



**Phosphorus speciation by  $^{31}\text{P}$  NMR spectroscopy 1 in bracken 2 (*Pteridium aquilinum* (L.) Kuhn) and bluebell (*Hyacinthoides non3 scripta* (L.) Chouard ex Rothm.) dominated semi-natural upland soil**

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1           **Phosphorus speciation by  $^{31}\text{P}$  NMR spectroscopy in bracken**  
2           **(*Pteridium aquilinum* (L.) Kuhn) and bluebell (*Hyacinthoides non-***  
3           ***scripta* (L.) Chouard ex Rothm.) dominated semi-natural upland soil**

4  
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25 **Abstract**

26

27 Access to P species is a driver for plant community composition based on nutrient  
28 acquisition. Here we investigated the distribution and accumulation of soil inorganic P (Pi)  
29 and organic P (Po) forms in a bracken and bluebell dominated upland soil for the period  
30 between bluebell above ground dominance until biomass is formed from half bluebells and  
31 half bracken. Chemical characterisation and  $^{31}\text{P}$  Nuclear Magnetic Resonance spectroscopy  
32 was used to determine the organic and inorganic P species. Total P concentration in soils was  
33  $0.87 \text{ g kg}^{-1}$ , while in plants (above- and below-ground parts) total P ranged between  $0.84 - 4.0$   
34  $\text{g kg}^{-1}$  and  $0.14 - 2.0 \text{ g kg}^{-1}$  for bluebell and bracken, respectively. The P speciation in the  
35 plant samples was reflected in the surrounding soil. The main forms of inorganic P detected  
36 in the NaOH-EDTA soil extracts were orthophosphate (20.0 - 31.5 %), pyrophosphate (0.6 -  
37 2.5 %) and polyphosphate (0.4 - 7.0 %). Phytate (*myo*-IP<sub>6</sub>) was the most dominant organic P  
38 form (23.6 - 40.0 %). Other major peaks were *scyllo*-IP<sub>6</sub> and  $\alpha$ - and  $\beta$ - glycerophosphate  
39 (glyP). In bluebells and bracken the main P form detected was orthophosphate ranging from  
40 (21.7 – 80.4 %) and 68.5 - 81.1 %, in above-ground and below-ground biomass, respectively.  
41 Other detected forms include  $\alpha$ -glyP (4.5-14.4 %) and  $\beta$ -glyP (0.9 -7.7 %) in bluebell, while  
42 in bracken they were detected only in stripe and blade in ranges of 2.5 - 5.5 % and 4.4 - 9.6  
43 %, respectively. Pyrophosphate, polyphosphate, *scyllo*-IP<sub>6</sub>, phosphonates, found in soil  
44 samples, were not detected in any plant parts. In particular, the high abundance of phytate in  
45 the soil and in bluebell bulbs, may be related to a mechanism through which bluebells create  
46 a recalcitrant phosphorus store which form a key part of their adaptation to nutrient poor  
47 conditions.

48

49 **Keywords:**  $^{31}\text{P}$  NMR spectroscopy; bluebell; bracken; phosphorus; phytate

## 50 **Introduction**

51

52 Plant growth depends strongly on the macronutrients NPK (Nitrogen-Phosphorus-Potassium),  
53 of which P is the least well understood in relation to the distribution and abundance of its  
54 different chemical species. Resource partitioning of P in soil dominated by diverse plant  
55 communities is critical in order to understand P cycling (Turner, 2008; Zemunik et al., 2015),  
56 particularly in view of accessing “legacy P” (Withers et al., 2014) because of predicted  
57 shortfalls of fossil P (Reijnders, 2014; Rockström et al., 2009). In soil, P can be present in  
58 various organic and inorganic forms with orthophosphate being the most readily available for  
59 organisms (Shen et al., 2011). The accurate characterisation of P species above and below  
60 ground in soil can thus provide useful information on its origins, availability and stability in a  
61 given ecosystem and it is thus essential for a better understanding of the ecology of the  
62 system under study, particularly in terms of plant accessibility versus loss through leaching  
63 that underpins primary productivity.

64 The abundance and distribution of the various P forms is strongly related to the specific  
65 environmental conditions and soil management practises (Condrón and Goh, 1990; Turner et  
66 al., 2003a; Turner and Newman, 2005; McDowell and Stewart, 2006; Cade-Menun et al.,  
67 2010; Stutter et al., 2015). In natural and semi-natural systems (i.e. no agricultural inputs), P  
68 cycling is closed with little in the way of losses, which are mainly due to leaching and largely  
69 dependent on factors such as soil parent material, topography, biomass and time (Solomon et  
70 al., 2002). In contrast, P imbalance is prominent in agricultural systems due to changes in P  
71 input and output reflected in the P speciation. For instance, inorganic orthophosphates usually  
72 account for a large proportion of total P in agricultural soils, whereas organic P forms mainly  
73 occur in natural or semi-natural soils (McDowell and Stewart, 2006; Stutter et al., 2015).

74 Bracken (*Pteridium aquilinum* (L.) Kuhn) and British bluebells (*Hyacinthoides non-scripta*  
75 (L.) Chouard ex. Rothm.) are often the dominant species in a late successional ecosystem on  
76 deforested Welsh uplands, exposed to extensive grazing pressure (Figure 1). The vegetation  
77 classification is U20: *Pteridium aquilinum* – *Galium saxatile* community (Rodwell, 1992)  
78 which is mostly found in areas of heath or grassland and has been characterised as species  
79 poor (Grime et al., 1988). The field site forms part of the Manod association that covers 5372  
80 km<sup>2</sup> in England and Wales and characterises loamy soils above 200 m AOD and annual  
81 rainfall of more than 1000 mm (Cranfield University, 2015). Extensive bluebell coverage is  
82 taken as an indicator of ancient woodlands (Rose, 1999). Equally, bracken has been described  
83 as “a plant of woodland origin, of moderate shade”. It often marks the sites of woods, which

84 have been destroyed, but when it is freed in the open, it can become a “pestilent weed” (cited  
85 in Marrs and Watt, 2006). The co-existence of both plant species is favoured by their  
86 different growth strategies with bluebells actively growing during winter and spring and  
87 bracken emerging above-ground mid to late spring and senescing in autumn. Dense bluebell  
88 and bracken populations are often associated with well-drained loamy soil (Knight, 1964) and  
89 both plants are present in all parts of the United Kingdom (for general ecological description  
90 on bluebells Blackman and Rutter, 1954 and on bracken Marrs and Watt, 2006).

91 The acquisition of P by bluebell and bracken below ground may support their dominance.  
92 Previous studies by Merryweather and Fitter (1995b) have showed an initial P inflow for  
93 bluebells during the subterranean phase, when roots are actively growing and colonised by  
94 arbuscular endophytes; while at the end of the growth phase P was lost through seeds and  
95 senesced biomass. Blackman and Rutter (1947, 1948 and 1949) instead found no significant  
96 increase in leaf weight on P addition and no significant interaction with light intensity. Seed  
97 production was only increased in response to added P and K, and flower production was  
98 enhanced only through increased light intensity. Overall Blackman and Rutter (1950) reached  
99 the conclusion that “the bluebell does not require a high level of mineral nutrients”.

100 Co-existence with bracken is attributed to their different phenologies, the reduction of  
101 competitive species by bracken and absence of heavy grazing. The acquisition of P by  
102 bracken has been less well studied. Mitchell (1973) reported that bracken acquires  
103 phosphorus through mobilisation from inorganic sources. However, her study utilised ground  
104 bracken rhizome only.

105 Some P forms are routinely measured on an operational basis by solution-based techniques  
106 (available and residual pools). These methods include sequential extraction schemes (Chang  
107 and Jackson 1957; Hedley et al., 1982; He et al., 2003), which provide valuable information  
108 on P lability and solubility in soil (Hedley et al., 1982), but it is often time consuming and  
109 tends to over or underestimate the P form in a specific defined fraction (Turner et al., 2005).  
110 For instance, the classification of organic P bioavailability based on chemical solubility can  
111 be misleading as recent studies have suggested that plants can access the supposedly  
112 “unextractable” fractions of soil organic P (Chen et al., 2002; Turner, 2008).

113 <sup>31</sup>P Nuclear Magnetic Resonance spectroscopy (NMR) is one of the most promising  
114 analytical tools which allows the identification of inorganic P forms (i.e. orthophosphate,  
115 pyrophosphate and polyphosphate) and most of the organic P forms (i.e. orthophosphate  
116 monoesters and diesters and phosphonates) simultaneously (Cade-Menun et al., 2010; Turner  
117 et al., 2005). The technique has already been used for the characterisation of P forms in soil

118 samples and for the evaluation of the effects of different soil types, farming practises and land  
119 use on the distribution and transformations of P forms in soils (Cade-Menun and Preston,  
120 1996; McDowell et al., 2005; McDowell and Stewart, 2006; Cade-Menun and Liu, 2014;  
121 Stutter et al., 2015). However, while there is an increasing number of publications on soils  
122 under native vegetation (McDowell and Stewart, 2006; McDowell et al., 2007; Turner et al.,  
123 2007; Stutter et al., 2015), fewer have investigated soils under native vegetation and the  
124 related contribution of the most dominant plant species to the soil P pools using  $^{31}\text{P}$  NMR. In  
125 particular, to our knowledge, no studies have reported the characterisation of P species found  
126 in a bracken and bluebell native vegetation and soil system; and only a few studies have  
127 reported the use of  $^{31}\text{P}$  NMR for the determination of P species in various plant parts  
128 (Makarov et al., 2002; Makarov et al., 2005; Noack et al., 2012).

129

130 This study thus investigated the P species in soil and plants from a natural vegetation system  
131 dominated by bracken and bluebell using  $^{31}\text{P}$  NMR and established assays targeting labile P  
132 species (Mehlich-3 extractable) when both plants show active above ground growth. The aim  
133 was to probe the mechanisms regulating the composition and nature of P forms in this type of  
134 soil and vegetation community.

135

## 136 **Materials and methods**

137

### 138 Sample collection and preliminary analysis

139

140 The soil and plant samples used in this study were collected in an area located at 250 m above  
141 sea level in the Snowdonia National Park (Llanberis, United Kingdom, N53°07' W 04°08').  
142 The whole area encompasses about half a hectare with full bracken (*Pteridium aquilinum* (L.)  
143 Kuhn) and bluebell (*Hyacinthoides non-scripta* (L.) Chouard ex Rothm) coverage (Figure 1  
144 and Supplementary Information (SI)). Most of the root systems of both plants, rhizome for  
145 bracken and bulbs for bluebells, grow intertwined. Bracken rhizomes form a dense 10 cm  
146 thick layer located approximately 5 to 10 cm below the soil surface. Bluebells propagate  
147 predominantly from seeds with small bulblets forming in the first growth period. As perennial  
148 plants, the bulb increases in size every year and the roots are contractile. Hence, bluebell  
149 bulbs extend downwards during active growth. Young bulbs are located above the bracken  
150 rhizome and with increasing age they find their way through the rhizome layer. Mature bulbs  
151 are found below the rhizome layer. Above and below-ground biomass of other plant species

152 accounted for less than 5 % of the total biomass. The area falls under the upland vegetation  
153 type U20a (*Pteridium aquilinum-Gallium saxatile* community U20, *Anthoxanthum odoratum*  
154 sub-community U20a) (Rodwell, 1992) with well-drained and infertile soils. No history of  
155 fertiliser application on these fields was reported and no grazing has been applied as a  
156 management regime, with the surface litter mostly dominated by dead bracken fronds. The  
157 site is hence classed as semi-natural.

158 A stratified random sampling approach was undertaken, because of the near-level surface of  
159 the field. Soil samples (0-15 cm) were collected proportionally around segments with high  
160 density of both plants, based on previous growth history and visual inspection. A 0-15 cm  
161 sampling depth was chosen instead of the recommended 0 -7 cm for undisturbed soil, because  
162 bluebell bulbs on the field site grow in colonies occurring at depths between 5 to 20 cm. Due  
163 to the heavy nature of the soil (68 % slit), bulbs usually occur between the Ah horizon and  
164 the upper Bs horizon (Grabham and Packham, 1983; Merryweather and Fitter, 1995a). The  
165 rationale was that since no form of agricultural land use (i.e ploughing) was reported for the  
166 field site, a depth of 0-15 cm was sufficient in estimating the soil nutritional properties. That  
167 was based on the assumption that there was very little variability (less than 15 %) on the  
168 field. We however, acknowledge that a small amount of error may have occurred due to  
169 incomplete sampling of some horizons (less than 15cm), because of the nature of the soil  
170 parent material, which consists of mainly metamorphic rock deposits of dark purple slate.

171 Soil and plant sampling was carried out using a 15-cm soil auger (Eijkelkamp, Holland). Two  
172 soil cores were collected weekly during the main growing stage of the plants, from 7<sup>th</sup> May  
173 2013 (week 1) to 25<sup>th</sup> June 2013 (week 8), a total of 8 samplings (W1-W8).

174 A total of 16 soil cores (two per week) were collected and processed in the laboratory by  
175 hand. Soil samples were air dried, ground in a porcelain mortar, passed through a 2 mm sieve  
176 and combined to form a composite sample for each week of collection (W<sub>s</sub>1- W<sub>s</sub>8). Plants  
177 were separated from the soil, thoroughly cleaned from any soil remains and further separated  
178 into below and above-ground parts. In particular, for bluebells, the below-ground part  
179 included bulbs and roots and the above- ground included scapes, leaves and flowers; while  
180 for bracken, rhizome was separated from the frond (stipes and blades). These parts were  
181 freeze-dried for one week and the percentage of the total dry weight was calculated. Each  
182 plant part was then ground in a porcelain mortar and stored at 4 °C until further analysis.

183 The following chemical and physical parameters were determined on the composite soil  
184 samples W<sub>s</sub>1- W<sub>s</sub>8: *i*) Particle size analysis was carried out using a particle size analyzer  
185 (Malvern Mastersizer 2000, UK) and the soil texture was classified using the USDA triangle.

186 *ii*) Soil pH was measured in H<sub>2</sub>O (ratio soil: water 1:2.5 w/v) using an Orion 420A pH meter  
187 (Boston, USA). *iii*) Soil organic matter (OM) was determined by loss of ignition at 450 °C  
188 (4hrs) in a muffle furnace (Carbolite, UK) after oven drying at 110°C. Total carbon (C) and  
189 nitrogen (N) in soil, were determined on a LECO Truspec, CN Analyzer.

190 Total P, aluminium (Al), iron (Fe) and calcium (Ca) were determined in soil and plant (P  
191 only) samples, using a nitric acid (HNO<sub>3</sub>) digestion method. Soil extractable nutrients (P, Al,  
192 and Fe) were also determined by a Mehlich-3 extraction (Mehlich, 1984). The resulting  
193 solutions were analysed via an ICP-AES Varian 710ES (Agilent Technologies, USA).

194 Soil total organic P was determined using the ignition method of Saunders and Williams  
195 (1955). Ignited and unignited extracts were determined based on the colorimetric method of  
196 Murphy and Riley, (1962).

197 Calculation of the Phosphorus Saturation Ratio (PSR) was performed using the formula

$$PSR = \frac{P}{Fe + Al}$$

198

199 where P, Fe and Al concentrations were determined by Mehlich -3 extraction

200

201 Pearson pair-wise correlations between sets of data was performed using the statistical  
202 Package IBM SPSS Statistics (version 22.0) with significance set at  $p < 0.05$ .

203

204 <sup>31</sup>P NMR: Sample preparation and analysis

205

206 The sample preparation for solution <sup>31</sup>P NMR Nuclear Magnetic Resonance spectroscopy  
207 was performed using a modified version of the Cade-Menun and Preston (1996) procedure.

208 Three grams of air dried soil or 1 g of crushed freeze-dried plant sample was mixed with 25  
209 mL of a solution of 0.25 M NaOH and 0.05 M Na<sub>2</sub>EDTA and shaken at 250 rpm and 20 °C  
210 for 6 hrs (soil) or 4 hrs (plant) hours. The extracts were then centrifuged for 20 min at 5000  
211 rpm and filtered using Whatman No.42 filter paper. An aliquot of 0.5 mL was then diluted for  
212 ICP-AES analysis and the remaining solution was freeze-dried. The efficiency of P extraction  
213 had a mean value of 74 % for soil sample and 91 % for plant sample (individual data are  
214 given in Table 1 and 2).

215 Approximately 100 mg of each freeze dried extract was redissolved in 1 mL of D<sub>2</sub>O, 0.6 mL  
216 10 M NaOH and 0.4 mL extracting solution (0.25M NaOH + 0.05M Na<sub>2</sub>EDTA) (Cade-  
217 Menun and Liu, 2014). A post extraction step was carried out only for soil samples as



218 described by Verstergren et al., 2012: An excess of sodium sulphide ( $\text{Na}_2\text{S}$ ) was added to the  
219 redissolved sample to ensure precipitation of some of the metals. The solution was then  
220 allowed to stand for 2 hours. Samples were then centrifuged for 40 minutes at 5000 rpm (to  
221 remove particles that might contribute to line broadening), transferred to a 5 mm NMR tube  
222 and analysed via  $^{31}\text{P}$  NMR spectroscopy. A comparison of spectral quality between the  
223 addition of sodium sulphide and no addition show improved spectroscopic resolution (Figure  
224 S2 SI) and hence reduced scan time.

225 Spectra were acquired on a Bruker Advance DRX 400 MHz NMR spectrometer (7.5T, 161.9  
226 MHz), equipped with a 5 mm broadband probe at 20 °C. Instrument parameters were a 90°  
227 pulse, 0.68 s acquisition time and recovery delay of 4.32s to 15s and inverse gated proton  
228 decoupling (waltz 16) were used, and set to at least five times the  $T_1$  (lattice relaxation time)  
229 based on the P / (Fe + Mn) mass ratios. The experiments required between 1000 and 2500  
230 scans (1-2 h running time) for plant and 4000 to 5000 scans (6-7 h running time) for soil  
231 samples to achieve a good signal to noise ratio. The spectral width used was 8090.6 Hz and  
232 the number of data points was 11002. A delay time of between 3 to 5 seconds has previously  
233 been reported to be sufficient to obtain quantitative spectra of NaOH-EDTA in similar soil  
234 extracts (McDowell et al., 2006, Stutter et al., 2015). The chemical shift (ppm) of the signals  
235 was indirectly referenced to an external 85 %  $\text{H}_3\text{PO}_4$  standard via the lock signal. Peaks were  
236 defined by three parameters: chemical shift, line width and peak height. Peak assignment was  
237 based on soil and plant extracts spiked with standard solutions and by comparisons to  
238 literature data (Turner et al., 2003b, 2003c; Makarov, 2005; McDowell et al., 2005; Smernik  
239 and Dougherty, 2007; Doolette et al., 2009; Cade-Menun et al., 2010; Cade-Menun ,2015).  
240 Spiked solutions were used for the identification of phytate (*myo*-IP<sub>6</sub>),  $\alpha$  and  $\beta$   
241 glycerophosphate and adenosine 5-mono phosphate peaks. Soil or plant extracts were spiked  
242 either with 0.1 mL of a 2.1 g L<sup>-1</sup> aqueous phytate solution (Na salt hydrate from rice, Sigma  
243 Aldrich P8810) or with 0.1 mL aqueous solutions of 4.0 g L<sup>-1</sup> of an isomeric mixture of  $\alpha$   
244 and  $\beta$  (1:1) glycerophosphate disodium salt hydrate (Sigma Aldrich G6501). Soil extracts  
245 were also spiked with 0.1 mL of a 4.4 g L<sup>-1</sup> of adenosine-5-monophosphate disodium salt  
246 (Sigma Aldrich 01930).

247 Integration of peak areas were calculated on spectra processed with a line broadening of 1-3  
248 Hz using a Bruker Topspin 2.0 software and MestReNova v.6.0. Quantification of P species  
249 was done by spectra deconvolution analysis, which proved to be successful in particular for  
250 areas such as the monoester region containing a number of peaks, sometimes overlapping; the

251 relative P concentration in the NaOH-EDTA extracts was estimated on the based on the total  
252 NMR signal area and presented as percentages of each species. If specific identification could  
253 not be made, they were grouped into compounds or compound classes (Cade-Menun, et al.,  
254 2010, Doolette et al., 2009).

255

## 256 **Results**

257

### 258 Soil and plant chemical characteristics

259

260 The soil, a brown podzolic soil termed Manod Association (Cranfield University, 2015), was  
261 classified as silt-loam soil (sand: 24 %, silt: 70 %, clay: 6 %). All collected soil samples  
262 showed an acidic pH in water, between 4.0 and 4.7. Soil organic matter content (OM %)   
263 ranged from 21.2 % - 37.1 %; total C and N content ranged from 10.8 % - 20.2 % and 0.65 %  
264 - 0.95 %, respectively with no major changes in the C/N ratio (19.4 mean value) (Table 1)

265 Total P in soil was between 0.70 - 1.1 g kg<sup>-1</sup> (mean value 0.87 g kg<sup>-1</sup>) of which between 64 %  
266 - 98 % (mean value 77 %) was organic P (Po) (Table 1). Of the total metals analysed, Fe and  
267 Al ranged from 18.5 – 22.8 g kg<sup>-1</sup> (mean value 22.1 g kg<sup>-1</sup>) and 11.6 – 16.7 g kg<sup>-1</sup> (mean  
268 value 14.1 g kg<sup>-1</sup>), respectively, while Ca ranged from 0.29 – 0.70 g kg<sup>-1</sup> (mean value 0.47 g  
269 kg<sup>-1</sup>). Low concentration of Ca were also reflected into the low pH value (mean value 4.5) of  
270 the soil. Mehlich 3-extractable P ranged from 21.4 – 48.7 mg kg<sup>-1</sup> (mean value 32.8 mg kg<sup>-1</sup>)  
271 and was negatively correlated with total P ( $r = - 0.74$   $p < 0.01$ ), but was strongly positively  
272 correlated with the Phosphorus Saturation Ration (PSR) ( $r = 0.97$   $p < 0.01$ ).

273 The total C to P and C to organic P (Po) ratios are given in Table 1 and showed a highly  
274 significant positive correlation ( $r = 0.85$ ,  $p < 0.01$ ), while the N/P to C/P ratio and the N/P  
275 ratio to the C/Po ratio showed significant correlations ( $r = 0.74$ ,  $p < 0.05$ ,  $r = 0.81$ ,  $p < 0.05$ ,  
276 respectively). None of the correlations for total N with total C/P, C/Po, total P or Mehlich 3-  
277 extractable P were significant. (Table S1a in SI)

278 Total P in plant samples was higher in the above ground part for both bluebells and bracken  
279 (Table 2) with total P ranging between 0.84 - 4.0 g kg<sup>-1</sup> and 0.14 - 2.8 g kg<sup>-1</sup>, respectively. In  
280 particular, the P concentrations were in the order flowers>leaves>scapes>roots>bulbs for  
281 bluebells and blade>stipe>rhizome for bracken.

282 During the sampling period bluebell leaves were dominating above-ground for weeks 1 to 3  
283 (Figure 2a). Peak flowering started in week 4 and a bluebell flower carpet was domineering  
284 during weeks 4 and 5 with few bracken fronds emerging. These weeks showed biomass

285 accumulation solely occurring for bluebells whose most active photosynthetic phase was  
286 occurring between weeks 1 to 5. At week 6 bluebell flowers started to fade and seed capsules  
287 started to form while bracken frond density increased. At week 7 bracken shoots were higher  
288 than fading bluebell flowers. From week 8 onwards bracken was the visually dominant plant  
289 above ground on the site and the bluebell flowers have turned into seed capsules (photo  
290 record is available in SI). As shown in Figure 2b and 2c, the below ground processes  
291 contributed a constant 40 % to the total biomass allocation. Until week 4, active  
292 photosynthesis contributed to bluebell biomass gains, while bracken biomass stayed constant  
293 up to week 6.

294

295 Phosphorus forms in soil

296

297 Solution  $^{31}\text{P}$  NMR results showed the presence of the same P species in all soil samples  
298 (Figure 3a and SI). Mean extraction efficiency of total P in the NaOH-EDTA extract was 74  
299 %, and was negatively correlated to pH ( $r = -0.73$ ,  $p < 0.05$ ).

300 Detected inorganic P compound classes accounted for a total amount of 30 – 41 % including  
301 orthophosphate between 5.95 ppm and 6.11 ppm in the range of 105.1 - 131.0 mg kg<sup>-1</sup> (20.0 –  
302 31.5 %), pyrophosphate at -3.75 ppm in the range 3.2 - 17.6 mg kg<sup>-1</sup> (0.6 – 2.5 %) and  
303 polyphosphates at -3.56 ppm ranging from 2.1- 48.6 mg kg<sup>-1</sup> (0.4 - 7.0 %). Organic P  
304 compound classes (59 – 70 % of total NaOH-EDTA extractable P) included phosphonolipids  
305 (18.0 ppm) between 4.2 - 21.9 mg kg<sup>-1</sup> (0.8 % - 3.6 %) and phosphonates (20.1 ppm) ranging  
306 from 3.0 - 10.6 mg kg<sup>-1</sup> (0.5 - 2.0 %). The orthophosphate diesters were divided into  
307 deoxyribonucleic acid (DNA) at -1.0 ppm and other diesters from 2.1 to -2.6 ppm. In the  
308 orthophosphate monoester region (2.9 – 5.7 ppm), the four peaks for phytate (*myo*-IP<sub>6</sub>) at  
309 5.27 ppm, 4.38 ppm, 3.98 ppm and 3.84 ppm were identified. Other major peaks detected in  
310 this region were *scyllo*-IP<sub>6</sub> at 3.7 ppm,  $\alpha$ - and  $\beta$ -glycerophosphate ( $\alpha$ -glyp and  $\beta$ -glyp,  
311 respectively), that are phospholipid degradation products, and adenosine-5-monophosphate  
312 (AMP). *myo*-IP<sub>6</sub> was confirmed after spiking, while degradation products;  $\alpha$ -glyp,  $\beta$ -glyp and  
313 AMP were also identified (Figure 3b). Other unidentified monoesters between 2.9 ppm and  
314 5.7 ppm were grouped as other monoesters (Table 3a-b). From our results, NMR-based Po  
315 speciation (average Po = 71 %) was in line with the ignition method of Saunders and  
316 Williams (1955) (average Po=77 %, in Table 1) and showed a significant correlation with  $r =$   
317 0.73.

318 Correlation coefficients for soil physico-chemical properties and P species determined in the  
319 NaOH- EDTA soil extracts are shown in Table S1 in SI. Focussing mainly on the significant  
320 correlations for C, N and extractable P, total NaOH-EDTA P was positively correlated with  
321 total P ( $r = 0.84, p < 0.01$ ), total C ( $r = 0.82, p < 0.05$ ), total N ( $r = 0.78, P < 0.05$ ) and C/N  
322 ratio ( $r = 0.72, p < 0.05$ ). It was however, negatively correlated with and Mehlich-3  
323 extractable P ( $r = - 0.61$ ). For the P species quantified using  $^{31}\text{P}$  NMR, inorganic  
324 orthophosphate concentration was positively correlated with total C ( $r = 83, p < 0.05$ ), total N  
325 ( $r = 0.72, p < 0.05$ ), C/N ratio ( $r = 0.82, p < 0.05$ ), total P ( $r = 0.74, p < 0.05$ ) and total  
326 NaOH-EDTA P ( $r = 0.88, p < 0.01$ ). Polyphosphate on the other hand, was strongly  
327 negatively correlated with pH ( $r = - 0.85, p < 0.01$ ) and strongly positively correlated with  
328 Mehlich-3 extractable Fe ( $r = 0.77, p < 0.05$ ). Orthophosphate monoesters were the most  
329 dominant group of Po compounds in the field. Their concentration as whole (sum of all  
330 detected monoesters) was negatively correlated Mehlich-3 extractable P ( $r = - 0.63, p < 0.05$ )  
331 and strongly positively correlated with total P ( $r = 0.79, p < 0.05$ ) and total NaOH-EDTA P ( $r$   
332  $= 0.84, p < 0.01$ ). The most dominant orthophosphate monoester *myo*-IP<sub>6</sub>, did not show any  
333 significant correlation with most of the soil physio-chemical properties. However, it was  
334 strongly negatively correlated with N/Po ratio ( $r = - 0.80 p < 0.05$ ).

335

### 336 Phosphorus forms in plants

337

338 Figure 4a shows solution  $^{31}\text{P}$  NMR results for all plant parts. The main P form in bluebells  
339 was orthophosphate (5.34 – 5.76 ppm), found in the range 302 -2573 mg kg<sup>-1</sup> (21.7 – 80.4  
340 %). Orthophosphate decreased in absolute and relative amounts from leaves > scapes >  
341 flowers > roots > bulbs > seeds. *myo*-IP<sub>6</sub> (5.08 ppm, 4.18ppm, 3.81 ppm and 3.79 ppm) was  
342 the major P form in bluebell seeds 1939 mg kg<sup>-1</sup> (60%) and bulbs 283.7 mg kg<sup>-1</sup> (39.4 %).  
343 The other species detected in all bluebell plant parts were phospholipid degradation products  
344  $\alpha$ -glyp (67.7- 347.2 mg kg<sup>-1</sup>, 4.5-14.4 %) and  $\beta$ -glyp (22.5-130 mg kg<sup>-1</sup>, 0.9 – 7.7 %) detected  
345 at 4.45 ppm and 4.10 ppm respectively. Ribonucleic acid derived AMP (4.02 ppm) was in the  
346 range 39.6 - 140.8 mg kg<sup>-1</sup> (1.4 - 6.4 %), but absent in bulbs. *myo*-IP<sub>6</sub>,  $\alpha$ -glyp and  $\beta$ -glyp and  
347 AMP were confirmed after spiking (Figure 4b). Deoxyribonucleic acid was only detected in  
348 bluebell flowers (55.8 mg kg<sup>-1</sup>, 1.8 %). Other monoesters, likely to include sugar phosphates,  
349 and lower inositol phosphates were between 90.5 - 350.3 mg kg<sup>-1</sup> (2.8 % - 11.3 %) and were  
350 not detected in bulbs. Other diester P forms, e.g. non-hydrolysed phospholipids, were in the  
351 range 32.5 - 73.6 mg kg<sup>-1</sup> (0.5 -14.3 %).

352 The main P form detected in bracken was also orthophosphate (102.8 - 2189.7 mg kg<sup>-1</sup>, 68.5 -  
353 81.1%, blade > stipe > rhizome), followed by monoester P forms (47.3 - 237.6 mg kg<sup>-1</sup>, 8.8 -  
354 31.5 %, rhizome > stipe > blade).  $\alpha$ -glyph and  $\beta$ -glyph were detected only in stipes and blades  
355 in ranges (67.5 - 88 mg kg<sup>-1</sup>, 2.5 - 5.5 %) and (118.8 - 153.6 mg kg<sup>-1</sup>, 4.4 - 9.6 %)   
356 respectively, with stipes showing higher values. Adenosine-5-monophosphate was similar  
357 between stipes and blades (about 2 %) and absent in rhizome. Other possible diester P forms  
358 were detected only in bracken blades in very small amounts 29.7 mg kg<sup>-1</sup> (1.1 %).  
359 Pyrophosphate, polyphosphate, *scyllo*-IP<sub>6</sub>, phosphonolipids and phosphonates, which were  
360 found in soil samples, were not detected in any plant parts (Figure 4).

361

## 362 **Discussion**

363

### 364 Soil and plant chemical characteristics

365

366 Bluebell and bracken often form dense co-existing communities on acidic, nutrient poor, and  
367 well drained (sandy) loamy soils with few other plant species present (Knight, 1964;  
368 Grabham and Packham, 1983; Merryweather and Fitter, 1995). Comparably, the field site  
369 used in this study presented a loamy texture and low pH but high total P content (McDowell  
370 and Stewart, 2006), mostly organically bound (Condon and Goh, 1990; Hawkes et al., 1984)  
371 and hence not directly bioavailable. Thus the limited availability of P could be the limiting  
372 factor for plant growth and access to this limited pool could thus contribute to the successful  
373 establishment and maintenance of specific species, i.e. bluebells and brackens. Organic P can  
374 be an essential component of soil solution pool during periods of P limitation (Shen et al.,  
375 2011). The positive relationship observed between total P and total NaOH-EDTA P (both  
376 consisting of large amounts of Po) suggest that fractions of the soil's organic P is labile and  
377 may contribute to the soil solution phase. The period under study described the shift from  
378 bluebells dominating, with their period of most active growth and biomass accumulation  
379 terminating with the onset of seed ripening in W5, to bracken dominating the above ground  
380 growth from W6 onwards. A decline in Mehlich-3 extractable P and an increase in total P  
381 from W1 to W5 was observed. The shift from a bluebell dominance culminating in seed  
382 setting in W6 and concurring with higher bracken biomass showed a decline in total P and a  
383 doubling of Mehlich-3 extractable P in W6. Bracken dominance in W7 and W8 was reflected  
384 in a decline in Mehlich-3 extractable P. but an increase in total P and total NaOH-EDTA P  
385 when compared to W6 (Figure S3 SI).

386 The Phosphorus Saturation Ratio (PSR), which is a measure of the soil capacity to retain P,  
387 gives an estimate of the extent to which potential adsorption sites (Fe and Al) in the soil have  
388 been saturated with P. In this study, the PSR (Al + Fe) did not exceeded the environmentally  
389 critical PSR limit of 0.15 (Table 1), in fact it was a magnitude smaller indicating the  
390 limitation of available P with a large capacity to adsorb phosphate, should it become  
391 available (Nair, 2014). Amorphous oxides of Fe and Al are influential for P sorption in acidic  
392 soils reducing its availability (Turner et al., 2006). The Fe and Al content in our soil indicates  
393 the possibility of P being fixed with Fe and Al hydrous oxides or being precipitated as  
394 insoluble Al and Fe phosphates. In addition, the total C to Po ratio used as an estimate for  
395 determining if net mineralization (< 200) or immobilization (> 300) is occurring in soils  
396 (Dalal, 1977). For most of the weeks sampled, (Table 1) the values were greater than 300,  
397 suggesting that net immobilization of P (imbalanced in the P cycle) was occurring in the soil,  
398 favoring the accumulation of P in organic form.

399 The P content in the different plant parts agreed with previous studies on bluebells (Blackman  
400 and Rutter, 1949; Merryweather and Fitter, 1995) and bracken (Ferguson and Armitage,  
401 1944; Moon and Pal, 1949). The amount of P varied according to the different plant parts,  
402 with higher values in the above-ground parts (leaves, scapes, flowers for bluebell, stipes and  
403 blades for bracken) and with bluebells showing the highest concentrations. The bluebell  
404 species under study is triploid (Grundmann et al., 2010) and has a larger genomic DNA size,  
405 as it is often found for early spring flowering species (Hendry, 1987). This implies higher  
406 demand for P during growth and subsequent higher P concentration in tissue. In general, P in  
407 plants is preferentially transferred to leaves and flowers where it is needed for photosynthesis,  
408 pollen and seed formation (Schachtman et al., 1998; Shen et al., 2011) and the highest P  
409 content was determined in bluebell flowers and seeds followed by leaves and bracken blades.

410

#### 411 Phosphorus forms in soil

412

413 The Pi content of the soil ranged from 20.0 to 31.5 %, this low Pi content seems to be  
414 reflected in the nature of the soil, largely controlled by its associations with Al and Fe –  
415 oxides. This is shown by the relationship between orthophosphate and Mehlich-3 extractable  
416 Al and Fe. However, due to the closer correlation of orthophosphate with Mehlich-3  
417 extractable Al, the formation of Al-P compounds in the soil is highly favoured. The strong  
418 relationship observed between orthophosphate with total C, also suggest that its sorption to  
419 OM increases as the level of total P in the soil increases.

420 Organic P content comprised a large part of the total P (up to 80 %) in soil mainly consisting  
421 of orthophosphate monoesters: *myo*-IP<sub>6</sub>, followed by its stereoisomers *scyllo*-IP<sub>6</sub>. They are  
422 derived from plant and microbial sources, but may also include NaOH-EDTA extraction  
423 degradation products of phospholipids and RNA (Turner et al 2002; Makarov et al., 2002;  
424 Makarov et al., 2005; Bünemann et al., 2008; Cade-Menun, 2015). The large charge density  
425 of higher orthophosphate monoesters contributes to their strong sorption to soil by metal  
426 oxide in preference to orthophosphate. Complexation and precipitation reaction with  
427 polyvalent cations inhibit both chemical and enzyme-mediated biological attack (Turner et  
428 al., 2002). This study showed a closer relationship between Mehlich-3 extractable Al, rather  
429 than Fe, and orthophosphate monoesters thus suggesting that Al is more essential for P  
430 sorption. These associations can either act as sources of P during periods of limitation,  
431 supplying labile forms of P to the solution phase or as soil P sinks. The negative correlation  
432 between orthophosphate monoester and pH suggests that the stability of their association may  
433 decline with increasing pH. The high C/P<sub>o</sub> ratio (>333 mean value) supports the formation of  
434 OM - Al - *myo*-IP<sub>6</sub> - complexes, as *myo*-IP<sub>6</sub> is the most dominant soil P species (159.4 –  
435 259.1 mg kg<sup>-1</sup>, 39 - 52 % of organic P), with resulting immobilisation of P in the soil.  
436 The other inorganic P species detected in soil were pyrophosphate and polyphosphates. The  
437 other major monoesters were α-glyp, β-glyp (phospholipid degradation products) and RNA  
438 derived AMP; and the diesters, including nucleic acids (DNA and RNA) and non-hydrolysed  
439 phospholipids. The last class of organic P compounds detected were the phosphonates and  
440 phosphonolipids found in soil samples only. Their likely origins have been extensively  
441 studied by various authors (Makarov et al., 2002; Makarov et al., 2005; Turner et al., 2005;  
442 Doolette et al., 2009; Bünemann et al., 2008; Turner, 2008; Cade-Menun et al., 2010).

443

#### 444 Phosphorus forms in plants

445

446 All bracken and actively growing bluebell parts (roots, scapes, leaves and flowers) contained  
447 a significant percentage of P<sub>i</sub> (60 to 80 %) in the form of orthophosphate  
448 i.e.  $H_2PO_4^-$  and  $HPO_4^{2-}$ . This was consistent with previous studies on plant material that  
449 found a range of 25 to 75 % P<sub>i</sub>. Bluebell seeds and bulbs on the other hand, contained only  
450 21.7 % and 42.0 % orthophosphate, respectively, which is within the range usually found in  
451 seeds (Noack et al., 2012). *myo*-IP<sub>6</sub> was the most abundant organic P form detected in  
452 bluebell bulbs (39.4 %) and seeds (60 %) but not in any other bluebell or all bracken parts.  
453 This is consistent with previous work on seeds, which reports that *myo*-IP<sub>6</sub> represents about

454 50 to 80% of seed P. The high *myo*-IP<sub>6</sub> content in bluebell bulbs is, to the best of our  
455 knowledge, the first report of *myo*-IP<sub>6</sub> in bulbs. Additionally, the bulbs did not contain other  
456 monoesters The other inorganic P species detected only in bluebell seeds was pyrophosphate.

457

#### 458 Ecological implications

459 The bracken and bluebell dominated ecosystem presented in this study could be an example  
460 of the co-existence of two species with different nutrient acquisition strategies in relation to P  
461 based on their differences in P speciation in plant. This P speciation in the plant biomass is  
462 linked through litter input with the P speciation in the soil. Overall P returned to soil from the  
463 vegetative plant parts (leaves, roots and inflorescences) would contribute mainly to the  
464 orthophosphate and diesters fractions, while bluebell bulbs and seeds would predominately  
465 contribute to the P<sub>O</sub> fraction in soil. Abiotic factors such as sorption / desorption, weathering  
466 and microbially-mediated P immobilization may change the P speciation in soil away from  
467 that originally found in the plant litter. The soils forming part of the Manod Association,  
468 however, are described as having a thick surface mat or roots and plant remains when on  
469 slopes to steep or rocky to cultivate. The comparatively high phytate content in bluebell bulbs  
470 thus determines the equally high phytate content in the surrounding soil as during flowering  
471 the old bulb is shed. At this stage there is significant loss of P to the surrounding soil.  
472 Previous studies have shown that *myo*-IP<sub>6</sub> additions to soil can led to the release of  
473 orthophosphate and OM into soil solutions (Anderson et al., 1974; Leytem et al., 2002). From  
474 our results, towards the end of flowering, W5 to W7, the bluebell plant also shed its leaves  
475 and inflorescences, at this point the plants not only losses P, but carbon to the surrounding  
476 soil, reflected by the increase in available P and the PSR ratio in weeks 6. This might also be  
477 the likely reason for the change in OM content noticed in Weeks 6 and 7.

478 Bluebells had been shown to store only half their acquired P in the new bulb because of the  
479 old bulb disappearing. The P speciation reported here support this bulb shedding which  
480 results in a net P flux from the dense bluebell population into the surrounding soil  
481 (Merryweather and Fitter, 1995b). With the remnants of the “old” bulb, a store is created near  
482 the growing location which includes essential elements for future growth and P in the form of  
483 *myo*-IP<sub>6</sub>. Access to this stored phytate may be achieved through arbuscular mycorrhiza (AM)  
484 on which bluebells are dependent (Merryweather and Fitter, 1995a). The roots of the bluebell  
485 plant are usually colonised by AM immediately after emergence. Studies have shown that this  
486 AM association can significantly increase the surface area of the bulbs root system, enabling  
487 it to access deeper layers of the soil profile (Merryweather and Fitter, 1995a,1995b), thus



488 likely increasing its proximity to substrates (i.e. *myo*-IP<sub>6</sub>), which is one of the essential  
489 requirement for plants who may likely utilize *myo*-IP<sub>6</sub>. (Richardson et al., 2006). Since the  
490 soil used in the present study was undisturbed, inositol phosphates would likely accumulate  
491 on the soil surface through shedded seeds and directly in the soil through the shedded bulb.  
492 Bluebells have contractile roots through which they are able to migrate to deeper depth in the  
493 soil profile (> 20 cm) (Grabham and Packham, 1983). AM increases the root phosphatase  
494 activity of their host and also produce an extracellular membrane-bound phytate degrading  
495 enzyme in its hyphae, which could aid root phosphatases in the hydrolysis of Po compounds  
496 (Richardson et al., 2001). The bracken plant, on the other hand, contains a thick root like  
497 rhizomes with tiny hair-like black roots forming a vast network located about 10 to 20 cm  
498 underground. Bracken has no reported AM association with its rhizome thus, P uptake is  
499 based on its rhizomes root phosphatase activity, exudation of acids and microbially-mediated  
500 hydrolysis. Exudates from plants root alone are not capable of utilizing P directly from *myo*-  
501 IP<sub>6</sub>, due the very low level of extracellular phytase they contain, but are dependent on  
502 microbial-mediated (i.e. fungi) hydrolysis. Hence bracken rhizome would probably not be  
503 able to utilize *myo*-IP<sub>6</sub> directly unlike plant roots with AM association (Richardson et al.,  
504 2006). This further supports our theory that bluebells are more likely to utilize *myo*-IP<sub>6</sub>  
505 compared to bracken with its non-AM rhizome counterpart. We however, acknowledge that  
506 some form of microbial-mediated hydrolysis may also be involved, likely fungi due to the  
507 acidic nature of the soil. Determination of phospholipid fatty acids (PLFAs) from this  
508 bluebell and bracken site showed a higher concentration of fungal biomarkers compared to a  
509 bracken only site on similar soil (unpublished data).

510 The suppression of bracken crozier emergence during bluebell flowering of the dense  
511 population was observed (Figure 1) compared to that of croziers emerging at the same  
512 geographic location but with lower bluebell density three weeks earlier (SI). Week 5 showed  
513 the highest total P concentration in soils for all sampling occasions combined with a low  
514 extraction efficiency, caused by litter input through shed bulbs. Croziers start to emerge  
515 during week 5 (see photos in SI). Week 6 showed both a reduction in total P in soil and a  
516 predominance of organic P (98%), thought to indicate the assimilation of Pi by bracken roots  
517 to support above ground growth.

518 The ecological consequence of the results presented here support a number of observation in  
519 relation to the ecology of established bluebell populations and means of spread. Bluebells are  
520 taken as indicators of ancient woodlands, where nutrient status is poor and P is mostly stored  
521 as phytate (Attiwill and Adams, 1993, Turner et al., 2002). Observations on the field site

522 included the establishment of bluebell clumps from seed stores, which could supply a  
523 preferred source of phytate from ungerminated seeds as their content was 60 % *myo*-IP<sub>6</sub>. In  
524 addition, plant establishment from bulbs is more successful if bulbs are planted close  
525 together, which again increases concentration of phytate through the disappearing old bulbs  
526 (Merryweather and Fitter 1995b). The retention of phytate in acidic and Al and Fe rich soils,  
527 preferred bluebell habitats, is hence a chemical mechanism that supports the long-term  
528 maintenance of bluebell populations on natural soils. The bluebell bracken dominated  
529 ecosystem is an example where access to resources is determined by both phenology for  
530 primary production or nutrient acquisition and storage for P.

531

## 532 **Conclusions**

533

534 In this study, investigations on the major P forms in a semi-natural upland soil dominated by  
535 bluebell and bracken showed that the distribution of the major soil P species was determined  
536 by the present vegetation. <sup>31</sup>P NMR spectroscopy showed that there was a dominance of more  
537 recalcitrant organic P forms (i.e. monoesters) compared to more readily available inorganic P  
538 form (i.e. orthophosphate). In particular, *myo*-IP<sub>6</sub>, the most dominant monoester form in the  
539 soil, was also found at similar concentrations in bluebell bulbs, suggesting that annual  
540 shedding of the old bulb could be a key contributor to the build-up of residual forms of P in  
541 the soil over time.

542 The data collected during this study suggest that the bracken and bluebell plant community  
543 was able to dominate for two main reasons: phenology and different nutrient acquisition and  
544 storage strategies, as demonstrated by the different P storage in the plants and soil through  
545 litter input above and below ground. The concentration of P in bluebell above-ground parts  
546 was between 2 to 5 times higher compared to bracken. In addition, bluebells store P in form  
547 of *myo*-IP<sub>6</sub> in bulbs, possibly as a survival mechanism against P supply interruption during its  
548 growth cycle. The shed bulbs then might extend the P store outside its physiological limit into  
549 the surrounding soil, increasing the resilience for the population. The semi-natural system  
550 used for this study suggests an accumulation of organic P over time. We thus conclude that  
551 the build-up of soil P in the field is a result of the plants' biomass contributions over time  
552 particularly in the form of *myo*-IP<sub>6</sub>. These findings support observations on bluebell ecology  
553 in relation to being a woodland plant, or an indicator of ancient woodlands, and often  
554 appearing in clumps.

555

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557

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563

564

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**Table 1.** pH, organic matter (OM), base cations, Total P, organic P (Po) and total NaOH-EDTA extractable P and extraction efficiency in air-dried soil samples taken during the period when bluebells formed sole above-ground biomass (W1-W4) and biomass was equally distributed between bracken and bluebell above-ground (W5-W8).

| Soil sample                        | ← Bracken fronds emergence → |       |       |       |       |       |       |      | Mean <sup>§</sup> | SE <sup>§§</sup> |
|------------------------------------|------------------------------|-------|-------|-------|-------|-------|-------|------|-------------------|------------------|
|                                    | ← Bluebell flowering →       |       |       |       |       |       |       |      |                   |                  |
|                                    | W1                           | W2    | W3    | W4    | W5    | W6    | W7    | W8   |                   |                  |
| pH                                 | 4.7                          | 4.0   | 4.6   | 4.6   | 4.7   | 4.5   | 4.4   | 4.6  | 4.5               | 0.1              |
| OM (%)                             | 21.2                         | 28.7  | 28.6  | 23.9  | 23.9  | 32.9  | 37.1  | 26.1 | 27.8              | 1.7              |
| C (%)                              | 10.8                         | 16.8  | 14.4  | 16.5  | 14.7  | 13.9  | 20.2  | 12.6 | 15.0              | 2.7              |
| N (%)                              | 0.64                         | 0.86  | 0.72  | 0.78  | 0.73  | 0.76  | 0.95  | 0.70 | 0.77              | 0.09             |
| C/N                                | 16.8                         | 19.5  | 20.1  | 21.0  | 20.2  | 18.3  | 21.2  | 18.0 | 19.4              | 1.5              |
| C/P                                | 144                          | 192   | 167   | 212   | 137   | 208   | 200   | 139  | 175               | 28               |
| C/Po                               | 309                          | 361   | 341   | 407   | 296   | 333   | 366   | 255  | 333               | 41               |
| Ca (g kg <sup>-1</sup> )           | 0.29                         | 0.31  | 0.39  | 0.43  | 0.57  | 0.63  | 0.70  | 0.42 | 0.47              | 0.05             |
| Al (g kg <sup>-1</sup> )           | 11.6                         | 12.7  | 14.0  | 12.3  | 14.8  | 14.3  | 14.9  | 16.7 | 14.0              | 0.6              |
| Fe (g kg <sup>-1</sup> )           | 18.5                         | 18.7  | 19.0  | 20.7  | 23.8  | 22.2  | 21.8  | 22.7 | 22.1              | 0.7              |
| Total P (g kg <sup>-1</sup> )      | 0.75                         | 0.87  | 0.86  | 0.78  | 1.1   | 0.70  | 1.0   | 0.90 | 0.87              | 0.05             |
| Po (%)                             | 64                           | 77    | 78    | 70    | 71    | 98    | 75    | 84   | 77                | 3                |
| NaOH-EDTA P (g kg <sup>-1</sup> )  | 0.50                         | 0.70  | 0.61  | 0.61  | 0.71  | 0.53  | 0.81  | 0.65 | 0.64              | 0.03             |
| NaOH-EDTA Po (%)                   | 71                           | 66    | 70    | 66    | 70    | 79    | 69    | 76   | 71                | 4                |
| Extraction efficiency (%)          | 66                           | 81    | 71    | 79    | 66    | 79    | 80    | 72   | 74                | 2                |
| Mehlich-3 P (mg kg <sup>-1</sup> ) | 48.7                         | 39.6  | 27.8  | 31.3  | 22.1  | 42.7  | 29.0  | 21.4 | 32.8              | 9.2              |
| Mehlich-3 Al (g kg <sup>-1</sup> ) | 2.0                          | 2.3   | 2.2   | 2.0   | 2.2   | 2.0   | 2.1   | 2.0  | 2.1               | 0.1              |
| Mehlich-3 Fe (g kg <sup>-1</sup> ) | 0.26                         | 0.38  | 0.23  | 0.30  | 0.27  | 0.26  | 0.27  | 0.23 | 0.28              | 0.04             |
| PSR (Al + Fe)*                     | 0.022                        | 0.015 | 0.011 | 0.013 | 0.009 | 0.019 | 0.012 | 0.01 | 0.014             | 0.004            |

<sup>§</sup> Mean value of the results obtained for W1-W8.

<sup>§§</sup>SE is the standard error of measurements.

Total (Al, Ca and Fe) and Mehlich -3 (Al, Fe and P) values are average of  $n = 3$  (RSD $\leq$ 15).

\* PSR, phosphorus saturation ratio



**Table 2.** Total P ( $n = 3$ ,  $RSD \leq 10$ ) and NaOH-EDTA extractable P and extraction efficiency in dry plant samples divided in the different below and above-ground parts.

|                                    | Bluebell |       |        |        |       |         | Bracken  |        |        |
|------------------------------------|----------|-------|--------|--------|-------|---------|----------|--------|--------|
|                                    | Roots    | Bulbs | Scapes | Leaves | Seeds | Flowers | Rhizomes | Stipes | Blades |
| Total P ( $\text{g kg}^{-1}$ )     | 2.7      | 0.84  | 2.9    | 3.5    | 3.3   | 4.0     | 0.14     | 1.5    | 2.8    |
| NaOH-EDTA P ( $\text{g kg}^{-1}$ ) | 2.2      | 0.72  | 2.5    | 3.2    | 3.2   | 3.1     | 0.15     | 1.6    | 2.7    |
| Extraction efficiency (%)          | 81       | 86    | 86     | 91     | 99    | 77      | 107      | 107    | 96     |

**Table 3.** Relative amount (%) of the major P forms detected in the soil, bluebell and bracken plant samples.

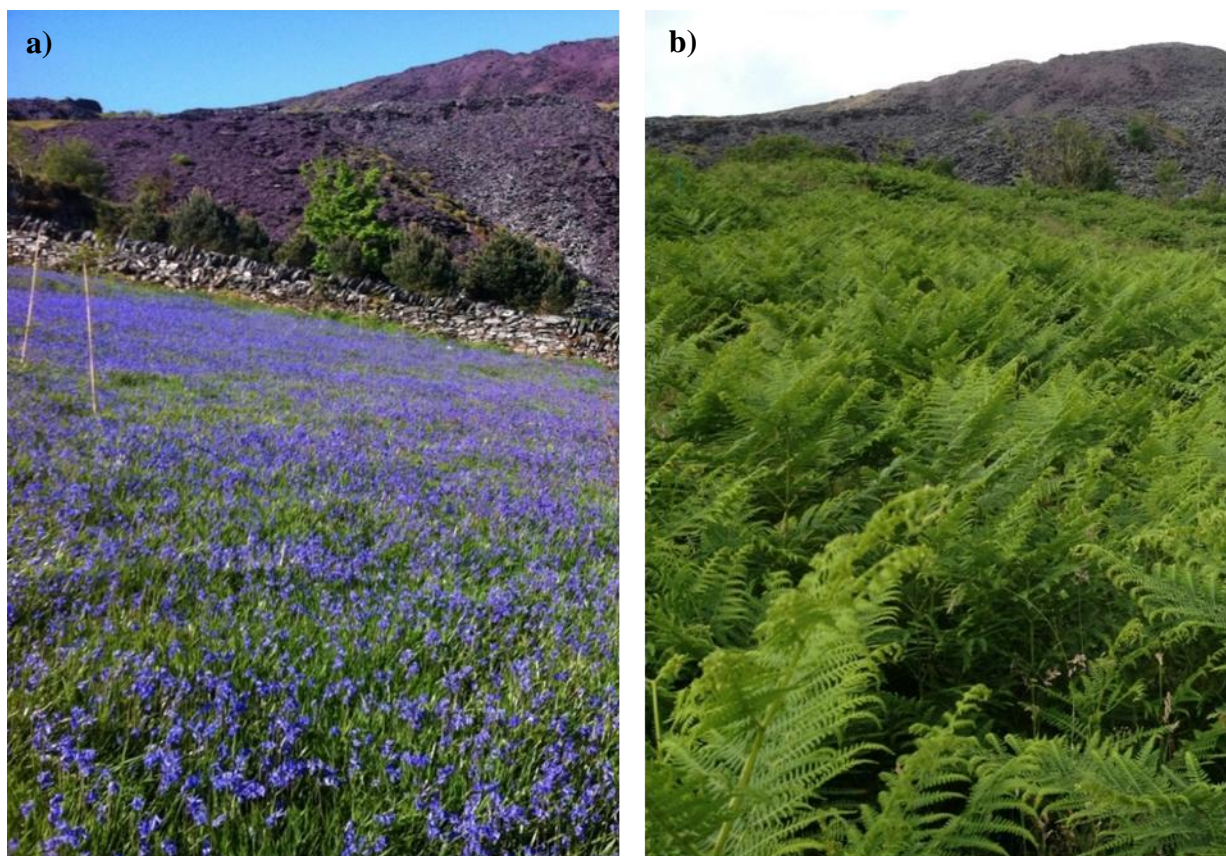
|                 |         | Inorganic P   |             |             |                                 |                                    |                   |                  | Organic P   |               |             |                   |                     |                  |
|-----------------|---------|---------------|-------------|-------------|---------------------------------|------------------------------------|-------------------|------------------|-------------|---------------|-------------|-------------------|---------------------|------------------|
|                 |         | Ortho<br>P    | Pyro<br>P   | Poly<br>P   | <i>myo</i> -<br>IP <sub>6</sub> | <i>scyllo</i> -<br>IP <sub>6</sub> | $\alpha$ -<br>gly | $\beta$ -<br>gly | AMP         | Other<br>mono | DNA         | Other<br>diesters | Phospho<br>nolipids | Phospho<br>nates |
| <i>Soil</i>     | W1-W8   | 20.0-<br>31.5 | 0.6-<br>2.5 | 0.4-<br>7.0 | 26.3-<br>40.0                   | 12.3-<br>17.5                      | 0.4-<br>2.4       | 0.8-<br>4.4      | 0.7-<br>5.3 | 8.4-16        | 1.1-<br>2.5 | 0.9-2.7           | 0.8-3.6             | 0.5-2            |
| <i>Bluebell</i> | Roots   | 63.2          | -           | -           | -                               | -                                  | 14.4              | 6.0              | 6.4         | 10.0          | -           | -                 | -                   | -                |
|                 | Bulbs   | 42.0          | -           | -           | 39.4                            | -                                  | 9.4               | 3.6              | 5.5         | -             | -           | -                 | -                   | -                |
|                 | Seeds   | 21.7          | 1.5         | -           | 60                              | -                                  | 4.5               | 7.7              | 1.4         | 2.8           | -           | 0.5               | -                   | -                |
|                 | Scapes  | 75.5          | -           | -           | -                               | -                                  | 8.1               | 0.9              | 4.1         | 10.1          | -           | 1.3               | -                   | -                |
|                 | Leaves  | 80.4          | -           | -           | -                               | -                                  | 6.8               | 2.9              | 3.5         | 4.0           | -           | 2.3               | -                   | -                |
|                 | Flowers | 70.4          | -           | -           | -                               | -                                  | 11.2              | 0.9              | 3.0         | 11.3          | 1.8         | 1.4               | -                   | -                |
| <i>Bracken</i>  | Rhizome | 68.5          | -           | -           | -                               | -                                  | -                 | -                | -           | 31.5          | -           | -                 | -                   | -                |
|                 | Stipes  | 74.1          | -           | -           | -                               | -                                  | 5.5               | 9.6              | 1.8         | 8.9           | -           | -                 | -                   | -                |
|                 | Blades  | 81.1          | -           | -           | -                               | -                                  | 2.5               | 4.4              | 2.0         | 8.8           | -           | 1.1               | -                   | -                |

**Table 3b.** Absolute amount (mg kg<sup>-1</sup>) of the major P forms detected in the soil, bluebell and bracken plant samples.

|                 |         | Inorganic P |          |          |                             |                                | Organic P     |              |          |            |          |                |                  |               |
|-----------------|---------|-------------|----------|----------|-----------------------------|--------------------------------|---------------|--------------|----------|------------|----------|----------------|------------------|---------------|
|                 |         | OrthoP      | Pyro P   | Poly P   | <i>myo</i> -IP <sub>6</sub> | <i>scyllo</i> -IP <sub>6</sub> | $\alpha$ -gly | $\beta$ -gly | AMP      | Other mono | DNA      | Other diesters | Phospho nolipids | Phospho nates |
| <i>Soil</i>     | W1-W8   | 105.1-131.0 | 3.2-17.6 | 2.1-48.6 | 159.4-259.1                 | 77.5-114.3                     | 4.4-14.7      | 4.9-27.4     | 4.3-32.5 | 50.2-113.5 | 6.9-15.3 | 5.5-21.7       | 4.2-21.9         | 3.0-10.6      |
| <i>Bluebell</i> | Roots   | 1390.4      | -        | -        | -                           | -                              | 316.8         | 132.0        | 140.8    | 220.0      | -        | -              | -                | -             |
|                 | Bulbs   | 302.4       | -        | -        | 283.7                       | -                              | 67.7          | 25.9         | 39.6     |            |          |                | -                | -             |
|                 | Seeds   | 701.4       | 48.5     |          | 1939                        |                                | 145.4         | 248.9        | 45.2     | 90.5       |          | 16.2           |                  |               |
|                 | Scapes  | 1887.5      | -        | -        | -                           | -                              | 202.5         | 22.5         | 102.5    | 252.5      |          | 32.5           | -                | -             |
|                 | Leaves  | 2572.8      | -        | -        | -                           | -                              | 217.6         | 92.8         | 112.0    | 128.0      |          | 73.6           | -                | -             |
|                 | Flowers | 2182.4      | -        | -        | -                           | -                              | 347.2         | 27.9         | 93.0     | 350.3      | 55.8     | 43.4           | -                | -             |
| <i>Bracken</i>  | Rhizome | 102.8       | -        | -        | -                           | -                              | -             | -            | -        | 47.3       | -        | -              | -                | -             |
|                 | Stipes  | 1185.6      | -        | -        | -                           | -                              | 88.0          | 153.6        | 28.8     | 142.4      | -        | -              | -                | -             |
|                 | Blades  | 2189.7      | -        | -        | -                           | -                              | 67.5          | 118.8        | 54.0     | 237.6      | -        | 29.7           | -                | -             |

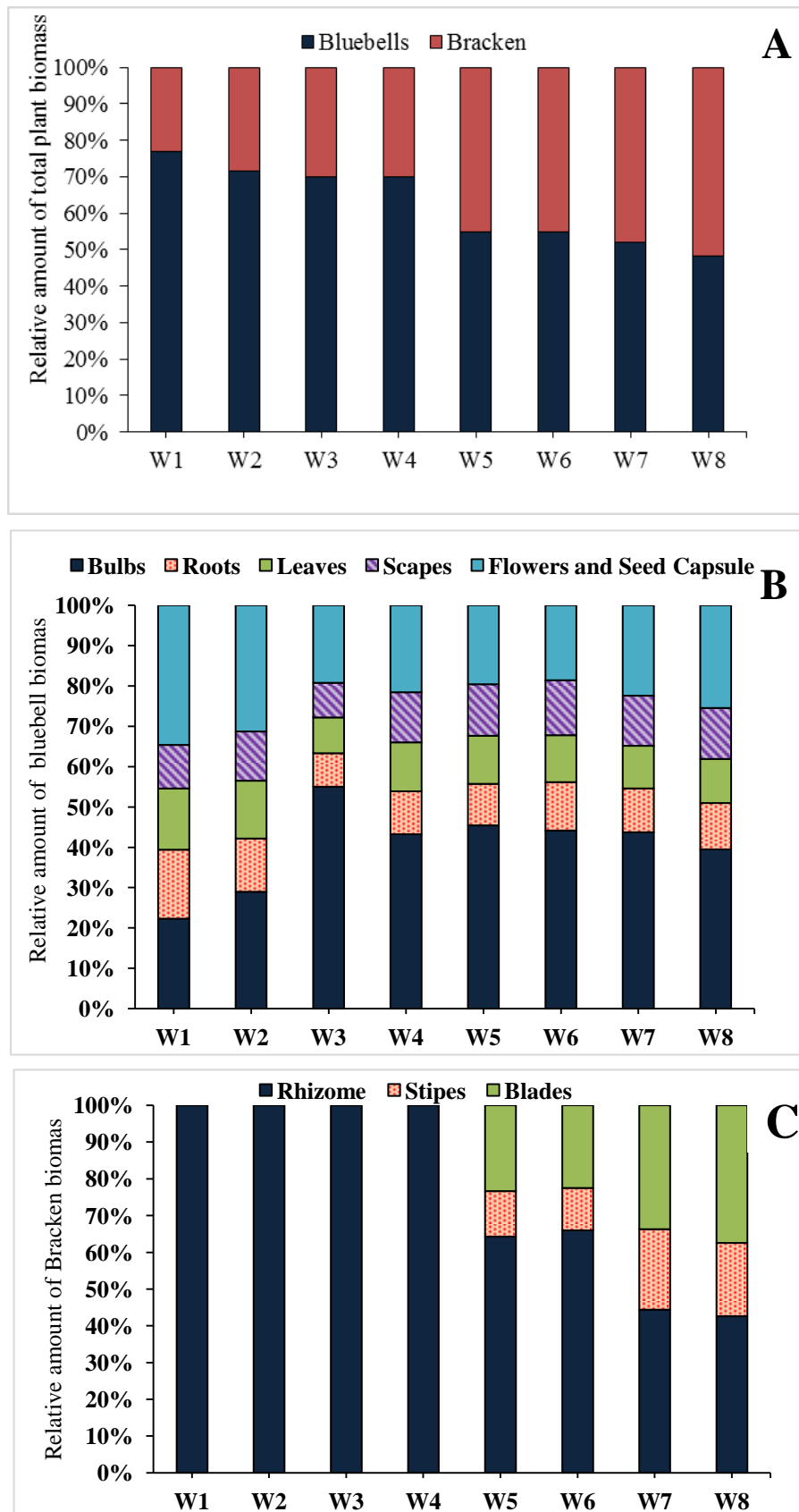
## Figure

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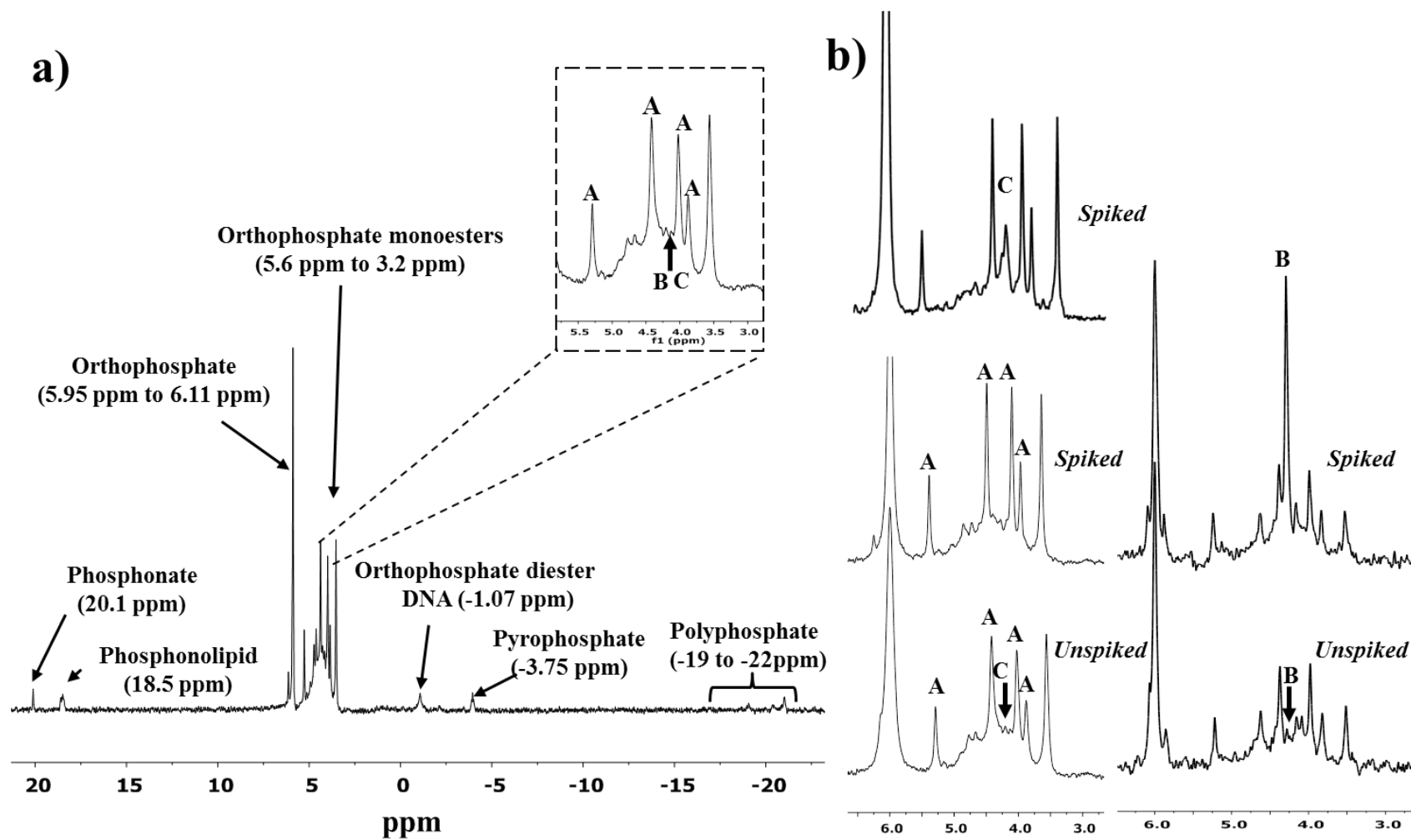


**Figure 1:** Field site shown with a) bluebell dominance in mid spring and b) bracken dominance in mid summer

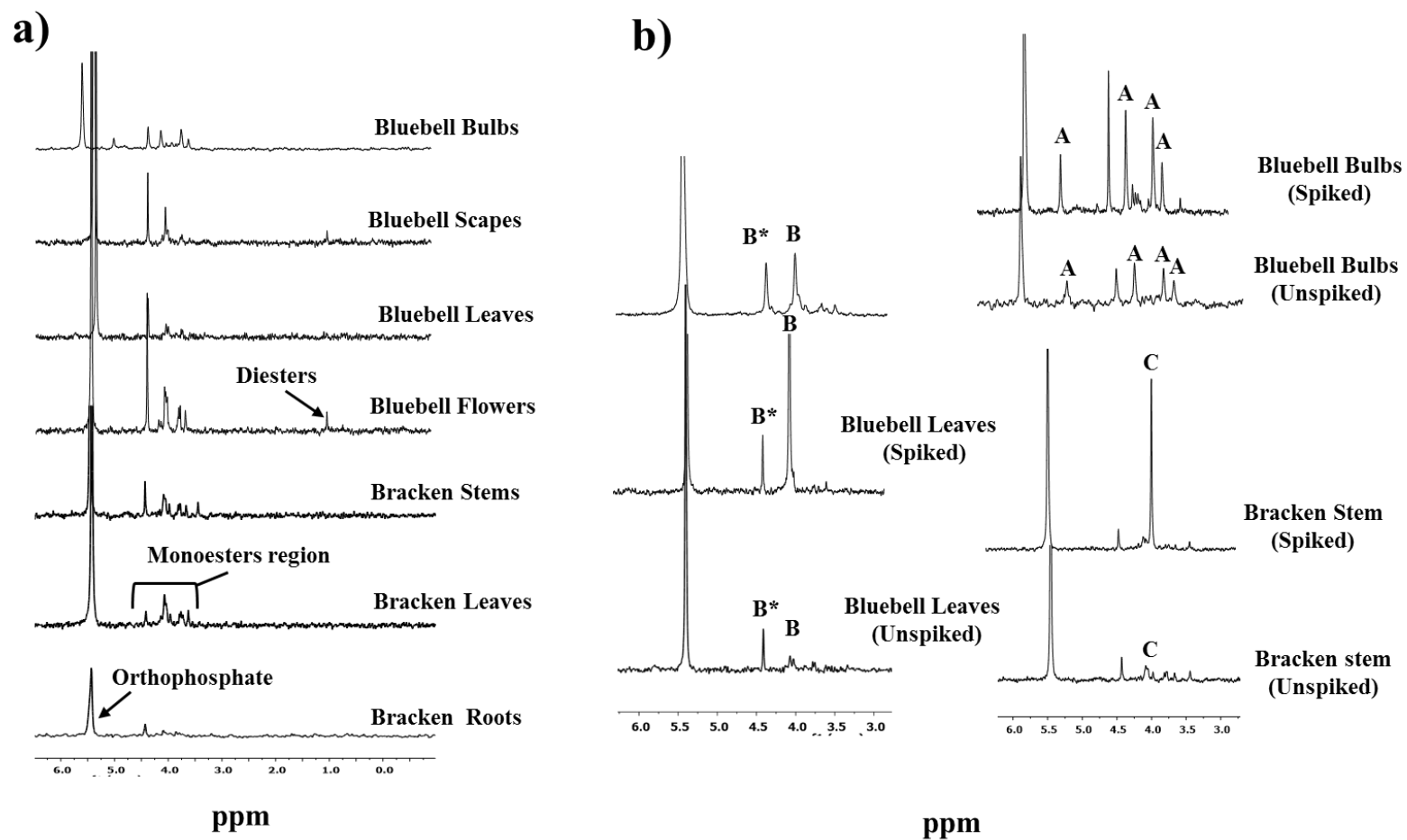
**Figure 2.** Percentage of biomass contribution above and below ground on dry weight basis of bluebell and bracken plants (A) and bluebell (B) and bracken (C) plant parts. W1 to W8 refers to the sampling weeks from 7<sup>th</sup> May to 25<sup>th</sup> June.



**Figure 3.** a) Solution  $^{31}\text{P}$  NMR spectra of a representative composite soil sample and b)  $^{31}\text{P}$  NMR unspiked and spiked spectra for the identification of *myo*-IP<sub>6</sub> (A),  $\beta$ -glycerophosphate (B), adenosine 5 monophosphate AMP (C).



**Figure 4.** a) Solution  $^{31}\text{P}$  NMR of bluebell and bracken plants parts and b)  $^{31}\text{P}$  NMR unspiked and spiked spectra for the identification of *myo*-IP<sub>6</sub> (A),  $\alpha$ -glycerophosphate (B\*),  $\beta$ -glycerophosphate (B), adenosine 5 monophosphate AMP (C).



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