various organic and inorganic forms with orthophosphate being the most readily available for organisms (Shen et al., 2011). The accurate characterisation of P species above and below 59 ground in soil can thus provide useful information on its origins, availability and stability in a 60 given ecosystem and it is thus essential for a better understanding of the ecology of the 61 system under study, particularly in terms of plant accessibility versus loss through leaching 62 that underpins primary productivity. 63

The abundance and distribution of the various P forms is strongly related to the specific 64 65 environmental conditions and soil management practises (Condron and Goh, 1990; Turner et al., 2003a; Turner and Newman, 2005; McDowell and Stewart, 2006; Cade-Menun et al., 66 67 2010; Stutter et al., 2015). In natural and semi-natural systems (i.e. no agricultural inputs), P cycling is closed with little in the way of losses, which are mainly due to leaching and largely 68 69 dependent on factors such as soil parent material, topography, biomass and time (Solomon et 70 al., 2002). In contrast, P imbalance is prominent in agricultural systems due to changes in P 71 input and output reflected in the P speciation. For instance, inorganic orthophosphates usually 72 account for a large proportion of total P in agricultural soils, whereas organic P forms mainly 73 occur in natural or semi-natural soils (McDowell and Stewart, 2006; Stutter et al., 2015).

74 Bracken (Pteridium aquilinum (L.) Kuhn) and British bluebells (Hyacinthoides non-scripta 75 (L.) Chouard ex. Rothm.) are often the dominant species in a late successional ecosystem on deforested Welsh uplands, exposed to extensive grazing pressure (Figure 1). The vegetation 76 classification is U20: Pteridium aquilinum – Galium saxatile community (Rodwell, 1992) 77 which is mostly found in areas of heath or grassland and has been characterised as species 78 poor (Grime et al., 1988). The field site forms part of the Manod association that covers 5372 79 km<sup>2</sup> in England and Wales and characterises loamy soils above 200 m AOD and annual 80 81 rainfall of more than 1000 mm (Cranfield University, 2015). Extensive bluebell coverage is taken as an indicator of ancient woodlands (Rose, 1999). Equally, bracken has been described 82 as "a plant of woodland origin, of moderate shade". It often marks the sites of woods, which 83

have been destroyed, but when it is freed in the open, it can become a "pestilent weed" (cited in Marrs and Watt, 2006). The co-existence of both plant species is favoured by their different growth strategies with bluebells actively growing during winter and spring and bracken emerging above-ground mid to late spring and senescing in autumn. Dense bluebell and bracken populations are often associated with well-drained loamy soil (Knight, 1964) and both plants are present in all parts of the United Kingdom (for general ecological description on bluebells Blackman and Rutter, 1954 and on bracken Marrs and Watt, 2006).

The acquisition of P by bluebell and bracken below ground may support their dominance. 91 92 Previous studies by Merryweather and Fitter (1995b) have showed an initial P inflow for bluebells during the subterranean phase, when roots are actively growing and colonised by 93 arbuscular endophytes; while at the end of the growth phase P was lost through seeds and 94 senesced biomass. Blackman and Rutter (1947, 1948 and 1949) instead found no significant 95 increase in leaf weight on P addition and no significant interaction with light intensity. Seed 96 production was only increased in response to added P and K, and flower production was 97 enhanced only through increased light intensity. Overall Blackman and Rutter (1950) reached 98 the conclusion that "the bluebell does not require a high level of mineral nutrients". 99

100 Co-existence with bracken is attributed to their different phenologies, the reduction of 101 competitive species by bracken and absence of heavy grazing. The acquisition of P by 102 bracken has been less well studied. Mitchell (1973) reported that bracken acquires 103 phosphorus through mobilisation from inorganic sources. However, her study utilised ground 104 bracken rhizome only.

105 Some P forms are routinely measured on an operational basis by solution-based techniques (available and residual pools). These methods include sequential extraction schemes (Chang 106 107 and Jackson 1957; Hedley et al., 1982; He et al., 2003), which provide valuable information on P lability and solubility in soil (Hedley et al., 1982), but it is often time consuming and 108 109 tends to over or underestimate the P form in a specific defined fraction (Turner et al., 2005). For instance, the classification of organic P bioavailability based on chemical solubility can 110 be misleading as recent studies have suggested that plants can access the supposedly 111 "unextractable" fractions of soil organic P (Chen et al., 2002; Turner, 2008). 112

<sup>31</sup>P Nuclear Magnetic Resonance spectroscopy (NMR) is one of the most promising analytical tools which allows the identification of inorganic P forms (i.e. orthophosphate, pyrophosphate and polyphosphate) and most of the organic P forms (i.e. orthophosphate monoesters and diesters and phosphonates) simultaneously (Cade-Menun et al., 2010;Turner et al., 2005). The technique has already been used for the characterisation of P forms in soil

samples and for the evaluation of the effects of different soil types, farming practises and land 118 use on the distribution and transformations of P forms in soils (Cade-Menun and Preston, 119 1996; McDowell et al., 2005; McDowell and Stewart, 2006; Cade-Menun and Liu, 2014; 120 Stutter et al., 2015). However, while there is an increasing number of publications on soils 121 under native vegetation (McDowell and Stewart, 2006; McDowell et al., 2007; Turner et al., 122 2007; Stutter et al., 2015), fewer have investigated soils under native vegetation and the 123 related contribution of the most dominant plant species to the soil P pools using <sup>31</sup>P NMR. In 124 particular, to our knowledge, no studies have reported the characterisation of P species found 125 in a bracken and bluebell native vegetation and soil system; and only a few studies have 126 reported the use of <sup>31</sup>P NMR for the determination of P species in various plant parts 127 (Makarov et al., 2002; Makarov et al., 2005; Noack et al., 2012). 128

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This study thus investigated the P species in soil and plants from a natural vegetation system dominated by bracken and bluebell using <sup>31</sup>P NMR and established assays targeting labile P species (Mehlich-3 extractable) when both plants show active above ground growth. The aim was to probe the mechanisms regulating the composition and nature of P forms in this type of soil and vegetation community.

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### 136 Materials and methods

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138 Sample collection and preliminary analysis

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The soil and plant samples used in this study were collected in an area located at 250 m above 140 141 sea level in the Snowdonia National Park (Llanberis, United Kingdom, N53°07' W 04°08'). The whole area encompasses about half a hectare with full bracken (*Pteridium aquilinum* (L.) 142 Kuhn) and bluebell (Hyacinthoides non-scripta (L.) Chouard ex Rothm) coverage (Figure 1 143 and Supplementary Information (SI)). Most of the root systems of both plants, rhizome for 144 bracken and bulbs for bluebells, grow intertwined. Bracken rhizomes form a dense 10 cm 145 thick layer located approximately 5 to 10 cm below the soil surface. Bluebells propagate 146 predominantly from seeds with small bulblets forming in the first growth period. As perennial 147 plants, the bulb increases in size every year and the roots are contractile. Hence, bluebell 148 bulbs extend downwards during active growth. Young bulbs are located above the bracken 149 rhizome and with increasing age they find their way through the rhizome layer. Mature bulbs 150 are found below the rhizome layer. Above and below-ground biomass of other plant species 151

accounted for less than 5 % of the total biomass. The area falls under the upland vegetation type U20a (*Pteridium aquilinum-Gallium saxatile* community U20, *Anthoxanthum odoratum* sub-community U20a) (Rodwell, 1992) with well-drained and infertile soils. No history of fertiliser application on these fields was reported and no grazing has been applied as a management regime, with the surface litter mostly dominated by dead bracken fronds. The site is hence classed as semi-natural.

A stratified random sampling approach was undertaken, because of the near-level surface of 158 the field. Soil samples (0-15 cm) were collected proportionally around segments with high 159 160 density of both plants, based on previous growth history and visual inspection. A 0-15 cm sampling depth was chosen instead of the recommended 0 -7 cm for undisturbed soil, because 161 bluebell bulbs on the field site grow in colonies occurring at depths between 5 to 20 cm. Due 162 to the heavy nature of the soil (68 % slit), bulbs usually occur between the Ah horizon and 163 the upper Bs horizon (Grabham and Packham, 1983; Merryweather and Fitter, 1995a). The 164 rational was that since no form of agricultural land use (i.e ploughing) was reported for the 165 field site, a depth of 0-15 cm was sufficient in estimating the soil nutritional properties. That 166 was based on the assumption that there was very little variability (less than 15 %) on the 167 field. We however, acknowledge that a small amount of error may have occurred due to 168 169 incomplete sampling of some horizons (less than 15cm), because of the nature of the soil parent material, which consists of mainly metamorphic rock deposits of dark purple slate. 170

Soil and plant sampling was carried out using a 15-cm soil auger (Eijkelkamp, Holland). Two
soil cores were collected weekly during the main growing stage of the plants, from 7<sup>th</sup> May
2013 (week 1) to 25<sup>th</sup> June 2013 (week 8), a total of 8 samplings (W1-W8).

A total of 16 soil cores (two per week) were collected and processed in the laboratory by 174 175 hand. Soil samples were air dried, ground in a porcelain mortar, passed through a 2 mm sieve and combined to form a composite sample for each week of collection (W<sub>s</sub>1- W<sub>s</sub>8). Plants 176 177 were separated from the soil, thoroughly cleaned from any soil remains and further separated into below and above-ground parts. In particular, for bluebells, the below-ground part 178 included bulbs and roots and the above- ground included scapes, leaves and flowers; while 179 for bracken, rhizome was separated from the frond (stipes and blades). These parts were 180 181 freeze-dried for one week and the percentage of the total dry weight was calculated. Each plant part was then ground in a porcelain mortar and stored at 4 °C until further analysis. 182

183 The following chemical and physical parameters were determined on the composite soil 184 samples  $W_s1$ -  $W_s8$ : *i*) Particle size analysis was carried out using a particle size analyzer 185 (Malvern Mastersizer 2000, UK) and the soil texture was classified using the USDA triangle.

- *ii*) Soil pH was measured in H<sub>2</sub>O (ratio soil: water 1:2.5 w/v) using an Orion 420A pH meter
   (Boston, USA). *iii*) Soil organic matter (OM) was determined by loss of ignition at 450 °C
- 188 (4hrs) in a muffle furnace (Carbolite, UK) after oven drying at 110°C. Total carbon (C) and
- 189 nitrogen (N) in soil, were determined on a LECO Truspec, CN Analyzer.
- 190 Total P, aluminium (Al), iron (Fe) and calcium (Ca) were determined in soil and plant (P
- 191 only) samples, using a nitric acid (HNO<sub>3</sub>) digestion method. Soil extractable nutrients (P, Al,

and Fe) were also determined by a Mehlich-3 extraction (Mehlich, 1984). The resulting
solutions were analysed via an ICP-AES Varian 710ES (Agilent Technologies, USA).

- 194 Soil total organic P was determined using the ignition method of Saunders and Williams
- (1955). Ignited and unignited extracts were determined based on the colorimetric method ofMurphy and Riley, (1962).
- 197 Calculation of the Phosphorus Saturation Ratio (PSR) was performed using the formula

$$PSR = \frac{P}{Fe + Al}$$

198

199 where P, Fe and Al concentrations were determined by Mehlich -3 extraction

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201 Pearson pair-wise correlations between sets of data was performed using the statistical 202 Package IBM SPSS Statistics (version 22.0) with significance set at p < 0.05.

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<sup>31</sup>P NMR: Sample preparation and analysis

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The sample preparation for solution <sup>31</sup>P NMR Nuclear Magnetic Resonance spectroscopy
was performed using a modified version of the Cade-Menun and Preston (1996) procedure.

Three grams of air dried soil or 1 g of crushed freeze-dried plant sample was mixed with 25 mL of a solution of 0.25 M NaOH and 0.05 M Na<sub>2</sub>EDTA and shaken at 250 rpm and 20 °C for 6 hrs (soil) or 4 hrs (plant) hours. The extracts were then centrifuged for 20 min at 5000 rpm and filtered using Whatman No.42 filter paper. An aliquot of 0.5 mL was then diluted for ICP-AES analysis and the remaining solution was freeze-dried. The efficiency of P extraction

- had a mean value of 74 % for soil sample and 91 % for plant sample (individual data are
- given in Table 1 and 2).
- Approximately 100 mg of each freeze dried extract was redissolved in 1 mL of  $D_2O$ , 0.6 mL
- 216 10 M NaOH and 0.4 mL extracting solution (0.25M NaOH + 0.05M Na<sub>2</sub>EDTA) (Cade-
- 217 Menun and Liu, 2014). A post extraction step was carried out only for soil samples as

described by Verstergren et al., 2012: An excess of sodium sulphide (Na<sub>2</sub>S) was added to the redissolved sample to ensure precipitation of some of the metals. The solution was then allowed to stand for 2 hours. Samples were then centrifuged for 40 minutes at 5000 rpm (to remove particles that might contribute to line broadening), transferred to a 5 mm NMR tube and analysed via <sup>31</sup>P NMR spectroscopy. A comparison of spectral quality between the addition of sodium sulphide and no addition show improved spectroscopic resolution (Figure S2 SI) and hence reduced scan time.

Spectra were acquired on a Bruker Advance DRX 400 MHz NMR spectrometer (7.5T, 161.9 225 MHz), equipped with a 5 mm broadband probe at 20 °C. Instrument parameters were a 90° 226 pulse, 0.68 s acquisition time and recovery delay of 4.32s to 15s and inverse gated proton 227 decoupling (waltz 16) were used, and set to at least five times the  $T_1$  (lattice relaxation time) 228 based on the P / (Fe + Mn) mass ratios. The experiments required between 1000 and 2500 229 scans (1-2 h running time) for plant and 4000 to 5000 scans (6-7 h running time) for soil 230 samples to achieve a good signal to noise ratio. The spectral width used was 8090.6 Hz and 231 the number of data points was 11002. A delay time of between 3 to 5 seconds has previously 232 been reported to be sufficient to obtain quantitative spectra of NaOH-EDTA in similar soil 233 extracts (McDowell et al., 2006, Stutter et al., 2015). The chemical shift (ppm) of the signals 234 235 was indirectly referenced to an external 85 % H<sub>3</sub>PO<sub>4</sub> standard via the lock signal. Peaks were defined by three parameters: chemical shift, line width and peak height. Peak assignment was 236 237 based on soil and plant extracts spiked with standard solutions and by comparisons to literature data (Turner et al., 2003b, 2003c; Makarov, 2005; McDowell et al., 2005; Smernik 238 239 and Dougherty, 2007; Doolette et al., 2009; Cade-Menun et al., 2010; Cade-Menun ,2015). Spiked solutions were used for the identification of phytate (*myo*-IP<sub>6</sub>),  $\alpha$  and  $\beta$ 240 glycerophosphate and adenosine 5-mono phosphate peaks. Soil or plant extracts were spiked 241 either with 0.1 mL of a 2.1 g  $L^{-1}$  aqueous phytate solution (Na salt hydrate from rice, Sigma 242 Aldrich P8810) or with 0.1 mL aqueous solutions of 4.0 g  $L^{-1}$  of an isomeric mixture of  $\alpha$ 243 and  $\beta$  (1:1) glycerophosphate disodium salt hydrate (Sigma Aldrich G6501). Soil extracts 244 were also spiked with 0.1 mL of a 4.4 g  $L^{-1}$  of adenosine-5-monophosphate disodium salt 245 (Sigma Aldrich 01930). 246

Integration of peak areas were calculated on spectra processed with a line broadening of 1-3
Hz using a Bruker Topspin 2.0 software and MestReNova v.6.0. Quantification of P species
was done by spectra deconvolution analysis, which proved to be successful in particular for
areas such as the monoester region containing a number of peaks, sometimes overlapping; the

relative P concentration in the NaOH-EDTA extracts was estimated on the based on the total
NMR signal area and presented as percentages of each species. If specific identification could
not be made, they were grouped into compounds or compound classes (Cade-Menun, et al.,

- 254 2010, Doolette et al., 2009).
- 255
- 256 **Results**
- 257

258 Soil and plant chemical characteristics

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The soil, a brown podzolic soil termed Manod Association (Cranfield University, 2015), was classified as silt-loam soil (sand: 24 %, silt: 70 %, clay: 6 %). All collected soil samples showed an acidic pH in water, between 4.0 and 4.7. Soil organic matter content (OM %) ranged from 21.2 % - 37.1 %; total C and N content ranged from 10.8 % - 20.2 % and 0.65 % 264 | - 0.95 %, respectively with no major changes in the C/N ratio (19.4 mean value) (Table 1)

Total P in soil was between 0.70 - 1.1 g kg<sup>-1</sup> (mean value 0.87 g kg<sup>-1</sup>) of which between 64 % 265 - 98 % (mean value 77 %) was organic P (Po) (Table 1). Of the total metals analysed, Fe and 266 Al ranged from  $18.5 - 22.8 \text{ g kg}^{-1}$  (mean value 22.1 g kg<sup>-1</sup>) and  $11.6 - 16.7 \text{ g kg}^{-1}$  (mean 267 value 14.1 g kg<sup>-1</sup>), respectively, while Ca ranged from 0.29 - 0.70 g kg<sup>-1</sup> (mean value 0.47 g 268 kg<sup>-1</sup>). Low concentration of Ca were also reflected into the low pH value (mean value 4.5) of 269 the soil. Mehlich 3-extractable P ranged from  $21.4 - 48.7 \text{ mg kg}^{-1}$  (mean value  $32.8 \text{ mg kg}^{-1}$ ) 270 and was negatively correlated with total P ( $r = -0.74 \ p < 0.01$ ), but was strongly positively 271 correlated with the Phosphorus Saturation Ration (PSR) (r = 0.97 p < 0.01). 272

The total C to P and C to organic P (Po) ratios are given in Table 1 and showed a highly significant positive correlation (r = 0.85, p < 0.01), while the N/P to C/P ratio and the N/P ratio to the C/Po ratio showed significant correlations (r = 0.74, p < 0.05, r = 0.81, p < 0.05, respectively). None of the correlations for total N with total C/P, C/Po, total P or Mehlich 3extractable P were significant. (Table S1a in SI)

278 Total P in plant samples was higher in the above ground part for both bluebells and bracken

(Table 2) with total P ranging between  $0.84 - 4.0 \text{ g kg}^{-1}$  and  $0.14 - 2.8 \text{ g kg}^{-1}$ , respectively. In

- 280 particular, the P concentrations were in the order flowers>leaves>scapes>roots>bulbs for
- 281 bluebells and blade>stipe>rhizome for bracken.
- During the sampling period bluebell leaves were dominating above-ground for weeks 1 to 3 (Figure 2a). Peak flowering started in week 4 and a bluebell flower carpet was domineering during weeks 4 and 5 with few bracken fronds emerging. These weeks showed biomass

285 accumulation solely occurring for bluebells whose most active photosynthetic phase was occurring between weeks 1 to 5. At week 6 bluebell flowers started to fade and seed capsules 286 started to form while bracken frond density increased. At week 7 bracken shoots were higher 287 than fading bluebell flowers. From week 8 onwards bracken was the visually dominant plant 288 above ground on the site and the bluebell flowers have turned into seed capsules (photo 289 record is available in SI). As shown in Figure 2b and 2c, the below ground processes 290 contributed a constant 40 % to the total biomass allocation. Until week 4, active 291 photosynthesis contributed to bluebell biomass gains, while bracken biomass stayed constant 292 293 up to week 6.

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295 Phosphorus forms in soil

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Solution <sup>31</sup>P NMR results showed the presence of the same P species in all soil samples (Figure 3a and SI). Mean extraction efficiency of total P in the NaOH-EDTA extract was 74 %, and was negatively correlated to pH (r = -0.73, p < 0.05).

Detected inorganic P compound classes accounted for a total amount of 30 - 41 % including 300 orthophosphate between 5.95 ppm and 6.11 ppm in the range of  $105.1 - 131.0 \text{ mg kg}^{-1}$  (20.0 – 301 31.5 %), pyrophosphate at -3.75 ppm in the range 3.2 - 17.6 mg kg<sup>-1</sup> (0.6 - 2.5 %) and 302 polyphosphates at -3.56 ppm ranging from 2.1- 48.6 mg kg<sup>-1</sup> (0.4 - 7.0 %). Organic P 303 compound classes (59 - 70 % of total NaOH-EDTA extractable P) included phosphonolipids 304 (18.0 ppm) between 4.2 - 21.9 mg kg<sup>-1</sup> (0.8 % - 3.6 %) and phosphonates (20.1 ppm) ranging 305 from 3.0 - 10.6 mg kg<sup>-1</sup> (0.5 - 2.0 %). The orthophosphate diesters were divided into 306 deoxyribonucleic acid (DNA) at -1.0 ppm and other diesters from 2.1 to -2.6 ppm. In the 307 orthophosphate monoester region (2.9 - 5.7 ppm), the four peaks for phytate (*myo*-IP<sub>6</sub>) at 308 5.27 ppm, 4.38 ppm, 3.98 ppm and 3.84 ppm were identified. Other major peaks detected in 309 this region were *scyllo*-IP<sub>6</sub> at 3.7 ppm,  $\alpha$ - and  $\beta$ -glycerophosphate ( $\alpha$ -glyp and  $\beta$ -glyp, 310 respectively), that are phospholipid degradation products, and adenosine-5-monophosphate 311 (AMP). *myo*-IP<sub>6</sub> was confirmed after spiking, while degradation products;  $\alpha$ -glyp,  $\beta$ -glyp and 312 AMP were also identified (Figure 3b). Other unidentified monoesters between 2.9 ppm and 313 5.7 ppm were grouped as other monoesters (Table 3a-b). From our results, NMR-based Po 314 speciation (average Po = 71 %) was in line with the ignition method of Saunders and 315 Williams (1955) (average Po=77 %, in Table 1) and showed a significant correlation with r =316 0.73. 317

318 Correlation coefficients for soil physico-chemical properties and P species determined in the NaOH- EDTA soil extracts are shown in Table S1 in SI. Focussing mainly on the significant 319 correlations for C, N and extractable P, total NaOH-EDTA P was positively correlated with 320 total P (r = 0.84, p < 0.01), total C (r = 0.82, p < 0.05), total N (r = 0.78, P < 0.05) and C/N 321 ratio (r = 0.72, p < 0.05). It was however, negatively correlated with and Mehlich-3 322 extractable P (r = -0.61). For the P species quantified using <sup>31</sup>P NMR, inorganic 323 orthophosphate concentration was positively correlated with total C (r = 83, p < 0.05), total N 324 (r = 0.72, p < 0.05), C/N ratio (r = 0.82, p < 0.05), total P (r = 0.74, p < 0.05) and total 325 NaOH-EDTA P (r = 0.88, p < 0.01). Polyphosphate on the other hand, was strongly 326 negatively correlated with pH (r = -0.85, p < 0.01) and strongly positively correlated with 327 Mehlich-3 extractable Fe (r = 0.77, p < 0.05). Orthophosphate monoesters were the most 328 dominant group of Po compounds in the field. Their concentration as whole (sum of all 329 detected monoesters) was negatively correlated Mehlich-3 extractable P (r = -0.63, p < 0.05) 330 and strongly positively correlated with total P (r = 0.79, p < 0.05) and total NaOH-EDTA P (r331 = 0.84, p < 0.01). The most dominant orthophosphate monoester myo-IP<sub>6</sub>, did not show any 332 significant correlation with most of the soil physio-chemical properties. However, it was 333 strongly negatively correlated with N/Po ratio (r = -0.80 p < 0.05). 334

335

336 Phosphorus forms in plants

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Figure 4a shows solution <sup>31</sup>P NMR results for all plant parts. The main P form in bluebells 338 was orthophosphate (5.34 – 5.76 ppm), found in the range 302 -2573 mg kg<sup>-1</sup> (21.7 – 80.4 339 %). Orthophosphate decreased in absolute and relative amounts from leaves > scapes > 340 flowers > roots > bulbs > seeds. *myo*-IP<sub>6</sub> (5.08 ppm, 4.18ppm, 3.81 ppm and 3.79 ppm) was 341 the major P form in bluebell seeds 1939 mg kg<sup>-1</sup> (60%) and bulbs 283.7 mg kg<sup>-1</sup> (39.4 %). 342 The other species detected in all bluebell plant parts were phospholipid degradation products 343  $\alpha$ -glyp (67.7- 347.2 mg kg<sup>-1</sup>, 4.5-14.4 %) and  $\beta$ -glyp (22.5-130 mg kg<sup>-1</sup>, 0.9 – 7.7 %) detected 344 at 4.45 ppm and 4.10 ppm respectively. Ribonucleic acid derived AMP (4.02 ppm) was in the 345 range 39.6 - 140.8 mg kg<sup>-1</sup> (1.4 - 6.4 %), but absent in bulbs. *myo*-IP<sub>6</sub>,  $\alpha$ -glyp and  $\beta$ -glyp and 346 AMP were confirmed after spiking (Figure 4b). Deoxyribonucleic acid was only detected in 347 bluebell flowers (55.8 mg kg<sup>-1</sup>, 1.8 %). Other monoesters, likely to include sugar phosphates, 348 and lower inositol phosphates were between 90.5 - 350.3 mg kg<sup>-1</sup> (2.8 % - 11.3 %) and were 349 not detected in bulbs. Other diester P forms, e.g. non-hydrolysed phospholipids, were in the 350 range 32.5 - 73.6 mg kg<sup>-1</sup> (0.5 - 14.3 %). 351

The main P form detected in bracken was also orthophosphate (102.8 - 2189.7 mg kg<sup>-1</sup>, 68.5 -352 81.1%, blade > stipe >rhizome), followed by monoester P forms (47.3 - 237.6 mg kg<sup>-1</sup>, 8.8 -353 31.5 %, rhizome > stipe > blade).  $\alpha$ -glyp and  $\beta$ -glyp were detected only in stipes and blades 354 in ranges (67.5 - 88 mg kg<sup>-1</sup>, 2.5 - 5.5 %) and (118.8 - 153.6 mg kg<sup>-1</sup>, 4.4 - 9.6 %) 355 respectively, with stipes showing higher values. Adenosine-5-monophosphate was similar 356 between stipes and blades (about 2 %) and absent in rhizome. Other possible diester P forms 357 were detected only in bracken blades in very small amounts 29.7 mg kg<sup>-1</sup> (1.1 %). 358 Pyrophosphate, polyphosphate, scyllo-IP<sub>6</sub>, phosphonolipids and phosphonates, which were 359 360 found in soil samples, were not detected in any plant parts (Figure 4).

361

#### 362 **Discussion**

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364 Soil and plant chemical characteristics

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Bluebell and bracken often form dense co-existing communities on acidic, nutrient poor, and 366 well drained (sandy) loamy soils with few other plant species present (Knight, 1964; 367 Grabham and Packham, 1983; Merryweather and Fitter, 1995). Comparably, the field site 368 369 used in this study presented a loamy texture and low pH but high total P content (McDowell and Stewart, 2006), mostly organically bound (Condron and Goh, 1990; Hawkes et al., 1984) 370 371 and hence not directly bioavailable. Thus the limited availability of P could be the limiting factor for plant growth and access to this limited pool could thus contribute to the successful 372 establishment and maintenance of specific species, i.e. bluebells and brackens. Organic P can 373 be an essential component of soil solution pool during periods of P limitation (Shen et al., 374 2011). The positive relationship observed between total P and total NaOH-EDTA P (both 375 consisting of large amounts of Po) suggest that fractions of the soil's organic P is labile and 376 may contribute to the soil solution phase. The period under study described the shift from 377 bluebells dominating, with their period of most active growth and biomass accumulation 378 terminating with the onset of seed ripening in W5, to bracken dominating the above ground 379 growth from W6 onwards. A decline in Mehlich-3 extractable P and an increase in total P 380 from W1 to W5 was observed. The shift from a bluebell dominance culminating in seed 381 setting in W6 and concurring with higher bracken biomass showed a decline in total P and a 382 doubling or Mehlich-3 extractable P in W6. Bracken dominance in W7 and W8 was reflected 383 in a decline in Mehlich-3 extractable P. but an increase in total P and total NaOH-EDTA P 384 when compared to W6 (Figure S3 SI). 385

The Phosphorus Saturation Ratio (PSR), which is a measure of the soil capacity to retain P, 386 gives an estimate of the extent to which potential adsorption sites (Fe and Al) in the soil have 387 been saturated with P. In this study, the PSR (AI + Fe) did not exceeded the environmentally 388 critical PSR limit of 0.15 (Table 1), in fact it was a magnitude smaller indicating the 389 390 limitation of available P with a large capacity to adsorb phosphate, should it become available (Nair, 2014). Amorphous oxides of Fe and Al are influential for P sorption in acidic 391 soils reducing its availability (Turner et al., 2006). The Fe and Al content in our soil indicates 392 the possibility of P being fixed with Fe and Al hydrous oxides or being precipitated as 393 394 insoluble Al and Fe phosphates. In addition, the total C to Po ratio used as an estimate for determining if net mineralization (< 200) or immobilization (> 300) is occurring in soils 395 (Dalal, 1977). For most of the weeks sampled, (Table 1) the values were greater than 300, 396 suggesting that net immobilization of P (imbalanced in the P cycle) was occurring in the soil, 397 favoring the accumulation of P in organic form. 398

- The P content in the different plant parts agreed with previous studies on bluebells (Blackman 399 and Rutter, 1949; Merryweather and Fitter, 1995) and bracken (Ferguson and Armitage, 400 1944; Moon and Pal, 1949). The amount of P varied according to the different plant parts, 401 402 with higher values in the above-ground parts (leaves, scapes, flowers for bluebell, stipes and 403 blades for bracken) and with bluebells showing the highest concentrations. The bluebell species under study is triploid (Grundmann et al., 2010) and has a larger genomic DNA size, 404 405 as it is often found for early spring flowering species (Hendry, 1987). This implies higher demand for P during growth and subsequent higher P concentration in tissue. In general, P in 406 407 plants is preferentially transferred to leaves and flowers where it is needed for photosynthesis, pollen and seed formation (Schachtman et al., 1998; Shen et al., 2011) and the highest P 408 409 content was determined in bluebell flowers and seeds followed by leaves and bracken blades.
- 410

411 Phosphorus forms in soil

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The Pi content of the soil ranged from 20.0 to 31.5 %, this low Pi content seems to be reflected in the nature of the soil, largely controlled by its associations with Al and Fe – oxides. This is shown by the relationship between orthophosphate and Mehlich-3 extractable Al and Fe. However, due to the closer correlation of orthophosphate with Mehlich-3 extractable Al, the formation of Al-P compounds in the soil is highly favoured. The strong relationship observed between orthophosphate with total C, also suggest that its sorption to OM increases as the level of total P in the soil increases. 420 Organic P content comprised a large part of the total P (up to 80 %) in soil mainly consisting of orthophosphate monoesters: myo-IP<sub>6</sub>, followed by its stereoisomers scyllo-IP<sub>6</sub>. They are 421 derived from plant and microbial sources, but may also include NaOH-EDTA extraction 422 degradation products of phospholipids and RNA (Turner et al 2002; Makarov et al., 2002; 423 Makarov et al., 2005; Bünemann et al., 2008; Cade-Menun, 2015). The large charge density 424 of higher orthophosphate monoesters contributes to their strong sorption to soil by metal 425 oxide in preference to orthophosphate. Complexation and precipitation reaction with 426 polyvalent cations inhibit both chemical and enzyme-mediated biological attack (Turner et 427 428 al., 2002). This study showed a closer relationship between Mehlich-3 extractable Al, rather than Fe, and orthophosphate monoesters thus suggesting that Al is more essential for P 429 sorption. These associations can either act as sources of P during periods of limitation, 430 supplying labile forms of P to the solution phase or as soil P sinks. The negative correlation 431 between orthophosphate monoester and pH suggests that the stability of their association may 432 decline with increasing pH. The high C/Po ratio (>333 mean value) supports the formation of 433 OM - Al - myo-IP<sub>6</sub> - complexes, as myo-IP<sub>6</sub> is the most dominant soil P species (159.4 -434 259.1 mg kg<sup>-1</sup>, 39 - 52 % of organic P), with resulting immobilisation of P in the soil. 435

The other inorganic P species detected in soil were pyrophosphate and polyphosphates. The other major monoesters were  $\alpha$ -glyp,  $\beta$ -glyp (phospholipid degradation products) and RNA derived AMP; and the diesters, including nucleic acids (DNA and RNA) and non-hydrolysed phospholipids. The last class of organic P compounds detected were the phosphonates and phosphonolipids found in soil samples only. Their likely origins have been extensively studied by various authors (Makarov et al., 2002; Makarov et al., 2005; Turner et al., 2005; Doolette et al., 2009; Bünemann et al., 2008; Turner, 2008; Cade-Menun et al., 2010).

443

444 Phosphorus forms in plants

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All bracken and actively growing bluebell parts (roots, scapes, leaves and flowers) contained 446 a significant percentage of Pi (60 to 80 %) in the form of orthophosphate 447 i.e.  $H_2PO_4^-$  and  $HPO_4^{2-}$ . This was consistent with previous studies on plant material that 448 found a range of 25 to 75 % Pi. Bluebell seeds and bulbs on the other hand, contained only 449 450 21.7 % and 42.0 % orthophosphate, respectively, which is within the range usually found in seeds (Noack et al., 2012). myo-IP<sub>6</sub> was the most abundant organic P form detected in 451 bluebell bulbs (39.4 %) and seeds (60 %) but not in any other bluebell or all bracken parts. 452 This is consistent with previous work on seeds, which reports that myo-IP<sub>6</sub> represents about 453

454 50 to 80% of seed P. The high myo-IP<sub>6</sub> content in bluebell bulbs is, to the best of our 455 knowledge, the first report of myo-IP<sub>6</sub> in bulbs. Additionally, the bulbs did not contain other 456 monoesters The other inorganic P species detected only in bluebell seeds was pyrophosphate.

457

458 Ecological implications

The bracken and bluebell dominated ecosystem presented in this study could be an example 459 of the co-existence of two species with different nutrient acquisition strategies in relation to P 460 based on their differences in P speciation in plant. This P speciation in the plant biomass is 461 462 linked through litter input with the P speciation in the soil. Overall P returned to soil from the vegetative plant parts (leaves, roots and inflorescences) would contribute mainly to the 463 orthophosphate and diesters fractions, while bluebell bulbs and seeds would predominately 464 contribute to the Po fraction in soil. Abiotic factors such as sorption / desorption, weathering 465 and microbially-mediated P immobilization may change the P speciation in soil away from 466 that originally found in the plant litter. The soils forming part of the Manod Association, 467 however, are described as having a thick surface mat or roots and plant remains when on 468 469 slopes to steep or rocky to cultivate. The comparatively high phytate content in bluebell bulbs 470 thus determines the equally high phytate content in the surrounding soil as during flowering 471 the old bulb is shed. At this stage there is significant loss of P to the surrounding soil. Previous studies have shown that myo-IP<sub>6</sub> additions to soil can led to the release of 472 473 orthophosphate and OM into soil solutions (Anderson et al., 1974; Leytem et al., 2002). From our results, towards the end of flowering, W5 to W7, the bluebell plant also shed its leaves 474 475 and inflorescences, at this point the plants not only losses P, but carbon to the surrounding 476 soil, reflected by the increase in available P and the PSR ratio in weeks 6. This might also be 477 the likely reason for the change in OM content noticed in Weeks 6 and 7.

Bluebells had been shown to store only half their acquired P in the new bulb because of the 478 479 old bulb disappearing. The P speciation reported here support this bulb shedding which results in a net P flux from the dense bluebell population into the surrounding soil 480 (Merryweather and Fitter, 1995b). With the remants of the "old" bulb, a store is created near 481 the growing location which includes essential elements for future growth and P in the form of 482 myo-IP<sub>6</sub>. Access to this stored phytate may be achieved through arbuscular mycorrhiza (AM) 483 on which bluebells are dependent (Merryweather and Fitter, 1995a). The roots of the bluebell 484 plant are usually colonised by AM immediately after emergence. Studies have shown that this 485 AM association can significantly increase the surface area of the bulbs root system, enabling 486 487 it to access deeper layers of the soil profile (Merryweather and Fitter, 1995a, 1995b), thus

488 likely increasing its proximity to substrates (i.e. myo-  $IP_6$ ), which is one of the essential requirement for plants who may likely utilize myo-IP<sub>6</sub>. (Richardson et al., 2006). Since the 489 soil used in the present study was undisturbed, inositol phosphates would likely accumulate 490 on the soil surface through shedded seeds and directly in the soil through the shedded bulb. 491 492 Bluebells have contractile roots through which they are able to migrate to deeper depth in the soil profile (> 20 cm) (Grabham and Packham, 1983). AM increases the root phosphatase 493 activity of their host and also produce an extracellular membrane-bound phytate degrading 494 enzyme in its hyphae, which could aid root phosphatases in the hydrolysis of Po compounds 495 496 (Richardson et al., 2001). The bracken plant, on the other hand, contains a thick root like rhizomes with tiny hair-like black roots forming a vast network located about 10 to 20 cm 497 underground. Bracken has no reported AM association with its rhizome thus, P uptake is 498 based on its rhizomes root phosphatase activity, exudation of acids and microbially-mediated 499 hydrolysis. Exudates from plants root alone are not capable of utilizing P directly from myo-500 IP<sub>6</sub>, due the very low level of extracellular phytase they contain, but are dependent on 501 microbial-mediated (i.e. fungi) hydrolysis. Hence bracken rhizome would probably not be 502 able to utilize *myo*-IP<sub>6</sub> directly unlike plant roots with AM association (Richardson et al., 503 504 2006). This further supports our theory that bluebells are more likely to utilize myo-IP<sub>6</sub> 505 compared to bracken with its non-AM rhizome counterpart. We however, acknowledge that some form of microbial-mediated hydrolysis may also be involved, likely fungi due to the 506 507 acidic nature of the soil. Determination of phospholipid fatty acids (PLFAs) from this bluebell and bracken site showed a higher concentration of fungal biomarkers compared to a 508 509 bracken only site on similar soil (unpublished data).

510 The suppression of bracken crozier emergence during bluebell flowering of the dense 511 population was observed (Figure 1) compared to that of croziers emerging at the same geographic location but with lower bluebell density three weeks earlier (SI). Week 5 showed 512 the highest total P concentration in soils for all sampling occasions combined with a low 513 extraction efficiency, caused by litter input through shed bulbs. Croziers start to emerge 514 during week 5 (see photos in SI). Week 6 showed both a reduction in total P in soil and a 515 predominance of organic P (98%), thought to indicate the assimilation of Pi by bracken roots 516 517 to support above ground growth.

The ecological consequence of the results presented here support a number of observation in relation to the ecology of established bluebell populations and means of spread. Bluebells are taken as indicators of ancient woodlands, where nutrient status is poor and P is mostly stored as phytate (Attiwill and Adams, 1993, Turner et al., 2002). Observations on the field site

included the establishment of bluebell clumps from seed stores, which could supply a 522 preferred source of phytate from ungerminated seeds as their content was 60 % myo-IP<sub>6</sub>. In 523 addition, plant establishment from bulbs is more successful if bulbs are planted close 524 together, which again increases concentration of phytate through the disappearing old bulbs 525 (Merryweather and Fitter 1995b). The retention of phytate in acidic and Al and Fe rich soils, 526 preferred bluebell habitats, is hence a chemical mechanism that supports the long-term 527 maintenance of bluebell populations on natural soils. The bluebell bracken dominated 528 ecosystem is an example where access to resources is determined by both phenology for 529 530 primary production or nutrient acquisition and storage for P.

531

## 532 Conclusions

533

In this study, investigations on the major P forms in a semi-natural upland soil dominated by 534 bluebell and bracken showed that the distribution of the major soil P species was determined 535 by the present vegetation. <sup>31</sup>P NMR spectroscopy showed that there was a dominance of more 536 recalcitrant organic P forms (i.e. monoesters) compared to more readily available inorganic P 537 form (i.e. orthophosphate). In particular, myo-IP<sub>6</sub>, the most dominant monoester form in the 538 539 soil, was also found at similar concentrations in bluebell bulbs, suggesting that annual shedding of the old bulb could be a key contributor to the build-up of residual forms of Po in 540 541 the soil over time.

The data collected during this study suggest that the bracken and bluebell plant community 542 543 was able to dominate for two main reasons: phenology and different nutrient acquisition and storage strategies, as demonstrated by the different P storage in the plants and soil through 544 545 litter input above and below ground. The concentration of P in bluebell above-ground parts was between 2 to 5 times higher compared to bracken. In addition, bluebells store P in form 546 of mvo-IP<sub>6</sub> in bulbs, possibly as a survival mechanism against P supply interruption during its 547 growth cycle. The shed bulbs then might extend the P store outside its physiological limit into 548 the surrounding soil, increasing the resilience for the population. The semi-natural system 549 used for this study suggests an accumulation of organic P over time. We thus conclude that 550 the build-up of soil P in the field is a result of the plants' biomass contributions over time 551 particularly in the form of myo-IP<sub>6</sub>. These findings support observations on bluebell ecology 552 553 in relation to being a woodland plant, or an indicator of ancient woodlands, and often appearing in clumps. 554

555

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# 565 **References**

- Ahlgren, J., Djodjic, F., Börjesson, G., Mattsson, L., 2013. Identification and quantification
   of organic phosphorus forms in soils from fertility experiments. Soil Use Manag. 29,
   24–35. doi:10.1111/sum.12014
- Anderson, G., Williams, E.G., Moir, J.O., 1974. A comparison of the sorption of inorganic
  orthophosphate and inositol hexaphosphate by six acid soils. J. Soil Sci. 25, 51–62.
- 571 Attiwill, P.M., Adams, M.A., 1993. Nutrient cycling in forests. New Phytol. 124, 561–582.
- 572 Bieleski, R.L., 1973. Phosphate pools, phosphate transport, and phosphate availability. Annu.
  573 Rev. Plant Physiol. 24, 225–252.
- 574 Blackman G, Rutter A. 1954. *Endymion nonscriptus* (L.) Garcke. J Ecol, 42:629-38.
- Blackman G, Rutter A. 1950. Physiological and Ecological Studies in the Analysis of Plant
  Environment: V. An Assessment of the Factors controlling the Distribution of the
  Bluebell (*Scilla non-scripta*) in Different Communities. Annals of Botany.14:487-520.
- Blackman G, Rutter A. 1949. Physiological and Ecological Studies in the Analysis of Plant
  Environment: IV. The Interaction between Light Intensity and Mineral Nutrient Supply
  on the Uptake of Nutrients by the Bluebell (*Scilla non-scripta*). Annals of
  Botany.13:453-89.
- Blackman G, Rutter A. 1948. Physiological and ecological studies in the analysis of plant
  environment: III. The interaction between light intensity and mineral nutrient supply in
  leaf development and in the net assimilation rate of the bluebell (*Scilla non-scripta*).
  Annals of Botany.12:1-26.
- Blackman, G.E., Rutter, A.J., 1948. Physiological and ecological studies in the analysis of
  plant environment: III. The interaction between light intensity and mineral nutrient
  supply in leaf development and in the net assimilation rate of the bluebell (*Scilla non-scripta*). Ann. Bot. 12, 1–26.
- Blackman, G.E., Rutter, A.J., 1947. Physiological and Ecological Studies in the Analysis of
  Plant Environment: II. The Interaction between Light Intensity and Mineral Nutrient
  Supply in the Growth and Development of the Bluebell (*Scilla non-scripta*). Ann. Bot.
  11, 125–158.
- 594 Bünemann, E.K., Prusisz, B., Ehlers, K., 2011. Characterization of phosphorus forms in soil

- 595 microorganisms, in: Phosphorus in Action. Springer, 37–57.
- Bünemann, E.K.K., Smernik, R.J.J., Doolette, A.L.L., Marschner, P., Stonor, R., Wakelin, 596 S.A.A., McNeill, A.M.M., 2008. Forms of phosphorus in bacteria and fungi isolated 597 Australian soils. Soil Biol. Biochem. 40, 1908–1915. 598 from two doi:10.1016/j.soilbio.2008.03.017 599
- Cade-Menun BJ. 2015. Improved peak identification in 31 P-NMR spectra of environmental
   samples with a standardized method and peak library. Geoderma. 257:102-14
- Cade-Menun, B.J., 2005. Characterizing phosphorus in environmental and agricultural
   samples by 31P nuclear magnetic resonance spectroscopy. Talanta 66, 359–371.
   doi:10.1016/j.talanta.2004.12.024
- Cade-Menun, B.J., Carter, M.R., James, D.C., Liu, C.W., 2010. Phosphorus forms and 605 606 chemistry in the soil profile under long-term conservation tillage: a phosphorus-31 nuclear magnetic resonance study. J. Environ. Qual. 1647-1656. 607 39, doi:10.2134/jeq2009.0491 608
- Cade-Menun B, Liu CW. 2014. Solution phosphorus-31 nuclear magnetic resonance
   spectroscopy of soils from 2005 to 2013: A review of sample preparation and
   experimental parameters. Soil Sci Soc Am J.78:19-37
- Cade-Menun, B.J., Preston, C.M., 1996. A comparison of soil extraction procedures for 31p
   nmr spectroscopy. Soil Sci. 161, 770–785. doi:10.1097/00010694-199611000-00006
- Caldwell, A.G., Black, C.A., 1958. Inositol hexaphosphate: II. Synthesis by soil
  microorganisms. Soil Sci. Soc. Am. J. 22, 293–296.
- 616 Chang, S.C., Jackson, M.L., 1957. Fractionation of soil phosphorus. Soil Sci. 84, 133–144.
- 617 Chen, C.R., Condron, L.M., Davis, M.R., Sherlock, R.R., 2002. Phosphorus dynamics in the
  618 rhizosphere of perennial ryegrass (*Lolium perenne L.*) and radiata pine (*Pinus radiata D.*619 *Don.*). Soil Biol. Biochem. 34, 487–499.
- Condron, L.M., Goh, K.M., 1990. Nature and availability of residual phosphorus in longterm
   fertilized pasture soils in New Zealand. J. Agric. Sci. 114, 1–9.
- 622 Cranfield University 2015. *The Soils Guide*. Available: www.landis.org.uk. Cranfield
   623 University, UK. Last accessed 24/06/2015
- 624 Dalal R. 1977. Soil organic phosphorus. Adv Agron .29.
- Doolette A, Smernik R, Dougherty W. 2009. Spiking improved solution phosphorus-31 nuclear magnetic resonance identification of soil phosphorus compounds. Soil Sci Soc Am J .73:919-27.
- Ferguson W, Armitage E. 1944. The chemical composition of bracken (*Pteridium aquilinum*). The Journal of Agricultural Science .34:165-71.
- Grabham P, Packham J. 1983. A comparative study of the bluebell *Hyacinthoides non-scripta* (*L.*) *Chouard* in two different woodland situations in the West Midlands, England. Biol
   Conserv.26: 105 -126
- Grime, J.P., Hodgson, J.G., Hunt, R., 1988. Comparative plant ecology: a functional
  approach to common British species. Springer.

- Grundmann, M., Rumsey, F.J., Ansell, S.W., Russell, S.J., Darwin, S.C., Vogel, J.C.,
  Spencer, M., Squirrell, J., Hollingsworth, P.M., Ortiz, S., 2010. Phylogeny and
  taxonomy of the bluebell genus *Hyacinthoides, Asparagaceae* [Hyacinthaceae]. Taxon
  59, 68–82.
- Hawkes G, Powlson D, Randall E, Tate K. 1984. A 31P nuclear magnetic resonance study of
  the phosphorus species in alkali extracts of soils from long-term field experiments. J Soil
  Sci.35:35-45
- He, Z., Honeycutt, C.W., Griffin, T.S., 2003. Comparative investigation of sequentially
  extracted phosphorus fractions in a sandy loam soil and a swine manure. Commun. Soil
  Sci. Plant Anal. 34, 1729–1742.
- Hedley, M.J., Stewart, J.W.B., Chauhan, Bs., 1982. Changes in inorganic and organic soil
  phosphorus fractions induced by cultivation practices and by laboratory incubations.
  Soil Sci. Soc. Am. J. 46, 970–976.
- Hendry, G., 1987. The ecological significance of fructan in a contemporary flora. New
  Phytol. 106, 201–216.
- Knight G.1964. Some factors affecting the distribution of *Endymion nonscriptus* (L.) Garcke
   in Warwickshire woods. The Journal of Ecology.405-21.
- Leytem, A.B., Mikkelsen, R.L., Gilliam, J.W., 2002. Sorption of organic phosphorus
   compounds in Atlantic coastal plain soils. Soil Sci. 167, 652–658.
- Makarov, M.. I., Haumaier, L., Zech, W., 2002. Nature of soil organic phosphorus: An
  assessment of peak assignments in the diester region of 31P NMR spectra. Soil Biol.
  Biochem. 34, 1467–1477. doi:10.1016/S0038-0717(02)00091-3
- Makarov, M.I., Haumaier, L., Zech, W., Marfenina, O.E., Lysak, L. V., 2005. Can 31P NMR
  spectroscopy be used to indicate the origins of soil organic phosphates? Soil Biol.
  Biochem. 37, 15–25. doi:10.1016/j.soilbio.2004.07.022
- Marrs R, Watt A. 2006 .Biological flora of the British isles: *Pteridium aquilinum* (L.) Kuhn. J
  Ecol. 94:1272-321.
- McDowell R, Cade-Menun B, Stewart I. 2007. Organic phosphorus speciation and
   pedogenesis: analysis by solution 31P nuclear magnetic resonance spectroscopy. Eur J
   Soil Sci .58:1348-57.
- McDowell R, Condron L, Stewart I, Cave V. 2005. Chemical nature and diversity of
   phosphorus in New Zealand pasture soils using 31P nuclear magnetic resonance
   spectroscopy and sequential fractionation. Nutr Cycling Agroecosyst .72:241-54.
- McDowell R, Stewart I. 2006 The phosphorus composition of contrasting soils in pastoral,
  native and forest management in Otago, New Zealand: sequential extraction and 31 P
  NMR. Geoderma ,130:176-89.
- McDowell R, Stewart I, Cade-Menun B. 2006. An examination of spin–lattice relaxation
  times for analysis of soil and manure extracts by liquid state phosphorus-31 nuclear
  magnetic resonance spectroscopy. J Environ Qual, 35:293-302.
- Mehlich, A., 1984. Mehlich 3 soil test extractant: A modification of Mehlich 2 extractant.
  Commun. Soil Sci. Plant Anal. 15, 1409–1416.

- Merryweather J, Fitter A. 1995a. Arbuscular mycorrhiza and phosphorus as controlling
  factors in the life history of *Hyacinthoides non-scripta* (L.) Chouard ex Rothm. New
  Phytol, 129:629-36.
- Merryweather J, Fitter A. 1995b. Phosphorus and carbon budgets: mycorrhizal contribution
   in *Hyacinthoides non-scripta* (L.) Chouard ex Rothm. under natural conditions. New
   Phytol, 129:619-27
- Mitchell, J., 1973. Mobilisation of phosphorus by*Pteridium aquilinum*. Plant Soil 38, 489–
  491.
- Moon F, Pal A.1949.The composition and nutritive value of bracken. The Journal of
   Agricultural Science, 39:296-301.
- Murphy, J., Riley, J., 1962. A modified single solution method for the determination of
  phosphate in natural waters. Anal. Chim. Acta 27, 31–36.
- Nair VD. 2014. Soil phosphorus saturation ratio for risk assessment in land use systems.
   Frontiers in Environmental Science 2:6.
- Noack, S.R., McLaughlin, M.J., Smernik, R.J., McBeath, T.M., Armstrong, R.D., 2012. Crop
   residue phosphorus: Speciation and potential bio-availability. Plant Soil 359, 375–385.
   doi:10.1007/s11104-012-1216-5
- Reijnders, L., 2014. Phosphorus resources, their depletion and conservation, a review.
   Resour. Conserv. Recycl. 93, 32–49.
- Richardson, A.E., George, T.S., Jakobsen, I., Simpson, R.J., 2006. 15 Plant Utilization of
   Inositol Phosphates. Inositol phosphates Link. Agric. Environ. 242.
- Rockström, J., Steffen, W., Noone, K., Persson, Å., Chapin, F.S., Lambin, E.F., Lenton,
  T.M., Scheffer, M., Folke, C., Schellnhuber, H.J., 2009. A safe operating space for
  humanity. Nature 461, 472–475.
- Rodwell, J.S., 1992. British Plant Communities Volume 3. Grasslands and montane
   communities. 550 pp. Cambridger, UK.
- Rose, F., 1999. Indicators of ancient woodland. Br. Wildl. 10, 241–251.
- Saunders, W.M.H., Williams, E.G., 1955. Observations on the determination of total organic
   phosphorus in soils. J. Soil Sci. 6, 254–267.
- Schachtman, D.P., Reid, R.J., Ayling, S.M., 1998. Phosphorus Uptake by Plants: From Soil to Cell. Plant Physiol. 116, 447–453.
- Shen, J., Yuan, L., Zhang, J., Li, H., Bai, Z., Chen, X., Zhang, W., Zhang, F., 2011.
  Phosphorus dynamics: from soil to plant. Plant Physiol. 156, 997–1005.
  doi:10.1104/pp.111.175232 [doi]
- Smernik RJ, Dougherty WJ. 2007. Identification of phytate in phosphorus-31 nuclear
   magnetic resonance spectra: The need for spiking. Soil Sci Soc Am J. 71:1045-50.
- Solomon D, Lehman N. 2000. Loss of phosphorus from soil in semi-arid northern Tanzania
   as a result of cropping: evidence from sequential extraction and 31P-NMR spectroscopy.
   Eur J Soil Sci .51:699-708
- 715 Stutter, M.I., Shand, C.A., George, T.S., Blackwell, M.S.A., Dixon, L., Bol, R., MacKay,

- R.L., Richardson, A.E., Condron, L.M., Haygarth, P.M., 2015. Land use and soil factors
  affecting accumulation of phosphorus species in temperate soils. Geoderma 257, 29–39.
- 718 Turner BL. 2008 .Resource partitioning for soil phosphorus: a hypothesis. J Ecol .96:698-702
- Turner, B.L., Cade-Menun, B.J., Condron, L.M., Newman, S., 2005. Extraction of soil organic phosphorus. Talanta 66, 294–306.
- Turner BL, Condron LM, Richardson SJ, Peltzer DA, Allison VJ. 2007. Soil organic
   phosphorus transformations during pedogenesis. Ecosystems.10:1166-81.
- Turner BL, Mahieu N, Condron LM. 2003a. The phosphorus composition of temperate
   pasture soils determined by NaOH–EDTA extraction and solution 31 P NMR
   spectroscopy. Org Geochem 2003;34:1199-210.
- Turner, B.L., Mahieu, N., Condron, L.M., 2003b. Quantification of Myo-Inositol
  Hexakisphosphate in Alkaline Soil Extracts By Solution 31P Nmr Spectroscopy and
  Spectral Deconvolution. Soil Sci. 168, 469–478.
  doi:10.1097/01.ss.0000080332.10341.ed
- Turner, B.L., Mahieu, N., Condron, L.M., 2003c. Phosphorus-31 Nuclear Magnetic
   Resonance Spectral Assignments of Phosphorus Compounds in Soil NaOH–EDTA
   Extracts. Soil Sci. Soc. Am. J. 67, 497–510. doi:10.2136/sssaj2003.0497
- Turner, B.L., Newman, S., 2005. Phosphorus cycling in wetland soils. J. Environ. Qual. 34, 1921–1929.
- Turner, B.L., Paphazy, M.J., Haygarth, P.M., McKelvie, I.D., 2002. Inositol phosphates in
  the environment. Philos. Trans. R. Soc. London.Series B, Biol. Sci. 357, 449–469.
  doi:10.1098/rstb.2001.0837 [doi]
- Turner, B.L., Richardson, A.E., Mullaney, E.J., 2006. Inositol phosphates: linking agricultureand the environment. CABI.
- Vestergren, J., Vincent, A.G., Jansson, M., Persson, P., Ilstedt, U., Gröbner, G., Giesler, R.
  and Schleucher, J., 2012. High-resolution characterization of organic phosphorus in soil
  extracts using 2D 1H–31P NMR correlation spectroscopy. Environmental science &
  technology. 46, 3950-3956.
- Withers, P.J.A., Sylvester-Bradley, R., Jones, D.L., Healey, J.R., Talboys, P.J., 2014. Feed
  the crop not the soil: rethinking phosphorus management in the food chain. Environ. Sci.
  Technol. 48, 6523–6530.
- Zemunik, G., Turner, B.L., Lambers, H., Laliberté, E., 2015. Diversity of plant nutrient acquisition strategies increases during long-term ecosystem development. Nat. Plants
   1.15050
- 750
- 751
- 752

**Table 1.** pH, organic matter (OM), base cations, Total P, organic P (Po) and total NaOH-EDTA extractable P and extraction efficiency in airdried soil samples taken during the period when bluebells formed sole above-ground biomass (W1-W4) and biomass was equally distributed between bracken and bluebell above-ground (W5-W8).

	← Brac	ken fro	nds eme	rgence	$\rightarrow$					
	← Bl	uebell f	lowerin	$g \rightarrow$						
Soil sample	W1	W2	W3	W4	W5	W6	W7	W8	Mean <sup>§</sup>	SE <sup>§§</sup>
pH	4.7	4.0	4.6	4.6	4.7	4.5	4.4	4.6	4.5	0.1
OM (%)	21.2	28.7	28.6	23.9	23.9	32.9	37.1	26.1	27.8	1.7
C (%)	10.8	16.8	14.4	16.5	14.7	13.9	20.2	12.6	15.0	2.7
N (%)	0.64	0.86	0.72	0.78	0.73	0.76	0.95	0.70	0.77	0.09
C/N	16.8	19.5	20.1	21.0	20.2	18.3	21.2	18.0	19.4	1.5
C/P	144	192	167	212	137	208	200	139	175	28
C/Po	309	361	341	407	296	333	366	255	333	41
$\operatorname{Ca}(\operatorname{gkg}^{-1})$	0.29	0.31	0.39	0.43	0.57	0.63	0.70	0.42	0.47	0.05
Al $(g kg^{-1})$	11.6	12.7	14.0	12.3	14.8	14.3	14.9	16.7	14.0	0.6
$Fe (g kg^{-1})$	18.5	18.7	19.0	20.7	23.8	22.2	21.8	22.7	22.1	0.7
Total P (g kg <sup>-1</sup> )	0.75	0.87	0.86	0.78	1.1	0.70	1.0	0.90	0.87	0.05
Po (%)	64	77	78	70	71	98	75	84	77	3
NaOH-EDTA P (g kg <sup>-1</sup> )	0.50	0.70	0.61	0.61	0.71	0.53	0.81	0.65	0.64	0.03
NaOH-EDTA Po (%)	71	66	70	66	70	79	69	76	71	4
Extraction efficiency (%)	66	81	71	79	66	79	80	72	74	2
Mehlich-3 P (mg kg <sup>-1</sup> )	48.7	39.6	27.8	31.3	22.1	42.7	29.0	21.4	32.8	9.2
Mehlich-3 Al (g kg <sup>-1</sup> )	2.0	2.3	2.2	2.0	2.2	2.0	2.1	2.0	2.1	0.1
Mehlich-3 Fe (g kg <sup>-1</sup> )	0.26	0.38	0.23	0.30	0.27	0.26	0.27	0.23	0.28	0.04
$PSR (Al + Fe)^*$	0.022	0.015	0.011	0.013	0.009	0.019	0.012	0.01	0.014	0.004

<sup>§</sup> Mean value of the results obtained for W1-W8.

<sup>§§</sup>SE is the standard error of measurements.

Total (Al, Ca and Fe) and Mehlich -3 (Al, Fe and P) values are average of n = 3 (RSD $\leq 15$ ).

\* PSR, phosphorus saturation ratio

**Table 2.** Total P (n = 3, RSD $\leq 10$ ) and NaOH-EDTA extractable P and extraction efficiency in dry plant samples divided in the different below and above-ground parts.

		Blue	bell	Bracken					
	Roots	Bulbs	Scapes	Leaves	Seeds	Flowers	Rhizomes	Stipes	Blades
Total P (g kg <sup>-1</sup> )	2.7	0.84	2.9	3.5	3.3	4.0	0.14	1.5	2.8
NaOH-EDTA P $(g kg^{-1})$	2.2	0.72	2.5	3.2	3.2	3.1	0.15	1.6	2.7
Extraction efficiency (%)	81	86	86	91	99	77	107	107	96

	Inorganic P								Organic P					
		Ortho P	Pyro P	Poly P	myo- IP <sub>6</sub>	scyllo -IP <sub>6</sub>	α- gly	$\beta$ -gly	AMP	Other mono	DNA	Other diesters	Phospho nolipids	Phospho nates
Soil	W1-W8	20.0- 31.5	0.6- 2.5	0.4- 7.0	26.3- 40.0	12.3- 17.5	0.4- 2.4	0.8- 4.4	0.7- 5.3	8.4-16	1.1- 2.5	0.9-2.7	0.8-3.6	0.5-2
Bluebell	Roots Bulbs Seeds Scapes Leaves Flowers	63.2 42.0 21.7 75.5 80.4 70.4	- 1.5 - -	- - - -	39.4 60 - -	- - - -	14.4 9.4 4.5 8.1 6.8 11.2	6.0 3.6 7.7 0.9 2.9 0.9	6.4 5.5 1.4 4.1 3.5 3.0	10.0 2.8 10.1 4.0 11.3	- - - 1.8	0.5 1.3 2.3 1.4	- - - -	- - - -
Bracken	Rhizome Stipes Blades	68.5 74.1 81.1	- - -	- - -	- -	- - -	- 5.5 2.5	- 9.6 4.4	- 1.8 2.0	31.5 8.9 8.8	- - -	- - 1.1	- -	- -

**Table 3.** Relative amount (%) of the major P forms detected in the soil, bluebell and bracken plant samples.

Inorganic P								Organic P						
		OrthoP	Pyro P	Poly P	myo- IP <sub>6</sub>	scyllo -IP <sub>6</sub>	α-gly	β-gly	AMP	Other mono	DNA	Other diesters	Phospho nolipids	Phospho nates
Soil	W1-W8	105.1- 131.0	3.2- 17.6	2.1- 48.6	159.4- 259.1	77.5- 114.3	4.4- 14.7	4.9- 27.4	4.3- 32.5	50.2- 113.5	6.9- 15.3	5.5- 21.7	4.2-21.9	3.0-10.6
Bluebell	Roots Bulbs	1390.4 302.4	-	- -	283.7	-	316.8 67.7	132.0 25.9	140.8 39.6	220.0	-	-	-	-
	Seeds Scapes Leaves Flowers	701.4 1887.5 2572.8 2182.4	48.5	-		-	145.4 202.5 217.6 347.2	248.9 22.5 92.8 27.9	45.2 102.5 112.0 93.0	90.5 252.5 128.0 350.3	55.8	16.2 32.5 73.6	-	-
Bracken	Rhizome Stipes	102.8 1185.6	-	-	-	-	- 88.0	- 153.6	- 28.8	47.3 142.4			-	-
	Blades	2189.7	-	-	-	-	67.5	118.8	54.0	237.6	-	29.7	-	-

Table 3b. Absolute amount	$(mg kg^{-1})$ of the m	ajor P forms detecte	ed in the soil, bluebel	l and bracken plant samples.



**Figure 1:** Field site shown with a) bluebell dominance in mid spring and b) bracken dominance in mid summer

**Figure 2.** Percentage of biomass contribution above and below ground on dry weight basis of bluebell and bracken plants (A) and bluebell (B) and bracken (C) plant parts. W1 to W8 refers to the sampling weeks from 7<sup>th</sup> May to 25<sup>th</sup> June.





**Figure 3.** a) Solution <sup>31</sup>P NMR spectra of a representative composite soil sample and b) <sup>31</sup>P NMR unspiked and spiked spectra for the identification of *myo*-IP<sub>6</sub>(A),  $\beta$ -glycerophosphate (B), adenosine 5 monophosphate AMP (C).



**Figure 4.** a) Solution <sup>31</sup>P NMR of bluebell and bracken plants parts and b) <sup>31</sup>P NMR unspiked and spiked spectra for the identification of *myo*-IP<sub>6</sub>(A),  $\alpha$ -glycerophosphate (B\*),  $\beta$ -glycerophosphate (B), adenosine 5 monophosphate AMP (C).



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