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## N-Carbamate protected amino acid derived guanidine organocatalysts.

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#### Abstract

We report the preparation of a range of N-protected amino acid derived guanidine organocatalysts and their application to the Michael addition of 2-hydroxy-1,4-napthoquinone to  $\beta$ -nitrostyrene, achieving a maximum ee of 26%. Whilst these catalysts gave poor ees, the structural variation together with the X-ray crystallographic study of the intra- and intermolecular hydrogen bonding reported suggest that the C<sub>2</sub>-symmetric catalyst are lead compounds for the further development of this methodology.

#### Keywords

Organocatalysis, amino acids, guanidines, H-bonded extended networks.

#### Introduction

We recently reported<sup>1</sup> the preparation of a range of amino acid derived guanidines which were shown to have some potential as asymmetric organocatalysts however several problems were apparent from the work. This was supported by an investigation in the solid state demonstrating extensive intra- and intermolecular H-bonding abilities of the proline, guanidine and/or amide functional groups within these organic moieties. The best of these catalysts 1 was found to catalyse the formation of the Michael adduct 4 in 56% ee and 49% yield from the addition of 2-hydroxy-1,4-napthoquinone 2 to  $\beta$ -nitrostyrene 3. (Scheme 1)

Scheme 1: Catalytic Michael addition leading to 4: Catalyst (0.1 equiv.), -78 °C, 5 h then -20 °C 48h. 49%, 56% ee.

A major problem observed with this reaction were related to the integrity of the catalyst 1 which was formed from the CDI mediated coupling of *N*-Me-L-proliner with *N*-Cbz-guanidine in DMF. It was found that during the coupling the intermediate

imidazole amide was prone to racemisation in the presence of even weakly basic guanidine reactants or the internal *N*-methyl functionality. In order to alleviate this problem and to investigate the role of the guanidine in the catalytic process we considered removing the internal amine base. We visualised a group of potential catalyst structures **5a,b** and **6a,b** derived from L-proline and **7a-h** and **8a-d** derived from L-alanine or L-phenylalanine. In these catalysts the nitrogen is protected by a Boc or Cbz-protecting group as this will suppress racemisation.<sup>2</sup> One problem that might arise is a lowering of the basicity of the catalyst system which will lead to slower rates of conversion of the base catalysed reaction. (Figure 1)

Fig. 1: Proposed catalysts; PG = Boc, Cbz; R = Me, Bn.

#### Preparation of catalysts

The required N-protected amino acids and guanidines were available commercially or were prepared by literature methods.<sup>3,4</sup> Catalysts 5a,b were prepared (Scheme 2) by activating the required N-Cbz- 9a or N-Boc-L-proline 9b with CDI in dry DMF at 0 °C, followed by the addition of either N-Cbz-guanidine 10a or N-Bocguanidine 10b stirring for 24 h (Table 1, entries 1 and 2). Work up and chromatography gave the desired compounds, 5a and 5b, in 40 and 67% yield respectively. The C2-symmetric catalyst **6b** was prepared by similarly activating 2.0 equiv. of 9b and adding this solution via cannula to a DMF solution of guanidine, generated from guanidinium chloride and sodium hydride. (Table 1 entry 4) Work up and chromatography gave the 6b in 97% yield. We attempted the preparation of the corresponding Cbz-catalst 6a from 9a using this method unfortunately despite numerous attempts we were unable to prepare this as the catalyst was inseparable from reaction by-products due to its high polarity (Table 1, entry 3). Catalysts 7a-d (R = Me) and 7e-h (R = Bn) were prepared from either N-Boc-L-alanine 11a, N-Cbz-L-alanine 11b, N-Boc-Lphenylalanine 11c or N-Cbz-L-phenylalanine 11c in 19-84% yield, however heating to 40 °C for 48 h was required to effect reaction in the case of 7a and 7h. (Table 1, entry 5-12). The catalysts 8a,b (R = Me) and 8c,d (R = Bn) were similarly prepared in 49-80% yield after work-up and purification (Table 1, entries 13-16).

Scheme 2: Preparation of catlysts 5-8 (a) METHOD A: i) 9a (Pg = Cbz), 9b (Pg = Boc), 11a (Pg = Boc, R = Me), 11b (Pg = Cbz, R = Me), 11c (Pg = Boc, R = Bn) or 11d (Pg = Cbz, R = Bn), CDI, DMF. ii) N-Cbz-guanidine 10a, or N-Boc-guanidine 10b (1.0 equiv.), rt, 24-72 h. (b) METHOD B: i) 9a (Pg = Cbz) or 9b (Pg = Boc), CDI, DMF; ii) Guanidine HCl (0.5 equiv., NaH, DMF; iii) Combine and stir, 24-72 h, rt.

Table 1: Preparation of catalysts 5-8.

Entry	Catalyst	Yield	Method - SM	R	N-Pg	G-Pg
1	5a	67	A - 9a	-	Cbz	Cbz
2	5b	40	A - 9b	-	Boc	Cbz
3	6a	0	B - 9a	-	Cbz	-
4	6b	97	B - 9b	-	Boc	-
5	7a	63	A - 11a	Me	Boc	Boc
6	7b	71	A - 11a	Me	Boc	Cbz
7	7c	49	A - 11b	Me	Cbz	Boc
8	7d	75	A - 11b	Me	Cbz	Cbz
9	7e	44	A - 11c	Bn	Boc	Boc
10	7f	72	A - 11c	Bn	Boc	Cbz
11	7g	84	A - 11d	Bn	Cbz	Boc
12	7h	19	A - 11d	Bn	Cbz	Cbz
13	8a	65	A - 11a	Me	Boc	-
14	8b	55	A - 11b	Me	Cbz	-
15	8c	80	A - 11c	Bn	Boc	-
16	8d	49	A - 11d	Bn	Cbz	-

(i) See scheme 1. (ii) N-Pg = amino acid protection G-Pg = guanidine protection.

### Catalytic studies

Initially the *N*-proline derived catalysts **5a**, **5b** and **6** catalysts were applied to the Michael reaction to give adduct **4**. (Scheme 3, Table 2, entries 1-3). The first unsurprising observation is that the yields for these catalytic reactions were all low and the reaction times relatively long which must arise from the low basicity of these catalysts. The *N*-Boc-proline catalyst **5b** gave very poor ee's (1-5%) over the range of four solvents studied, whilst the *N*-Cbz-proline catalyst **5a** gave generally better 8-18% ees but slower reactions. The *C*<sub>2</sub>-symmetric catalyst **6b** gave better 12-22% ees and benzene and toluene appeared to be the best solvents (both gave 22% ee). The L-alanine and L-phenylalanine derived catalysts **7a-h** were then investigated and again the reactions using these catalysts (entries 4-11) were slow, requiring several days to reach a reasonable yield. Where the reactions were relatively quick and higher yielding (2-8 days; entries 4, 10 and 11) the ee's were very

poor. The best ees were observed with 7b (entry 5) and 7c (entry 6), which gave 16% and 14% ee in toluene respectively, but the reactions were very slow and yields poor. The C2-symmetric catalysts 8a-d, were next utilised in the Michael reaction to form 4 (entries 12-17). The N-Boc-L-alanine catalyst 8a effected the reaction in a high yield over 9 days (entry 12) however the ee observed for either solvents was poor. The corresponding N-Boc-Lphenylalanine catalyst 8c was however completely inactive in this reaction (entry 15) which we initially thought might be a consequence of high steric hindrance. Repeating the reaction using benzoic acid as a co-catalyst (entry 16) led to slow conversion to the product but the ees in both solvents were low. Surprisingly the assumed less sterically hindered N-Cbz-L-alanine catalyst 8b was also completely inactive in this reaction (entry 13). This might suggest that steric factors are not the primary consideration in these reactions and that intramolecular H-bonding plays a significant role. Similarly repeating the reaction using benzoic acid as a cocatalyst (entry 14) led to slow conversion to the product but the ees in both solvents were again low. Finaly the N-Cbz-L-phenylalanine catalyst 8d was studied (entry 17) and this was found to be the best of these catalysts leading to a slow conversion to the product in poor yield but with better ees in dichloromethane (26%) and toluene (15%).

$$\begin{array}{c} O \\ O \\ O \\ O \\ \end{array}$$

$$\begin{array}{c} O \\ O \\$$

Scheme 3: Catalytic studies: Conditions: See Table 10, Catalyst (0.1 equiv.), 0 °C 7-8 h then rt.

Table 2: Catalytic studies(i)

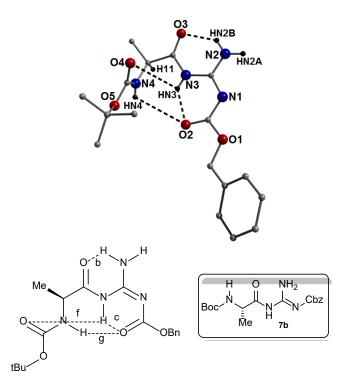
Entry	Cat.	CH <sub>2</sub> Cl <sub>2</sub>	PhMe	Xylene	PhH
1	5a	18 (5/55)	13 (5/31)	8 (5/32)	8 (2/30)
2	5b	5 (1/38)	4 (1/22)	1 (3/30)	4 (1/25)
3	6b	<b>18</b> (1/38)	<b>22</b> (48/30)	12 (23/41)	<b>22</b> (48/25)
4	7a	2 (6/49)	2 (7/81)		
5	7b	6 (18/47)	<b>14</b> (12/11)		
6	7c	12 (10/37)	<b>16</b> (11/37)		
7	7d	6 (10/51)	<b>10</b> (10/24)		
8	7e	5 (9/67)	0 (9/84)		
9	7 <b>f</b>	6 (17/68)	<b>12</b> (11/18)		
10	7g	3 (8/74)	2 (8/73)		
11	7h	5 (3/84)	0 (2/56)		
12	8a	3 (9/85)	3 (9/70)		
13	8b	NR	NR		
14	8b <sup>(ii)</sup>	10 (24/95)	6 (11/70)		
15	8c	NR	NR		
16	8c <sup>(ii)</sup>	5 (24/48)	8 (9/32)	_	
17	8d	<b>26</b> (22/22)	<b>15</b> (12/46)		

(i) Results are given as ee (time (d)/yield (%)) (ii) Benzoic acid (0.1 equiv.) was added

### 2.11 X-ray and NMR.

Analysis of the NMR data for the catalysts in all cases gave evidence of the presence of a rotamers in solution. In particular the catalysts with a Boc-carbamate protecting group gave evidence of multiple species in solution with multiple signals present. We attempted to crystallisation all of the catalysts from a variety of solvents and in the majority of cases only amorphous solids were obtained. Only compound 7b and 7f gave crystals suitable for X-

ray analysis. Pleasingly neither of these appeared to be racemised in the crystal form which supports our premise of using Nprotected amino acids. Catalyst 7b, crystallises in the P2<sub>1</sub> space group and comprises two complete molecules within the asymmetric unit. Each dimer is bound through a strong and highly directional intermolecular H-bond between an amide proton and a nearby carbonyl O-donor atom (N4(HN4)···O9 = 1.95 Å), as well as a number of aliphatic C-H···O interactions (e.g. C34(H34A)···O2 = 2.88 Å,  $C16(H16A) \cdot \cdot \cdot O6 = <math>2.87 \text{ Å}$  and  $C12(H12A) \cdot \cdot \cdot O8 = <math>2.69$ Å) (Fig. S1 in appendix 1). Catalyst 7b is derived from L-alanine, which has a E-amide arrangement with the carbonyl of the amide having a E-bc type H-bonding type1 (bond b; N2(HN2B)···O3 = 1.98 Å and bond c; N3(HN3)···O2 = 1.97 Å). Additionally an Hbond was observed between the guanidine NH and the Bocprotecting group (bond f; N3(HN3)···O4 = 3.04 Å) and possibly a weaker H-bond between the Boc-protecting group NH and the Cbz-protecting group carbonyl (bond g; N4(HN4)···O2 = 3.40 Å). This is an *E*-amide-bcf(g) pattern (Figure 2).



**Figure 2:** The X-ray crystal structure and corresponding ChemDraw representations of **7b**. The dashed lines represent intramolecular H-bonding at distances: (b) N2(HN2B)···O3 = 1.98 Å; (c)N3(HN3)···O2 = 1.97 Å along with the long-contacts (f) N3(HN3)···O4 = 3.04 Å and (g) N4(HN4)···O2 = 3.40 Å. The stereogenic proton has been labelled (H11).

Organocatalyst **7f** crystallises in the P1 space group and akin to **7b**, its asymmetric unit comprises two complete organocatalyst moieties (Fig. S3). In both molecules, extensive intramolecular H-bonds (e.g. N2(HN2A)···O2 = 1.97 Å, N4(HN4)···O5 = 2.27 Å and N6(HN6B)···O8 = 2.13 Å) and C-H··· $\pi$  interactions (e.g. [C18-C23]<sub>centroid</sub>···(H16C)C16 = 4.10 Å and [C41-C46]<sub>centroid</sub>···(H38B)C36 = 4.20 Å) are observed (Figure 4). On close examination of the structure of compound **7f**, it again possessed an *E*-amide arrangement but with the amide N-H not involved in any intramolecular H-bonds to the carbamate protecting group of the guanidine. Instead there was a long distance

interaction between the Boc-carbonyl and the amide N-H (bond f; N3(HN3)···O4 = 3.49 Å) was observed. The guanidine NH<sub>2</sub> had two H-bonds of the *E*-bd type (bond b; N2(HN2B)···O3 = 2.17 Å and bond d; N2(HN2A)···O2 = 1.97 Å) and a H-bond between the NH of the Boc protecting group and the amide oxygen was also observed (bond e; N4(HN4)···O3 = 3.12 Å). Overall this is an *E*-amide-bde(f) pattern (Figure 3).

As highlighted in Figure 4, the crystallographically equivalent organocatalyst units in 7f are connected via a number of intermolecular H-bonding interactions, resulting in planar sheets that propagate along the a unit cell direction (Figure 5b). In order to obtain packing efficiency, these individual H-bonded 2D sheets arrange in a parallel and interdigitated motif along the b unit cell direction and are held together by intermolecular C-H··· $\pi$  and C-H···O interactions at distance of 3.80 Å ([C41-C46]<sub>centroid</sub>···(H7B)C7) and 2.54 Å (C23(H23)···O7), respectively. This packing efficiency is best shown in Figure 5d, whereby each H-bonded sheet is represented in space-fill mode and using a different colour.

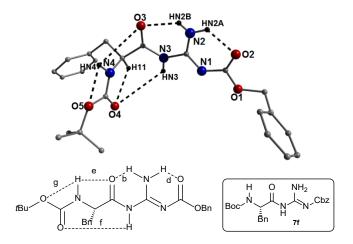
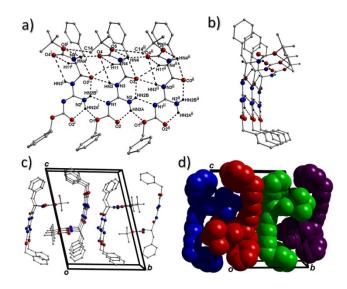


Figure 3: The X-ray crystal structure and ChemDraw representations of 7f. The dashes lines represent the H-bonds: (d) N2(HN2A)···O2 = 1.97 Å; (b) N2(HN2B)···O3 = 2.17 Å; (e) N4(HN4)···O3 = 3.12 Å; (f) N3(HN3)···O4 = 3.49 Å (long contact); (g) N4(HN4)···O5 = 2.27 Å and C11(H11)···O4 = 2.42 Å. The chiral proton has been labelled (H11).



**Figure 4** A planar 2-D sheet comprising on individual **7f** units connected through numerous intermolecular hydrogen bonds (dashed lines) as viewed perpendicular (a) and parallel to the plane.

Intermolecular H-bond distances: N2(HN2A)<sup>i</sup>···O1 = 2.09 Å; N2(HN2B)<sup>ii</sup>···N1 = 2.35 Å; N3(HN3)<sup>ii</sup>···O3 = 2.37 Å; C11(H11)<sup>ii</sup>···O3 = 2.93 Å; N4(HN4)<sup>ii</sup>···O4 = 2.14 Å and C14(H14B)<sup>ii</sup>···O5 = 3.02 Å. Symmetry code (i) = (-1+x, y, z); (ii) = (1+x, y, z). (c) Packing arrangement observed in 7f as viewed along the *a* unit cell direction along with its space-fill represented version (d). Each colour represents an individual 2D hydrogen bonded sheet in 7f (as shown in a).

Whilst X-ray conformations are not a measure of conformation in solution it was interesting to note that the two structures are structurally similar in that the carbamates are identical and the only differences are the amino acid R-group (Me versus Bn). The  $^1H$  spectra of **7b** and **7f** in CDCl<sub>3</sub> displays a mixture of rotamers in both cases and the guanidine NH-protons are broadened. The corresponding  $^1H$  spectra in DMSO-d6 however gave separate and relatively sharp signals for each NH of the guanidines of **7b** ( $\delta_H$  = 11.30/8.90/8.86 (3 x 1H, 3 x br s)) and **7f** ( $\delta_H$  = 11.36/8.94/8.85 (3 x 1H, 3 x br s)). This might suggest the two catalysts which give reasonable ees in the catalystic reaction possess similar structures in solution.

#### **Conclusions**

The overall conclusion from the reactions of the N-protected amino acids is that these processes are typically slower than the corresponding N-alkyl catalysts and that there is no appreciable increase in ee's. There is strong apparent correlation between the different general types of the catalysts however, two of the better catalysts of this class were the C2-symmetric examples 6 (18-22% ee over three solvents) and 8d (15-26% ee over two solvents), which gave encouraging results. Unfortunately, we could not obtain suitable crystals of either of these catalysts to investigate Hbonding patterns present and no real rationale for the higher ee's observed with these catalysts can be offered. However steric factors do not seem to be paramount and this possibly points to intramolecular hydrogen bonding being a key to developing an effective system. In addition, the level of disorder of these compounds in solution is apparent from NMR studies and it has been reported that the presence of multiple catalytic structures in solution leads to lower ees in reactions of related guanidines.<sup>5</sup> Our overall goals in this research was to develop a catalyst which takes part in strong associative interactions with the reactants in this process and this study has focused on one reaction. The modifications made to the catalyst structures from our previous work<sup>1</sup> have not lead to any improvement in ee and it is apparent that the H-bonding patterns observed are not predictable. This might suggest that the ability of the guanidine to form multiple strong H-bonds is not a favorable one and our original goal<sup>1</sup> to employ a simplified range of base catalysts might be more advantageous.<sup>6</sup> Interestingly, Liu et al. reported<sup>7</sup> the use of chiral bi-functional guanidines 12 to catalyse the aza-Henry reactions of isatin derived ketimines 13 with nitromethane. The authors put forward a model where the strong intramolecular N-H bond (2.260 Å; Scheme 4) is broken by the deprotonation of nitromethane. When this intramolecular hydrogen bonding is removed by this chelation, the amide groups is thought to act as a Brønsted acid to activate ketimine 13 eventually leading to the formation of the product 14. (Scheme 4) Obviously this report infers that changes in the H-bonding pattern are a consideration and targeting systems with weaker H-bonding interaction in the solid state might be a factor in the success of these reactions. We will report further findings in this area in due course.†

**Scheme 4:** (a) i) **12** (10 mol %), **13**, MeNO<sub>2</sub>, PhMe, -30 °C, 72 h. R<sup>1</sup> = Me, Bn; R<sup>2</sup> = Boc, EtO<sub>2</sub>C; R<sup>3</sup> = H, F, Cl, Br, I, Me, F<sub>3</sub>CO, CF<sub>3</sub>.

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#### **General Procedures:**

Unless otherwise noted, reactions were stirred and monitored by TLC. TLC plates were visualized using iodine, phosphomolybdic acid or under UV light. All anhydrous reactions were conducted under a static argon atmosphere using oven dried glassware that had previously been cooled under a constant stream of nitrogen. Reagents, dry solvents and starting materials (N-Boc-L-proline 9a, *N*-Boc-L-alanine *N*-Cbz-L-alanine 11a, 11b. N-Boc-Lphenylalanine 11c and N-Cbz-L-phenylalanine 11d) were purchased from commercial suppliers and used without further purification. Other starting materials (N-Cbz-L-proline 11b,<sup>3</sup> N-Cbz-guanidine 10a<sup>4</sup> and N-Boc-guanidine 10b<sup>4</sup>) were prepared according to literature procedures. Flash column chromatography was performed on Davisil® silica gel (35-70 microns) with the eluent specified in each case, TLC was conducted on precoated E. Merck silica gel 60 F<sub>254</sub> glass plates. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on a Bruker Avance 400 spectrometer with an internal deuterium lock at ambient temperature at 400/100 MHz with internal references of  $\delta_H$  7.26 and  $\delta_C$  77.016 ppm for CDCl<sub>3</sub>,  $\delta_H$ 3.31 and  $\delta_{\rm C}$  49.0 ppm for CD<sub>3</sub>OD. Infrared spectra were recorded on a Bruker Tensor 37 FT-IR. Mass spectra were determined on a Q Exactive Plus (Thermo Scientific) instrument run with positive electrospray ionization (ESI). Melting points were determined on a Stuart SMP10 apparatus and are uncorrected. Optical rotations were measured in a 0.25 dm cell on an ADP440 polarimeter (Bellinghan & Stanley Ltd.).

### **Experimental**

#### Crystallography

Data for **7b** and **7f** were collected at 100K on a Rigaku AFC11 quarter chi goniometer equipped with a Rigaku Hypix 6000 detector mounted at the window of 007 HF copper rotating anode generator with Varimax optics. Cell determination, Data collection, Data reduction and Absorption correction were performed with CrysAlisPro (Rigaku Oxford Diffraction, 2017, version 1.171.39.31c). Structures were solved using SHELXT<sup>8</sup> and refined with SHELXL-2014/7 within the Olex2<sup>9</sup> software suite. Non-

hydrogen atoms were refined anisotropically whilst hydrogen atoms were included isotropically in idealised positions based on the parent atom. Further data collection and refinement details can be found in Appendix 1 and full data, including structure factors, have been deposited with the Cambridge Crystallographic Data Centre (Deposition numbers CCDC2030901 and CCDC2030902 for 7b and 7f respectively).

#### General Methods for the Preparation of catalysts.

**Method** A: The *N*-protected amino acid (1.0-1.6 equiv.) was dissolved in DMF (1-2.5 mL per mmol), CDI (1.2 equiv.) was added and the mixture stirred for 5 min to 24 h. After cooling (0 °C) the mixture, the required guanidine (1.0 equiv.) was added as a solid and the mixture stirred to rt over 16-168 h. After evaporation under reduced pressure (freeze dryer) or dilution with water and extracting with ethyl acetate. After drying (MgSO<sub>4</sub>), filtering and evaporation under reduced pressure, the product was obtained and was purified by column chromatography (Et<sub>2</sub>O/petroleum ether or EtOAc/petroleum ether), recrystallization or tritutation.

**Method B**: (*C*<sub>2</sub>-catalysts): The *N*-alkyl-L-amino acid (1 equiv.) and CDI (1.20 equiv.) were added sequentially to dry DMF (0.5-2.5 mL per mmol) and the mixture stirred for 1-16 h. In a separate flask, NaH (0.60 equiv.) was suspended in dry DMF (1.0-2.0 mL per mmol) and dried (P<sub>2</sub>O<sub>5</sub>) guanidinium chloride (0.50 equiv.) was added. After stirring for 1 h the activated amino acid solution was transferred into this flask via cannula and the mixture stirred for 24-48 h. The mixture was diluted with water (100 mL) and EtOAc (100 mL), separated and the aqueous layer extracted with further EtOAc (2 × 100 mL) and the combined extracts washed with water (2 × 100 mL). After drying (MgSO<sub>4</sub>), filtering and evaporation under reduced pressure the residue was co-evaporated with heptane to remove residual DMF and purified by silica gel chromatography (EtOAc in petroleum ether).

### Benzyl (S)-2-((N-

#### ((benzyloxy)carbonyl)carbamimidoyl)carbamoyl)pyrrolidine-1- Carboxylate 5a.

Method A: N-Cbz-L-proline 9a (1.51 g, 6.07 mmol, 1.0 equiv.); DMF (10 mL), 0 °C; CDI (1.27 g, 7.83 mmol, 1.3 equiv.); 24 h; N-Cbz-guanidine 10a (1.17 g, 6.07 mmol, 1.0 equiv.); 2 d, rt. Extraction with EtOAc; washed with water (3 × 150 mL) and brine (3 × 50 mL); column chromatography (0-30% EtOAc in petroleum ether) gave 5a (1.73 g, 4.08 mmol) in 67% yield as a white solid. **R**<sub>f</sub> 0.49 (EtOAc); **Mp** 47-49 °C;  $[\alpha]_D^{20}$  -60.1 (CHCl<sub>3</sub>, c = 2.0);  $\delta_H$ (CDCl<sub>3</sub>) (mixture of rotamers) 7.82-9.96 (3H, br s, 3 × NH), 7.16-7.45 (10H, m, 2 × Ph), 5.19 (1H, br d, J 11.7 Hz, CH), 5.14 (1H, d, J 12.5 Hz, CH), 5.10 (1H, d, J 12.4 Hz, CH), 5.02 (1H, br d, J 11.7 Hz, CH), 4.24-4.47 (1H, m, CH), 3.37-3.65 (2H, m, CH<sub>2</sub>), 1.98-2.32 (2H, m, CH<sub>2</sub>), 1.81-1.97 (2H, m, CH<sub>2</sub>); δ<sub>C</sub> (CDCl<sub>3</sub>) (mixture of rotamers) 176, 162.7/162.1, 158.9/158.8, 155.6/154.4, 136.4/136.3/136.2 (2 × C), 128.5/128.1/128.1/128.0 (10 × CH), 67.6/67.4, 67.1, 61.7/61.4, 47.3/47.0, 31.2/29.6, 24.4/23.6;  $\nu_{max}$ 3383, 3274, 3032, 2957, 2886, 1703, 1662, 1627, 1539, 1498, 1446, 1414, 1379, 1355, 1277, 1170, 1116, 1090, 1026, 989, 914, 806, 748 cm<sup>-1</sup>; MS (ESI) m/z 425.2 (100%, [M+H]<sup>+</sup>), HRMS (ESI) m/z found 425.1814,  $C_{22}H_{25}N_4O_5^+$  ([M+H]+) requires 425.1819.

#### *tert*-Butyl (S)-2-((N-

### ((benzyloxy)carbonyl)carbamimidoyl)carbamoyl)pyrrolidine-1-carboxylate 5b.

Method A: N-Boc-L-proline 9b (1.52 g, 7.04 mmol, 1.0 equiv.); DMF (10 mL), 0 °C; CDI (1.78 g, 10.1 mmol, 1.4 equiv.); 24 h; N-Cbz-guanidine 10a (1.35 g, 6.97 mmol, 1.0 equiv.); 2 d, rt. Extraction with EtOAc; washed with water (150 mL × 3) and brine (50 mL × 2); column chromatography (0-20% EtOAc in petroleum ether) gave 5b (1.10 g, 2.82 mmol) in 40% yield as a white solid.  $R_f 0.14 (30\% \text{ EtOAc in hexane}); Mp 64-66 °C; <math>[\alpha]_D^{20}$  -53.8 (c = 2.1, CHCl<sub>3</sub>); δ<sub>H</sub> (CDCl<sub>3</sub>) (mixture of rotamers) 7.63-10.22 (3H, br. s, 3 × NH), 7.27-7.38 (5H, m, Ph), 5.11 (2H, s, CH<sub>2</sub>), 4.11-4.46 (1H, m, CH), 3.28-3.62 (2H, m, CH<sub>2</sub>), 1.94-2.30 (2H, m, CH<sub>2</sub>), 1.76-1.94 (2H, m, CH<sub>2</sub>), 1.44/1.40 (9H, 2 × s, 3 × CH<sub>3</sub>);  $\delta_C$ (CDCl<sub>3</sub>) (mixture of rotamers) 175.9/176.7, 162.8, 158.8, 155.1/153.8, 136.5/136.3, 128.4, 128.0, 128.0, 80.9, 67.0, 61.8/61.4, 46.8/47.2, 31.3/29.7, 28.3, 24.5/23.8; *v*<sub>max</sub> 3380, 3275, 2976, 2883, 1697, 1661, 1628, 1541, 1497, 1477, 1446, 1392, 1367, 1279, 1161, 1120, 1090, 1045, 1026, 991, 927, 854, 806, 751, 698, 666 cm<sup>-1</sup>; **MS (ESI) m/z** 391.2 (100%, [M+H]<sup>+</sup>); **HRMS** (ESI) found 391.1980,  $C_{19}H_{27}N_4O_5^+$  ([M+H]+) requires 391.1976.

#### Di-tert-butyl 2,2'-

# (((iminomethylene)bis(azanediyl))bis(carbonyl))(2S,2'S)-bis(pyrrolidine-1-carboxylate) 6b

**Method B**: *N*-Boc-L-proline **9b** (0.95 g, 4.40 mmol, 2.4 equiv.); DMF (10 mL); CDI (0.82 g, 5.13 mmol, 2.8 equiv.); 24 h; guanidinium hydrochloride (0.175 g, 1.83 mmol, 1.0 equiv.); DMF (10 mL); NaH (60%, 77.0 mg, 1.92 mmol, 1.05 equiv.); 2 d. Extraction with EtOAc; washed with water (3 × 100 mL) and brine (2 × 100 mL); column chromatography (75% Et<sub>2</sub>O in hexane) gave **6b** (0.81 g, 1.78 mmol) as a white solid in 97% yield. **R**<sub>f</sub> 0.28 (75%) EtOAc in hexane); **Mp** 65-69 °C;  $[\alpha]_D^{20}$ -77.1 (CHCl<sub>3</sub>, c = 2.3);  $\delta_H$ (CDCl<sub>3</sub>) (mixture of rotamers) 9.11 (3H, br s, 3 × NH), 4.06-4.37 (2H, m, 2 × CH), 3.26-3.61 (4H, m, 2 × CH<sub>2</sub>), 1.74-2.29 (8H, m, 4  $\times$  CH<sub>2</sub>), 1.45/1.39/1.36 (18H, s, 6  $\times$  Me);  $\delta$ <sub>C</sub> (CDCl<sub>3</sub>) (mixture of rotamers) 158.4/158.4, 155.2/155.1/154.2, 80.3/80.1, 62.9/62.1/ 62.0, 47.1/46.8, 31.2, 29.8/29.7/29.6, 28.4/28.4, 24.4/23.7;  $\mathbf{v}_{\text{max}}$ 3366, 3231, 2976, 2932, 2879, 1692, 1643, 1520, 1477, 1451, 1392, 1365, 1249, 1158, 1122; MS (ESI) m/z 454.3 (100%)  $[M+H]^+$ ); HRMS (ESI) found 454.2654,  $C_{21}H_{36}N_5O_6^+$  ( $[M+H]^+$ ) requires 454.2660.

## *tert*-Butyl (*S*)-(1-(3-(*tert*-butyloxycarbonyl)guanidino)-1-oxopropan-2-yl)carbamate 7a.

Method A: N-Boc-L-alanine 11a (1.00 g, 5.29 mmol, 1.0 equiv.); DMF (15 mL), 0 °C; CDI (0.94 g, 6.66 mmol, 1.26 equiv.); 30 min; N-Boc-guanidine 10b (0.93 g, 5.95 mmol, 1.12 equiv.); DMF (15 mL); 3 d, rt; 2 d; 40  $^{\circ}$ C. Extraction with EtOAc (3 × 50 mL); washed with HCl (0.1 M, 50 mL), NaHCO<sub>3</sub> (aq. sat. 50 mL) and brine (2 × 50 mL); column chromatography (0-40% EtOAc in hexane) gave 7a (1.10 g, 3.33 mmol) in 63% as a white solid. Rf 0.11 (25% EtOAc in hexane);  $[\alpha]p^{23}$  -24.0 (CHCl<sub>3</sub>, c = 1.0); **Mp** 113 °C;  $\delta_H$  (CDCl<sub>3</sub>) (mixture of rotamers) 7.50-9.75 (3H, br s, 3 × NH), 5.56-5.81/5.21-5.48 (1H, 2 x br s, NH), 4.13-4.33/3.96-4.12 (1H, 2 x m, CH), 1.47 (9H, s, 3 × Me), 1.42 (9H, s, 3 × Me), 1.36 (3H, d, J 7.1 Hz, Me);  $\delta_C$  (CDCl<sub>3</sub>) (mixture of rotamers) 179.8, 164.2/158.9/156.6/155.2/152.0/149.7 (3 × C), 85.3/82.6, 79.9, 52.6, 28.5/28.2/28.1/28.0 (2 × tBu), 19.3; v<sub>max</sub> 3382, 3282, 2978, 2934, 1712, 1643, 1539, 1496, 1447, 1148, 1047 cm<sup>-1</sup>; MS (ESI) m/z 331.2 (100, [M+H]+), **HRMS (ESI)** found 331.1973, C<sub>14</sub>H<sub>27</sub>N<sub>4</sub>O<sub>5</sub>+  $([M+H]^+)$  requires 331.1976.

# *tert*-Butyl (*S*)-(1-(3-(benzyloxycarbonyl)guanidino)-1-oxopropan-2-yl)carbamate 7b.

**Method A**: *N*-Boc-L-alanine **11a** (1.00 g, 5.29 mmol, 1.0 equiv.); DMF (15 mL), 0 °C; CDI (0.94 g, 6.66 mmol, 1.26 equiv.); 30 min; N-Cbz-guanidine 10a (1.12 g, 5.81mmol, 1.1 equiv.); DMF (15 mL); 3 d, rt. Extraction with EtOAc (3 × 50 mL); washed with HCl (0.1 M, 50 mL), NaHCO<sub>3</sub> (aq. sat. 50 mL) and brine (2 × 50 mL); Recrystallized from ethanol (ca. 15 mL) at -20 °C (2 days) to form white needles which were washed with ice-cold diethyl ether and dried under vacuum to give 7b (1.37 g, 3.76 mmol) in 71% yield. Column chromatography (50% Et<sub>2</sub>O in hexane) gave an analytical sample.  $\mathbf{R}_{\rm f}$  0.11 (50% Et<sub>2</sub>O in hexane);  $[\alpha]p^{25}$  -18.8 (CHCl<sub>3</sub>, c = 1.15); Mp 138-140 °C;  $\delta_H$  (CDCl<sub>3</sub>) (mixture of rotamers) 7.77-10.41 (3H, br. s, 3 × NH), 7.24-7.37 (5H, m, Ph), 7.34/5.40 (1H, br s/d, J 5.8 Hz, NH), 5.09 (2H, s, CH<sub>2</sub>), 4.24-4.39/3.86-3.99 (1H, 2 × br m, CH), 1.41 (9H, s, 3 × Me), 1.34/1/24 (3H, d/obscured d, J 6.5 Hz, CH<sub>3</sub>); δ<sub>C</sub> (CDCl<sub>3</sub>) (mixture of rotamers) 177.9/176.7, 162.7/161.9, 159.2/159.1, 156.1/155.2, 136.1, 128.5, 128.2, 128.2, 81.5/80.3, 67.1, 52.7/51.7, 28.3, 18.3/17.2;  $v_{max}$  3383, 3282, 2978, 2928, 1695, 1629, 1542, 1498, 1436, 1165, 1069 cm<sup>-1</sup>; **MS (ESI) m/z** 365.2 (100%, [M+H]<sup>+</sup>); HRMS (ESI) found 365.1819, C<sub>17</sub>H<sub>25</sub>N<sub>4</sub>O<sub>5</sub><sup>+</sup> ([M+H]<sup>+</sup>) requires 365.1820.

## (S)-(1-(3-(tert-butyloxycarbonyl)guanidino)-1-oxopropan-2-yl)carbamate 7c.

**Method A**: *N*-Cbz-L-alanine **11b** (1.03 g, 4.48 mmol, 1.0 equiv.); DMF (10 mL), 0 °C; CDI (0.91 g, 5.60 mmol, 1.25 equiv.); 90 min; N-Boc-guanidine 10b (0.79 g, 4.93 mmol, 1.1 equiv.); 24 h, rt. Extraction with CHCl<sub>3</sub> (3 × 50 mL); washed with HCl (0.1 M, 50 mL), NaHCO<sub>3</sub> (aq. sat. 50 mL) and brine (2 × 50 mL); column chromatography (50% Et<sub>2</sub>O in hexane) gave 7c (1.37 g, 3.76 mmol) in 49% yield.  $R_f$  0.13 (50% Et<sub>2</sub>O in hexane);  $[\alpha]_D^{28}$  -21.2 (CHCl<sub>3</sub>, c = 1.0); **Mp** 64-66 °C;  $\delta_H$  (CDCl<sub>3</sub>) (mixture of rotamers) 8.03-9.41 (3H, br s,  $3 \times NH$ ), 7.16-7.49 (5H, m, Ph), 5.92/5.87 (1H, 2 × d, J 6.6/7.2 Hz, NH), 5.03-5.20 (2H, m, CH<sub>2</sub>), 4.25-4.33/4.07-4.14 (1H, 2 × m, CH), 1.50 (9H, s, 3 × Me), 1.41 (3H, d, J 7.0 Hz, Me);  $\delta_C$  (CDCl<sub>3</sub>) (mixture of rotamers) 179.8, 159.0/159.0, 156.7/156.2, 155.9/152.6, 136.4/136.6, 128.5/128.6, 128.2/128.1,  $128.1/128.0, 84.4, 66.6/66.9, 52.0/53.0, 27.9/28.0, 19.1/19.2; \nu_{max}$ 3374, 3101, 2976, 1720, 1639, 1524, 1500, 1236, 1146 cm<sup>-1</sup>; MS (ESI) m/z 365.2 (100, [M+H]+), HRMS (ESI) found 365.1819,  $C_{17}H_{25}N_4O_5^+$  ([M+H]<sup>+</sup>) requires 365.1819.

## Benzyl (*S*)-(1-(3-(benzyloxycarbonyl)guanidino)-1-oxopropan-2-yl)carbamate 7d.

Method A: *N*-Cbz-L-alanine 11b (1.00 g, 5.48 mmol, 1.0 equiv.); DMF (5 mL), 0  $^{0}$ C; CDI (0.88 g, 5.40 mmol, 1.20 equiv.); 3 h; *N*-Cbz-guanidine 10a (1.95 g, 4.93 mmol, 1.1 equiv.); DMF (5 mL); 3 d, rt. Extraction with CHCl<sub>3</sub> (3 × 50 mL); washed with HCl (aq. 0.1 M, 50 mL), NaHCO<sub>3</sub> (aq. sat. 50 mL) and brine (2 × 50 mL); recrystallized from Et<sub>2</sub>O (ca. 15 mL) at -20  $^{0}$ C (2 days) to form white needles which were washed with ice-cold diethyl ether and dried under vacuum to give 7d (1.34 g, 3.36 mmol) in 75% yield. Column chromatography (70% Et<sub>2</sub>O in hexane) gave an analytical sample. R<sub>f</sub> 0.19 (70% Et<sub>2</sub>O in hexane); [α]<sub>D</sub><sup>23</sup> -17.5 (CHCl<sub>3</sub>, c = 1.0); Mp 95-98  $^{\circ}$ C; δ<sub>H</sub> (CDCl<sub>3</sub>) (mixture of rotamers) 7.96-10.69 (3H, br s, 3 × NH), 7.19-7.58 (10H, m, 2 × Ph), 5.98/7.48 (1H, d/br s *J* 7.2 Hz, NH), 5.11 (2H, s, CH<sub>2</sub>), 5.05-5.14 (2H, AB pattern, *J* 12.5 Hz, CH<sub>2</sub>), 4.35-4.53/4.03-4.23 (1H, 2 × m, CH), 1.23/1.36 (3H, br d/d, *J* 5.4/7.0 Hz, Me); δ<sub>C</sub> (CDCl<sub>3</sub>) (mixture of rotamers)

178.4, 161.7/161.1, 159.1/158.8, 156.5/155.8, 136.2, 135.9/135.7, 128.5, 128.2, 128.2, 128.1, 127.9 (10 x CH) 67.5/67.2, 67.0, 52.5/52.0, 18.3/17.5;  $\nu_{max}$  3332, 3272, 3033, 2951, 1708, 1687, 1627, 1526, 1259 cm<sup>-1</sup>; **MS (ESI) m/z** 399.2 (100, [M+H]<sup>+</sup>), **HRMS (ESI)** found 399.1664, C<sub>20</sub>H<sub>23</sub>N<sub>4</sub>O<sub>5</sub><sup>+</sup> ([M+H]<sup>+</sup>) requires 399.16630.

# *tert*-Butyl (*S*)-(1-(3-(tert-butyloxycarbonyl)guanidino)-1-oxo-3-phenylpropan-2-yl)carbamate 7e.

Method A: N-Boc-L-alanine 11a (1.01 g, 3.77 mmol, 1.0 equiv.); DMF (10 mL), 0°C; CDI (0.77 g, 4.71 mmol, 1.25 equiv.); 90 min; N-Boc-guanidine 10b (0.67 g, 4.15 mmol, 1.1 equiv.); DMF (15 mL); 2 d, rt. Extraction with CHCl<sub>3</sub> (3 × 50 mL); washed with (3 × 50 mL) and brine (2 × 50 mL); column chromatography (45% Et<sub>2</sub>O in hexane) gave 7e (0.68 g, 1.67 mmol) in 44% yield as a white solid.  $\mathbf{R_f} \ 0.23 \ (50\% \ \text{Et}_2\text{O} \ \text{in hexane}); \ [\alpha]_{\mathbf{D}^{21}} \ -21.0 \ (\text{CHCl}_3, \ c = 1.0);$ Mp 64-66 °C;  $\delta_H$  (CDCl<sub>3</sub>) (mixture of rotamers) 9.26 (2H, br s, 2 × NH), 8.81 (H, br s, NH) 7.14-7.25 (3H, m, 3 × CH), 7.07 (2H, br d, J 6.9 Hz, 2 × CH) 5.71-5.91/5.35 (2H, m/d, J 5.8 Hz, NH), 4.49-4.64/4.20-4.40 (1H, m/br m, CH), 3.15/3.00/2.75-3.21 (2H, dd J 5.0, 13.6/dd, J 5.6, 13.6/br m, 2 x CH), 1.43 (9H, s, 3 × Me), 1.34 (9H, s,  $3 \times \text{Me}$ );  $\delta_C$  (CDCl<sub>3</sub>) (mixture of rotamers) 182.4, 159.2, 155.0, 136.6, 129.5, 128.3, 126.7, 83.2, 80.3/79.4, 59.2/57.6, 39.2/38.5, 28.4, 27.9;  $\nu_{\text{max}}$  3378, 3005, 2977, 2933, 1709, 1641, 1542, 1493, 1240, 1146 cm<sup>-1</sup>; **MS (ESI) m/z** 407.2 (100, [M+H]<sup>+</sup>); HRMS (ESI) found 407.2291, C<sub>20</sub>H<sub>31</sub>N<sub>4</sub>O<sub>5</sub><sup>+</sup> ([M+H]<sup>+</sup>), requires 407.2289.

# *tert*-butyl (*S*)-(1-(3-(benzyloxycarbonyl)guanidino)-1-oxo-3-phenylpropan-2-yl)carbamate 7f.

Method A: N-Boc-L-phenylalanine 11c (1.00 g, 3.77 mmol, 1.0 equiv.); DMF (15 mL), 0 °C; CDI (0.67 g, 4.75 mmol, 1.26 equiv.); 30 min; N-Cbz-guanidine **10a** (0.80 g, 4.15 mmol, 1.10 equiv.); DMF (15 mL); 3 d, rt, 2 d, 40 °C. Extraction with EtOAc  $(3 \times 50 \text{ mL})$ ; washed with HCl (0.1 M, 50 mL), NaHCO<sub>3</sub> (aq. sat. 50 mL) and brine  $(2 \times 50 \text{ mL})$ ; Recrystallized from ethanol (ca. 15 mL) at -20 °C (12 h) to form white needles which were washed with ice-cold Et<sub>2</sub>O and dried under vacuum to give 7f (1.20 g, 2.72 mmol) in 72% yield. Column chromatography (50% Et<sub>2</sub>O in hexane) gave an analytical sample. Rf 0.15 (50% Et<sub>2</sub>O in hexane);  $[\alpha]_D^{25}$  -20.8 (CHCl<sub>3</sub>, c = 1.27); **Mp** 151-154 °C;  $\delta_H$  (CDCl<sub>3</sub>) (mixture of rotamers) 8.89 (2H, br s, 2 × NH), 8.26 (1H, br s, NH), 7.07-7.44 (10H, m,  $10 \times CH$ ), 5.20 (1H, br s, NH), 5.12/5.18 (2H, 2  $\times$  s, CH<sub>2</sub>), 4.49-4.69/4.09-4.30 (1H, 2  $\times$  m, CH), 2.72-3.23 (2H, m, CH<sub>2</sub>) 1.39/1.30 (9H, 2 × s, 3 × Me);  $\delta_C$  (CDCl<sub>3</sub>) (mixture of rotamers) 178.5, 161.2/158.9/156.5/156.0/155.6/153.8 (3 × C), 137.5/136.5/136.2/136.0/134.5 (2 × C), 129.6/129.3/128.8/128.7/ 128.7/128.6/128.4/128.3/128.2/128.2/127.1/127.1 (10 × CH) 127.1/ 127.0/126.5, 81.4/80.4/79.4, 68.5/67.8/67.3, 58.6/56.9/56.2, 38.5/ 38.0/37.6, 28.4/28.3;  $\nu_{\text{max}}$  3395, 3350, 3271, 3059, 3033, 2985, 2956, 1708, 1684, 1620, 1543, 1511, 1439, 1378, 1316, 1294, 1252, 1201 cm<sup>-1</sup>; MS (ESI) m/z 441.2 (100, [M+H]<sup>+</sup>); HRMS (ESI) found 441.2134,  $C_{23}H_{29}N_4O_5^+$  ([M+H]+), requires 441.2132.

# Benzyl (S)-(1-(3-(tert-butyloxycarbonyl)guanidino)-1-oxo-3-phenylpropan-2-yl)carbamate 7g.

**Method A**: *N*-Cbz-L-phenylalanine **11d** (1.00 g, 3.34 mmol, 1.0 equiv.); DMF (15 mL); CDI (0.60 g, 3.74 mmol, 1.12 equiv.); 30 min; N-Boc-guanidine **10b** (0.59 g, 4.14 mmol, 1.24 equiv.); DMF (15 mL); 3 d, rt. Extraction with EtOAc; washed HCl (0.1 M, 50 mL), NaHCO<sub>3</sub> (aq. sat. 50 mL) and brine (2 × 50 mL); column chromatography (50% Et<sub>2</sub>O in hexane) gave **7g** (1.23 g, 2.79

mmol) in 84% yield as a white solid.  $\mathbf{R_f}$  0.23 (50% Et<sub>2</sub>O in hexane);  $[\alpha]_{\mathbf{p}^{21}}$  -32.2 (CHCl<sub>3</sub>, c = 1.54);  $\mathbf{Mp}$  123-125 °C;  $\delta_{\mathbf{H}}$  (CDCl<sub>3</sub>) (mixture of rotamers) 8.75 (3H, br s, 3 × NH), 7.02-7.40 (10H, m, 2 × Ph), 5.60-5.71/5.58 (1H, m/d, J 6.6 Hz, NH), 4.96-5.10 (2H, m, CH<sub>2</sub>), 4.57-4.65/4.37-4.51 (1H, 2 × m, CH), 2.92-3.22/3.12/3.26 (2H, m/dd/dd, J 13.6, 5.0/13.6, 5.2 Hz, 2 x CH) 1.47 (9H, s, 3 × Me);  $\delta_{\mathbf{C}}$  (CDCl<sub>3</sub>) (mixture of rotamers) 182.6, 159.0, 155.7, 154.3, 136.7, 136.5. 129.6/128.5/128.4/128.2/126.8 (10 x CH), 83.7, 66.8/67.0, 58.1/59.0, 38.4/39.5, 28.0;  $\mathbf{v}_{max}$  3378, 3031, 2977, 1721, 1641, 1542, 1497, 1145, 1081 cm<sup>-1</sup>;  $\mathbf{MS}$  (ESI) m/z 441.2 (100%, [M+H]<sup>+</sup>);  $\mathbf{HRMS}$  (ESI) found 441.2136, C<sub>23</sub>H<sub>29</sub>N<sub>4</sub>O<sub>5</sub><sup>+</sup> ([M+H]<sup>+</sup>), requires 441.2133.

# Benzyl (*S*)-(1-(3-(benzyloxycarbonyl)guanidino)-1-oxo-3-phenylpropan-2-yl)carbamate 7h.

Method A: N-Cbz-L-phenylalanine 11d (1.00 g, 3.34 mmol, 1.0 equiv.); DMF (15 mL), 0 °C; CDI (0.70 g, 4.98 mmol, 1.49 equiv.); 30 min; N-Cbz-guanidine 10a (0.70 g, 3.67 mmol, 1.10 equiv.); DMF (15 mL); 3 d, rt; 2 d, 40 °C. Extraction with EtOAc (3 × 50 mL); washed with HCl (0.1 M, 50 mL), NaHCO<sub>3</sub> (aq. sat. 50 mL) and brine (2 × 50 mL); column chromatography (45% Et<sub>2</sub>O in hexane) gave 7h (0.30 g, 0.63 mmol) in 19% as a white solid. Rf 0.35 (70% Et<sub>2</sub>O in hexane);  $[\alpha]_D^{25}$ -19.6° (CHCl<sub>3</sub>, c = 1.0); **Mp** 83-86 °C;  $\delta_H$  (CDCl<sub>3</sub>) (mixture of rotamers) 8.80 (3H, br s, 3 × NH), 6.85-7.31 (15H, m,  $3 \times Ph$ ), 7.33-7.50/5.48-5.59 (1H, br s/m, NH), 5.01 (2H, s, CH<sub>2</sub>), 4.74-5.07 (2H, m, 2 × CH), 4.62-4.74/4.16-4.32 (1H, 2 × m, CH), 2.62-3.11 (2H, m, CH<sub>2</sub>);  $\delta_C$  (CDCl<sub>3</sub>) (mixture of rotamers) 177.9, 160.2, 158.9/158.7, 156.7/155.9, 136.6, 136.2/136.0, 135.8/135.6, 129.3, 128.6, 128.6, 128.5, 128.3, 128.2, 128.2, 128.1, 127.9, 127.1, 127.0 (15 × CH), 67.4, 67.1, 58.4/57.3, 38.1/38.0/37.7;  $\nu_{max}$  3390, 3338, 3278, 3063, 3031, 2962, 1687, 1664, 1630, 1523, 1497, 1268, 1111, 1087; MS (ESI) m/z 475.2 (100,  $[M+H]^+$ ); **HRMS** (ESI)  $C_{26}H_{27}N_4O_5^+$   $[M+H]^+$ ), requires 475.1976, found 475.1977.

# Di-*tert*-butyl ((2*S*,2'*S*)-((iminomethylene)bis(azanediyl))bis(1-oxopropane-1,2-diyl))dicarbamate 8a.

Method B: *N*-Boc-L-alanine 11a (1.78 g, 9.4 mmol, 2.2 equiv); DMF (5 mL); CDI (2.04 g, 12.7 mmol, 3.0 equiv); 0 °C, 90 min; guanidine hydrochloride (0.41 g, 4.3 mmol, 1.0 equiv); DMF (5 mL); NaH (60%, 0.16 g, 4.0 mmol, 0.93 equiv.), 24 h; 5 d; freeze dried then column chromatography (60% Et<sub>2</sub>O in hexane) gave 8a (1.05 g, 2.6 mmol) in 65% yield as a white solid. R<sub>f</sub> 0.30 (75% Et<sub>2</sub>O/PE); [α]<sub>0</sub>2<sup>1</sup> -30 (CHCl<sub>3</sub>, c = 1.0); Mp 76-78 °C; δ<sub>H</sub> (CDCl<sub>3</sub>) (mixture of rotamers) 9.18 (3H, br s, 3 × NH), 5.40-5.56/5.30-5.40 (2H, 2 × m, 2 × NH), 4.19-4.36/3.96-4.19 (2H, 2 × m, 2 × CH), 1.43 (18H, s, 6 × Me), 1.38 (6H, d, *J* 6.8 Hz, 2 × Me); δ<sub>C</sub> (CDCl<sub>3</sub>) 180.1, 158.7, 155.6, 80.3, 52.3, 28.4, 18.6; ν<sub>max</sub> 3245, 3219, 3001, 2977, 2932, 1690, 1644, 1603, 1509, 1247; cm<sup>-1</sup>; MS (ESI) m/z 402.2 (100, [M+H]<sup>+</sup>); HRMS (ESI) found 402.2348, C<sub>17</sub>H<sub>32</sub>N<sub>5</sub>O<sub>6</sub><sup>+</sup> ([M+H]<sup>+</sup>) requires 402.2347.

# $\label{linear_prop} Dibenzyl\ ((2S,2'S)-((iminomethylene)bis(azanediyl))bis(1-oxopropane-1,2-diyl))dicarbamate\ 8b.$

Method B: *N*-Cbz-L-alanine 11b (0.93 g, 4.16 mmol, 2.2 equiv.); DMF (5 mL); CDI (0.92 g, 5.7 mmol, 3.0 equiv.); 2 h; guanidine hydrochloride (0.18 g, 1.88 mmol, 1.0 equiv.); DMF (5 mL); NaH (60%, 0.07 g, 1.75 mmol, 0.93 equiv.); 24 h; 2 d; column chromatography (90-100% Et<sub>2</sub>O in hexane) gave 8b (0.71 g, 1.51 mmol) in 80% yield as a white solid.  $\mathbf{R}_{\mathbf{f}}$  0.33 (Et<sub>2</sub>O); [α] $\mathbf{p}^{27}$  -22.7 (CHCl<sub>3</sub>, c = 1.0); **Mp** 90-93 °C; δ<sub>H</sub> (CDCl<sub>3</sub>) (mixture of rotamers), 7.83 (1H, br s, 3 × NH), 7.27-7.39 (10H, m, 2 × Ph), 5.57-5.72

(2H, br m,  $2 \times NH$ ), 5.13 (2H, d, J 12.3 Hz,  $2 \times CH$ ), 5.08 (2H, d, J 12.3 Hz,  $2 \times CH$ ), 4.29-4.41/4.14-4.29 (2H, m/m,  $2 \times CH$ ), 1.42/1.40 (6H,  $2 \times d$ , J 7.0 Hz,  $2 \times Me$ );  $\delta_C$  (CDCl<sub>3</sub>) 176.0, 158.6, 156.0, 136.2, 128.7, 128.4, 128.3, 67.3, 52.8, 18.6;  $\nu_{max}$  3338, 3031, 2977, 1693, 1643, 1605, 1508, 1213 cm<sup>-1</sup>; **MS** (**ESI**) m/z 470.2 (100, [M+H]<sup>+</sup>), **HRMS** (**ESI**) found 470.2040, C<sub>23</sub>H<sub>28</sub>N<sub>5</sub>O<sub>6</sub><sup>+</sup> ([M+H]<sup>+</sup>) requires 470.2034.

## Di-tert-butyl ((2S,2'S)-((iminomethylene)bis(azanediyl))bis(1-oxo-3-phenylpropane-1,2-diyl))dicarbamate 8c.

**Method B**: *N*-Boc-L-phenylalanine **11c** (1.10 g, 4.15 mmol, 2.2 equiv); DMF (5 mL); CDI (0.92 g, 5.66 mmol, 3.0 equiv); 2 h; guanidine hydrochloride (0.19 g, 1.99 mmol, 1.0 equiv); DMF (5 mL); NaH (60%, 0.07 g, 1.75 mmol, 0.9 equiv.); 2 d; column chromatography (35% Et<sub>2</sub>O in hexane) gave **8c** (0.78 g, 1.40 mmol) in 80% yield as a white solid. **R**<sub>f</sub> 0.48 (70% Et<sub>2</sub>O in hexane); [α]<sub>D</sub><sup>27</sup> -21.4 (CHCl<sub>3</sub>, c = 1.0); **Mp** 83-86 °C; δ<sub>H</sub> (mixture of rotamers) 9.71 (3H, br s, 3 × NH), 7.09-7.33 (10H, m, 2 × Ph), 5.46-5.74/5.23-5.46 (2H, 2 × m, 2 × NH), 4.40-4.62/4.19-4.40 (2H, 2 × m, 2 × CH), 2.95-3.24 (4H, m, 2 × CH<sub>2</sub>), 1.38 (18H, s, 6 × Me); δ<sub>C</sub> (mixture of rotamers) 178.6, 158.3, 155.6, 136.4, 129.4, 128.6, 127.0, 80.6/80.3, 59.3/57.4, 38.7/38.1, 28.4; ν<sub>max</sub> 3368, 3008, 2977, 2932, 1689, 1645, 1604, 1496 cm<sup>-1</sup>; **MS** (**ESI**) m/z 554.3 (100, [M+H]<sup>+</sup>), **HRMS** (**ESI**) found 554.2982, C<sub>29</sub>H<sub>40</sub>N<sub>5</sub>O<sub>6</sub><sup>+</sup> ([M+H]<sup>+</sup>) requires 554.2973.

# Dibenzyl ((2S,2'S)-((iminomethylene)bis(azanediyl))bis(1-oxo-3-phenylpropane-1,2-diyl))dicarbamate 8d.

Method B: N-Cbz-L-phenylalanine 11d (1.24 g, 4.15 mmol, 2.2 equiv); DMF (5 mL); CDI (0.92 g, 5.70 mmol, 3.0 equiv); 0 °C, 90 min; guanidine hydrochloride (0.18 g, 1.88 mmol, 1.0 equiv); DMF (5 mL); NaH (60%, 73.0 mg, 1.83 mmol, 0.97 equiv.), 24 min; 4 d; column chromatography (50% Et<sub>2</sub>O in hexane) gave **8d** (0.55 g, 0.89 mmol) in 49% yield as a white solid.  $R_f$  0.26 (70% Et<sub>2</sub>O in hexane);  $[\alpha]_D^{28}$  -12.8 (CHCl<sub>3</sub>, c = 1.0); **Mp** 150-152 °C;  $\delta_H$ (CDCl<sub>3</sub>) (mixture of rotamers) 9.07 (3H, br s, 3 × NH), 6.76-7.38 (20H, m, 4 × Ph), 5.52-5.87 (2H, m, 2 × NH), 4.91-5.14 (4H, m, 2  $\times$  CH<sub>2</sub>), 4.52-4.91/4.29-4.50 (2H, 2  $\times$  m, 2  $\times$  CH), 2.75-3.20 (4H, m, 2 × CH<sub>2</sub>);  $\delta_C$  (mixture of rotamers) 178.4, 158.4/158.3, 156.2/156.1, 136.2, 136.0, 129.5, 129.4, 128.7, 128.6, 128.6, 128.3 128.2, 127.1, 67.3/66.8, 59.0/57.8, 39.3/38.1;  $\nu_{\text{max}}$  3351, 3063, 3030, 2952, 1689, 1644, 1604, 1496, 1212 cm<sup>-1</sup>; MS (ESI) m/z 622.3 (100, [M+H]<sup>+</sup>), **HRMS (ESI)** found 622.2660, C<sub>35</sub>H<sub>36</sub>N<sub>5</sub>O<sub>6</sub><sup>+</sup>  $([M+H]^+)$  requires 622.2667.

## Catalysed reaction of 2-hydroxy-1,4-napthoquinone 2 with $\beta$ -nitrostyrene 3

2-Hydroxy-1,4-napthoquinone **2** (100 mg, 0.574 mmol) and the required catalyst (0.04-0.1 equiv.) were dissolved in the requisite solvent and cooled to the required temperature (-20 to 0 °C).  $\beta$ -Nitrostyrene **3** (128.5 mg, 0.861 mmol, 1.5 equiv) was then added and the mixture stirred for the required time and temperature. Reaction progress was determined by sampling and determination by <sup>1</sup>H NMR. On completion the solvent was evaporated to give a deep red residue which was purified by column chromatography (2-4% EtOAc in petroleum ether to remove excess **3** then CH<sub>2</sub>Cl<sub>2</sub>) to give **4** as a yellow solid. Enantiomeric excesses were determined either on a Chiralcel AS-H column (250 × 4.6 mm, mobile phase 96% hexane, 4% isopropanol, 0.1% TFA, 1.5 mL/min at 40 °C, detecting at 254 nm; R enantiomer 23.5 min, S enantiomer 26.2 min); or on a Phenomenex Lux Amylose-1 column (250 × 4.6 mm, mobile phase 70% hexane, 30% isopropanol, 0.5 mL/min at 40 °C,

detecting at 254 nm; R enantiomer 13.2 min, S enantiomer 14.3 min).<sup>1</sup>

† Note added in proof. A communication<sup>10</sup> detailing the use of amino acylguanidines as bioinspired catalysts for the asymmetric aldol reaction, reports on a threonine derived amino acylguanidine organocatalyst. Catalysis of the asymmetric aldol addition of hydroxyacetone with this compound, afforded predominantly syn diastereselectivity and high ee. MMFF modelling suggested the presence of extensive hydrogen bonding network between the acylguanidinium group and the reaction intermediates.

#### References

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### Appendix 1: Slow catalysts – insights into structural dependency of amino acid derived Guanidine catalysts (P. J. Murphy et al).

### **Supporting information**

### X-Ray data

Table S1. Crystal data and structure refinement details for 7b.

Empirical formula	$C_{17}H_{24}N_4O_5$				
Formula weight	364.40				
Temperature	100(2) K				
Wavelength	1.54184 Å				
Crystal system	Monoclinic				
Space group	$P2_1$				
Unit cell dimensions	a = 9.3046(3)  Å	$\alpha = 90^{\circ}$			
	b = 20.5570(6)  Å	$\beta = 109.461(3)^{\circ}$			
	c = 10.3928(3)  Å	$\gamma = 90^{\circ}$			
Volume	$1874.31(10) \text{ Å}^3$				
Z	4				
Density (calculated)	$1.291 \text{ Mg} / \text{m}^3$				
Absorption coefficient	$0.802~{\rm mm}^{-1}$				
F(000)	776				
Crystal	plate; colourless				
Crystal size	$0.1 \times 0.05 \times 0.025 \text{ mm}^3$				
$\theta$ range for data collection	$4.301 - 70.508^{\circ}$				
Index ranges	$-11 \le h \le 11, -25 \le k \le 24, -12 \le l \le 12$				
Reflections collected	34589				
Independent reflections	$7018 [R_{int} = 0.1316]$				
Completeness to $\theta = 67.684^{\circ}$	100.0 %				
Absorption correction	Semi-empirical from equivalents				
Max. and min. transmission	1.00000 and 0.48398				
Refinement method	Full-matrix least-squares on $F^2$				
Data / restraints / parameters	7018 / 13 / 509				
Goodness-of-fit on $F^2$	1.053				
Final <i>R</i> indices $[F^2 > 2\sigma(F^2)]$	R1 = 0.0746, wR2 = 0.1866				
R indices (all data)	R1 = 0.0874, wR2 = 0.1971				
Absolute structure parameter	-0.03(19)				
Extinction coefficient	0.0036(8)				
Largest diff. peak and hole	$0.735 \text{ and } -0.303 \text{ e Å}^{-3}$				
CCDC Deposition Number	CCDC2030901				

**Table S2.** Crystal data and structure refinement details for **7f**.

Empirical formula	$C_{23}H_{28}N_4O_5$
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Formula weight 440.49
Temperature 100(2) K
Wavelength 1.54184 Å
Crystal system Triclinic

Space group P1

Unit cell dimensions a = 5.21590(10) Å  $\alpha = 76.385(2)^{\circ}$ 

b = 12.9476(2) Å  $\beta = 83.401(2)^{\circ}$ c = 17.1996(4) Å  $\gamma = 88.849(2)^{\circ}$ 

Volume 1121.40(4) Å<sup>3</sup>

Z 2

Density (calculated)  $1.305 \text{ Mg} / \text{m}^3$ Absorption coefficient  $0.767 \text{ mm}^{-1}$ 

F(000) 468

Crystal plate; colourless

Crystal size  $0.3 \times 0.05 \times 0.01 \text{ mm}^3$ 

 $\theta$  range for data collection  $3.512 - 70.456^{\circ}$ 

Index ranges  $-5 \le h \le 6, -15 \le k \le 15, -20 \le l \le 20$ 

Reflections collected 52890

Independent reflections 7847 [ $R_{int} = 0.0926$ ]

Completeness to  $\theta = 67.684^{\circ}$  99.0 %

Absorption correction Semi-empirical from equivalents

Max. and min. transmission 1.00000 and 0.31940

Refinement method Full-matrix least-squares on  $F^2$ 

Data / restraints / parameters 7847 / 3 / 609

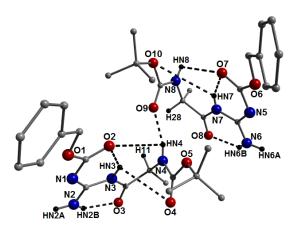
Goodness-of-fit on  $F^2$  1.071

Final *R* indices  $[F^2 > 2\sigma(F^2)]$  RI = 0.0572, wR2 = 0.1564*R* indices (all data) RI = 0.0606, wR2 = 0.1592

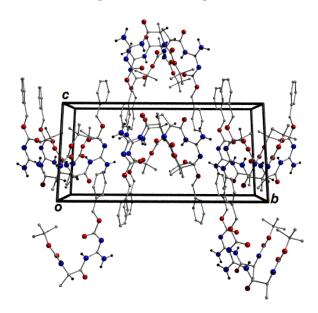
Absolute structure parameter 0.0(3)

Largest diff. peak and hole 0.252 and -0.235 e Å<sup>-3</sup>

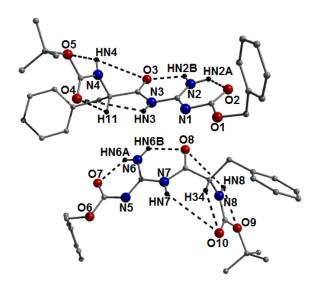
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**Figure S1** The two guanidine molecules comprising the asymmetric unit in **7b**. The dashes lines represent hydrogen bonding intra- and intermolecular interactions. Hydrogen bond distances: N2(HN2B)···O3 = 1.978 Å; N3(HN3)···O2 = 1.973 Å; N7(HN7)...O7 = 1.867 Å; N6(HN6B)···O8 = 2.035 Å; N4(HN4)...O9 = 1.953 Å along with the long-contacts N3(HN3)···O4 = 3.041 Å and N4(HN4)···O2 = 3.402 Å; N7(HN7)...O10 = 3.538 Å and N8(HN8)···O7 = 3.490 Å. The labels H11 and H28 represent the chiral protons in **7b**.



**Figure S2** Packing configuration observed in the crystal structure of **7b** as viewed along the *a* unit cell direction. All aromatic and *tert*-Butyl protons have been omitted for clarity.



**Figure S3** The two guanidine molecules that make up the asymmetric unit in **7f**. The dashes lines represent the H-bonds: N2(HN2A)···O2 = 1.97 Å; N2(HN2B)···O3 = 2.17 Å; N4(HN4)···O5 = 2.27 Å; C11(H11)···O4 = 2.42 Å; N6(HN6A)···O7 = 2.01 Å; N6(HN6B)···O8 = 2.13 Å; N8(HN8)···O9 = 2.36 Å; C34(H34)···O10 = 2.41 Å along with the long contacts with distances of 3.49 Å (N3(HN3)···O4); 3.12 Å (N4(HN4)···O3); 3.41 Å ((N7(HN7)···O10) and 3.02 Å (N8(HN8)···O8). The chiral protons have been labelled H11 and H34.

### Appendix 2: Carbamate protected amino acid derived guanidine organocatalysts.

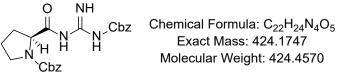
(P. J. Murphy et al).

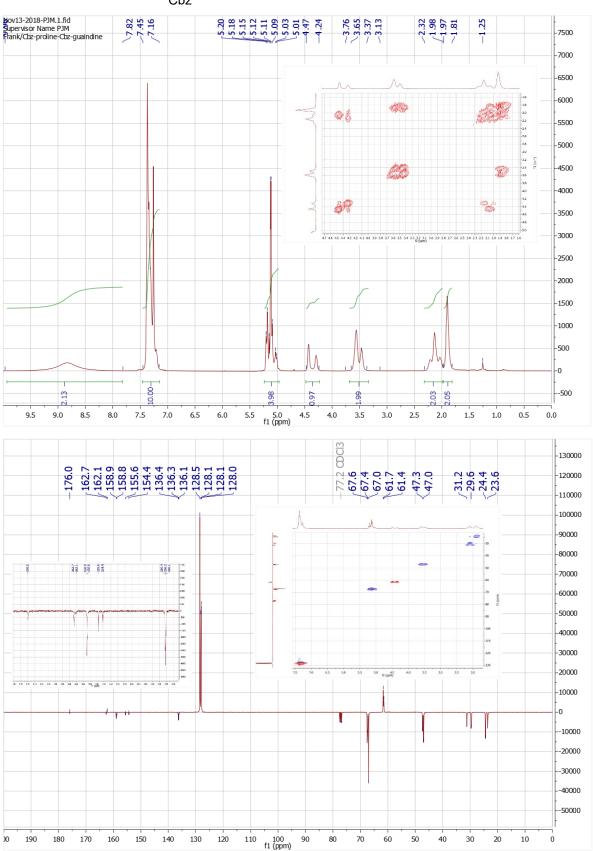
### **Supporting information**

### **NMR Spectra**

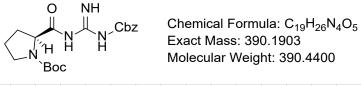
### **Cbz- and Boc-L-proline catalysts**

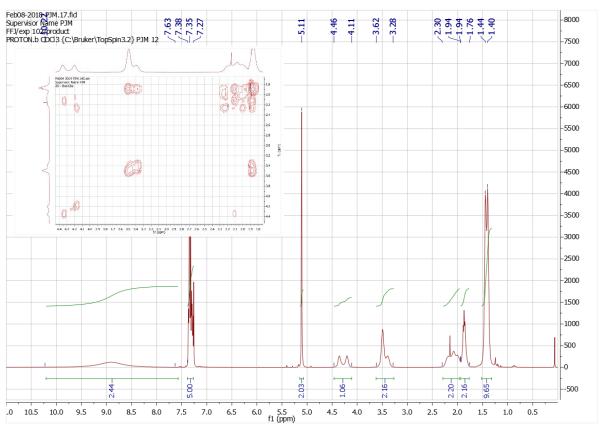
Benzyl (S)-2-((N-((benzyloxy)carbonyl)carbamimidoyl)carbamoyl)pyrrolidine-1-Carboxylate 5a.

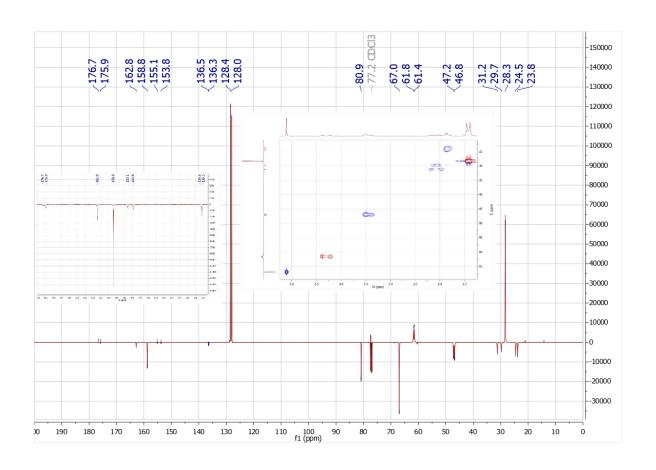




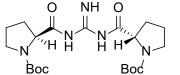
# tert-Butyl (S)-2-((N-((benzyloxy)carbonyl)carbamimidoyl)carbamoyl)pyrrolidine-1-carboxylate 5b.







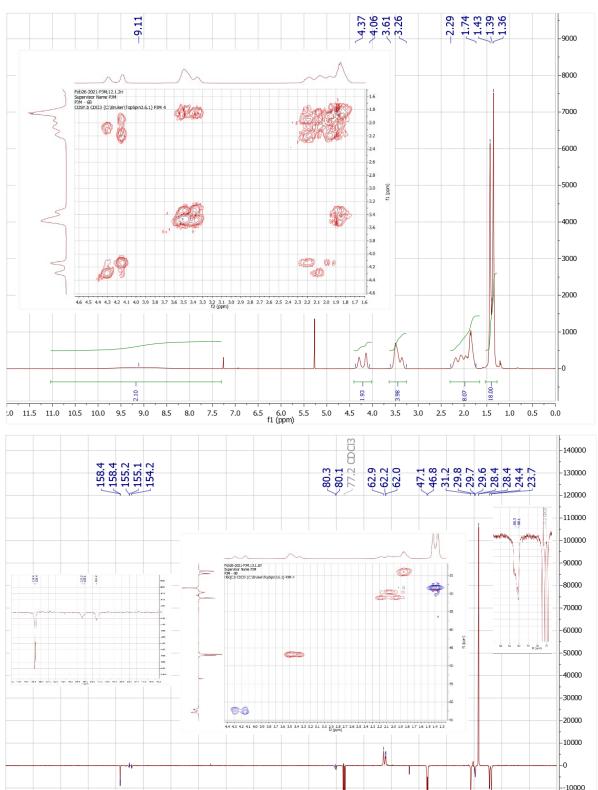
# Di-tert-butyl 2,2'-(((iminomethylene)bis(azanediyl))bis(carbonyl))(2S,2'S)-bis(pyrrolidine- 1-carboxylate) 6b.



Chemical Formula:  $C_{21}H_{35}N_5O_6$ 

-20000

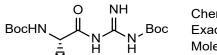
Exact Mass: 453.2587 Molecular Weight: 453.5400



100 f1 (ppm)

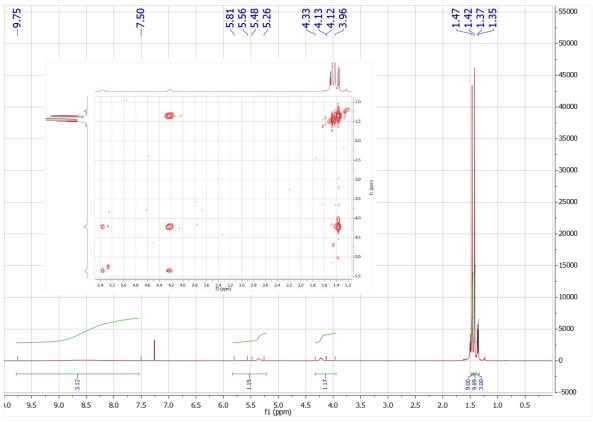
### N-Cbz-, N-Boc-L-alanine and N-Cbz-, N-Boc-L-phenylalanine catalysts.

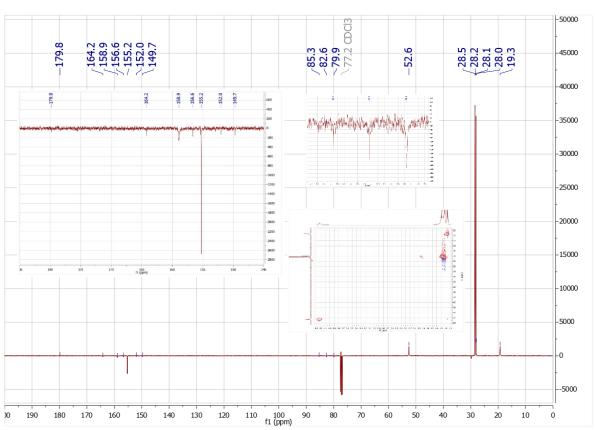
### tert-Butyl (S)-(1-(3-(tert-butyloxycarbonyl)guanidino)-1-oxopropan-2-yl)carbamate 7a.



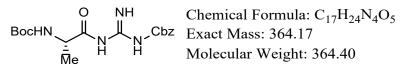
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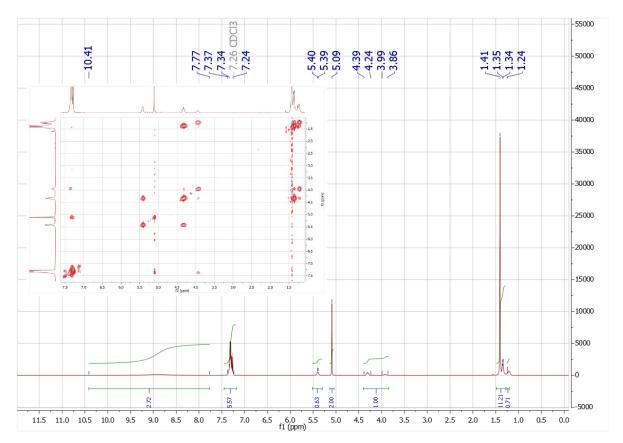
Exact Mass: 330.19 Molecular Weight: 330.39

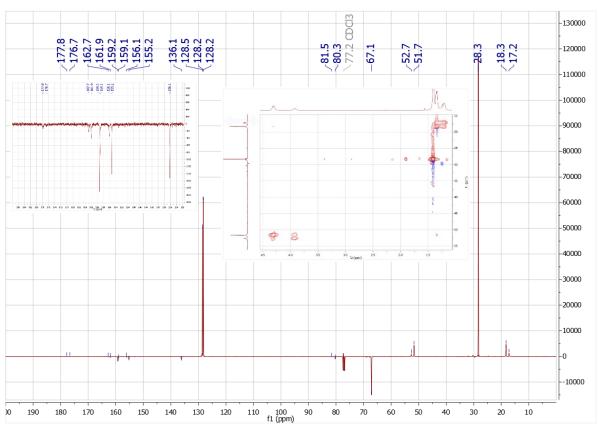




### tert-Butyl~(S)-(1-(3-(benzyloxycarbonyl)guanidino)-1-oxopropan-2-yl)carbamate~7b.

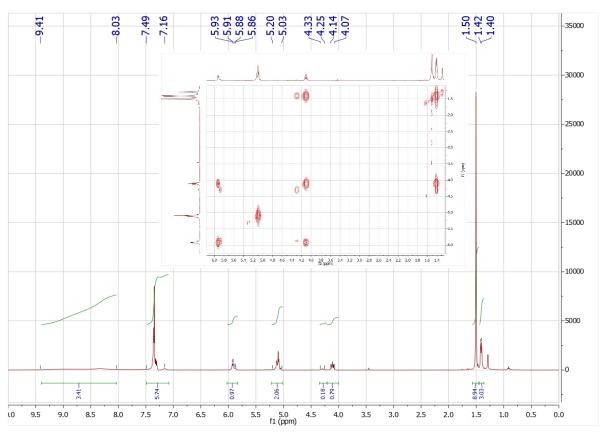


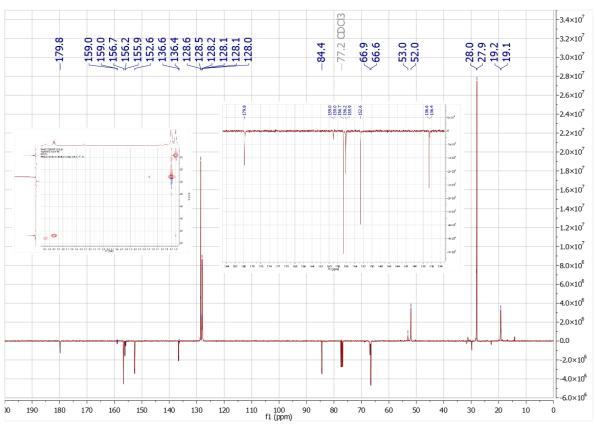




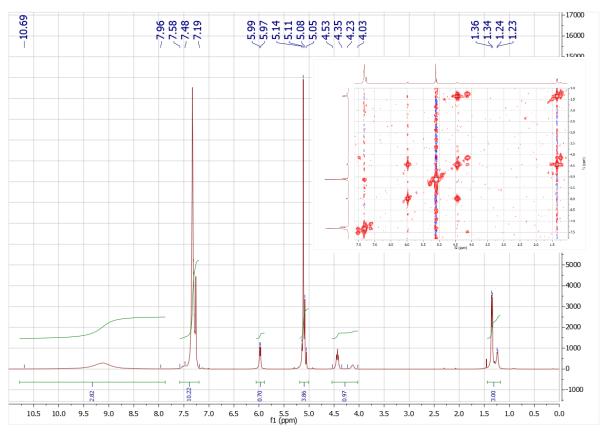
### Benzyl (S)-(1-(3-(tert-butyloxycarbonyl)guanidino)-1-oxopropan-2-yl)carbamate 7c.

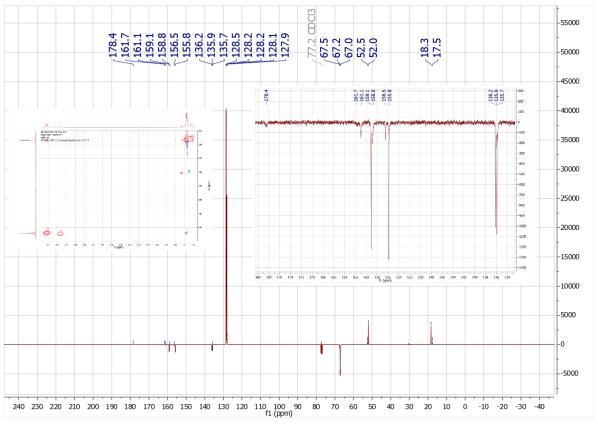
 $\begin{array}{c|cccc} \text{CbzHN} & \begin{array}{c} \text{O} & \text{NH} & \text{Chemical Formula: } C_{17}H_{24}N_4O_5 \\ \hline \text{N} & \text{H} & \text{H} & \text{Exact Mass: } 364.17 \\ \hline \text{Molecular Weight: } 364.40 \\ \end{array}$ 



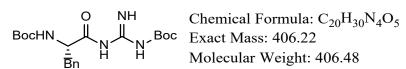


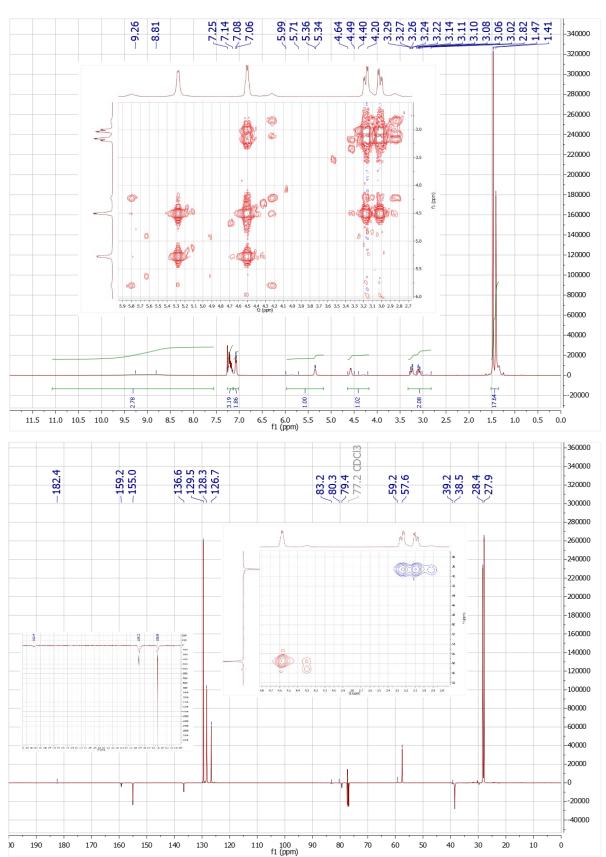
### Benzyl (S)-(1-(3-(benzyloxycarbonyl)guanidino)-1-oxopropan-2-yl)carbamate 7d.



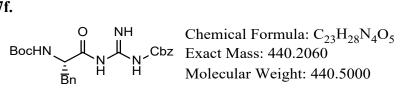


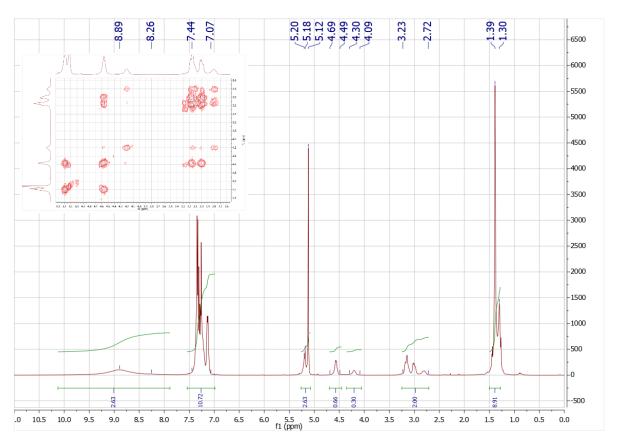
# *tert*-Butyl (S)-(1-(3-(tert-butyloxycarbonyl)guanidino)-1-oxo-3-phenylpropan-2-yl)carbamate 7e.

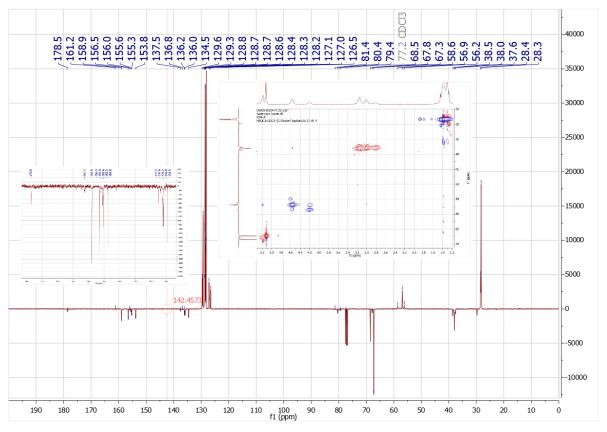




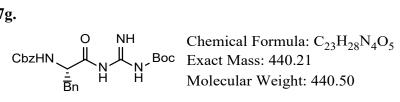
# *tert*-butyl (S)-(1-(3-(benzyloxycarbonyl)guanidino)-1-oxo-3-phenylpropan-2-yl)carbamate 7f.

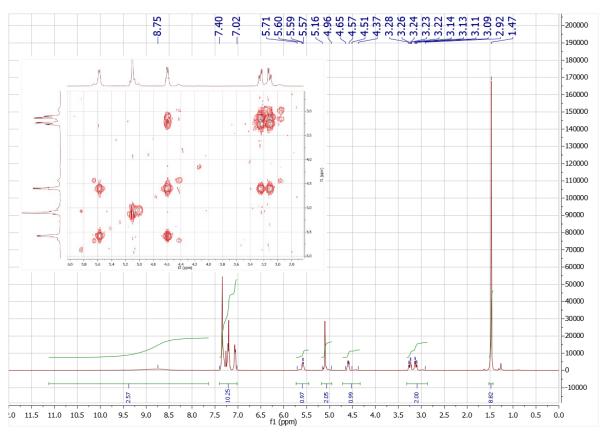


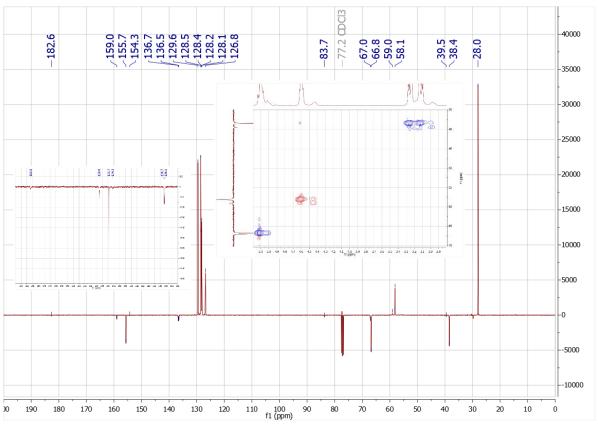




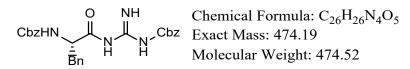
# Benzyl (S)-(1-(3-(tert-butyloxycarbonyl)guanidino)-1-oxo-3-phenylpropan-2-yl)carbamate 7g.

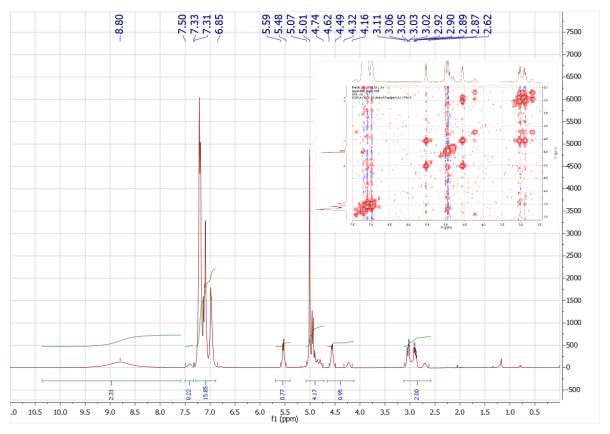


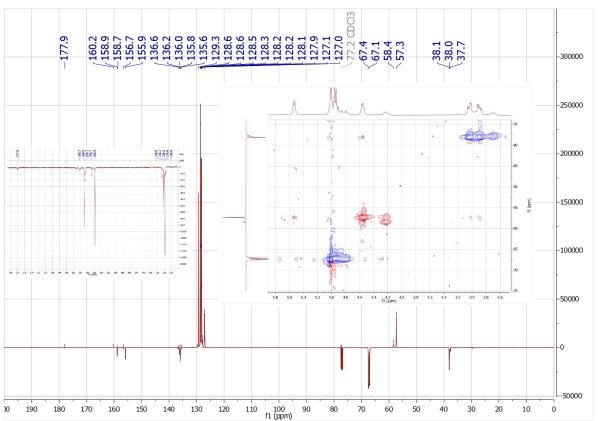




Benzyl (S)-(1-(3-(benzyloxycarbonyl)guanidino)-1-oxo-3-phenylpropan-2-yl)carbamate 7h.

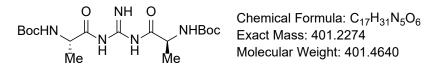


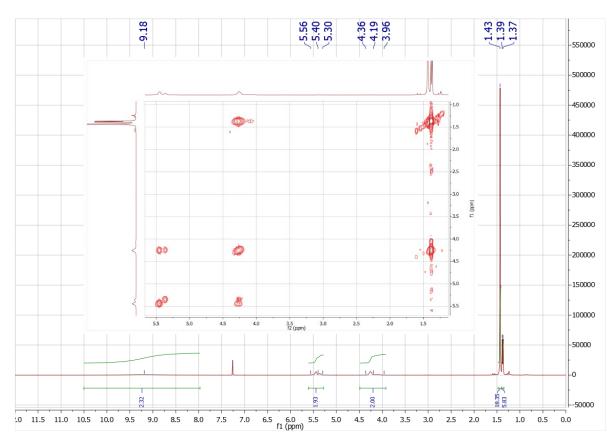


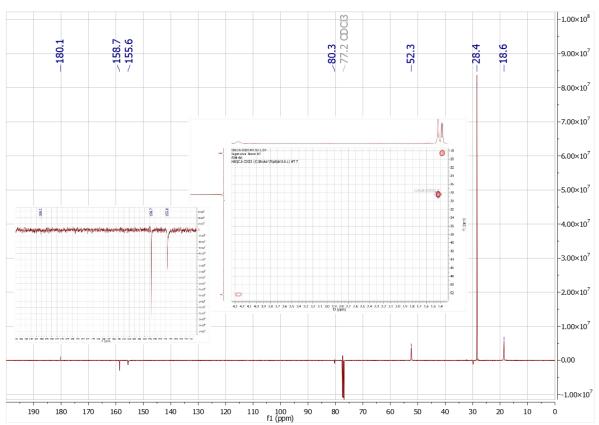


### N-Cbz- and N-Boc-L-alanine and L-phenylalanine C2-symmetric catalysts.

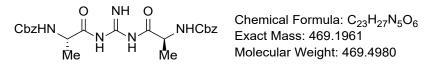
# Di-tert-butyl ((2S,2'S)-((iminomethylene)bis(azanediyl))bis(1-oxopropane-1,2-diyl))dicarbamate 8a.

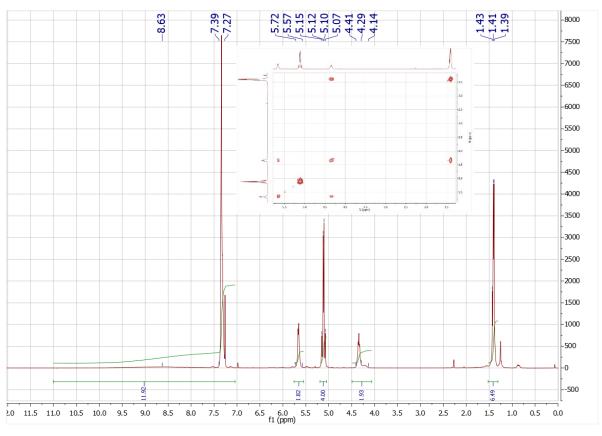


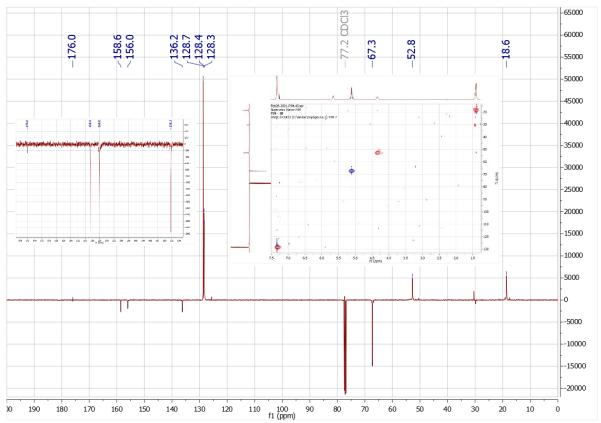




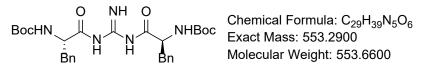
# Dibenzyl ((2S,2'S)-((iminomethylene)bis(azanediyl))bis(1-oxopropane-1,2-diyl))dicarbamate 8b.

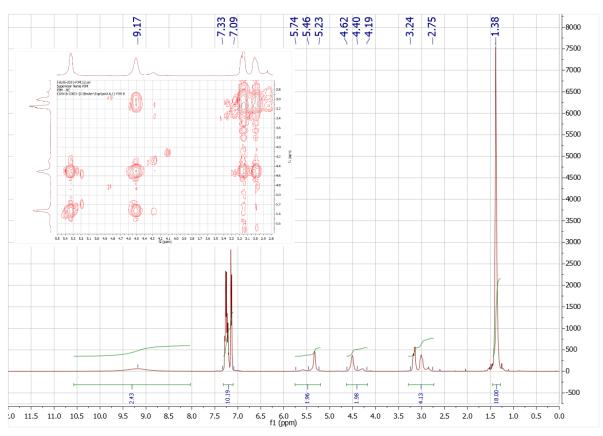


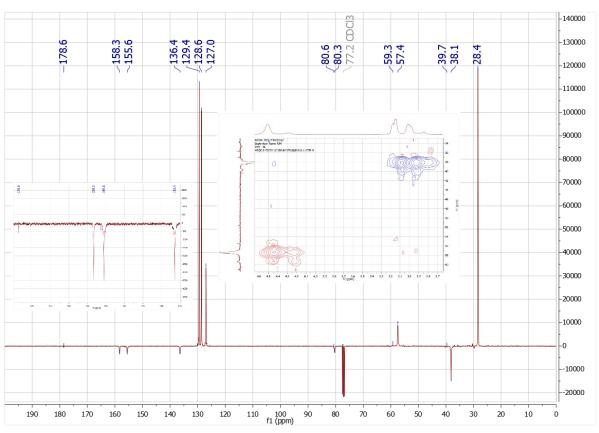




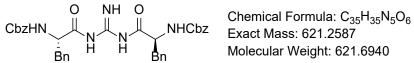
# Di-tert-butyl ((2S,2'S)-((iminomethylene)bis(azanediyl))bis(1-oxo-3-phenylpropane-1,2-diyl))dicarbamate 8c.

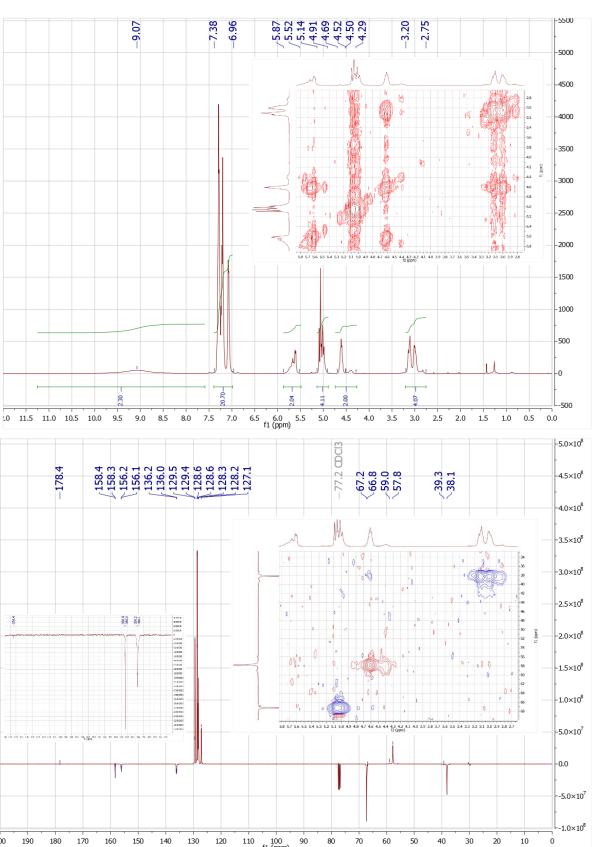




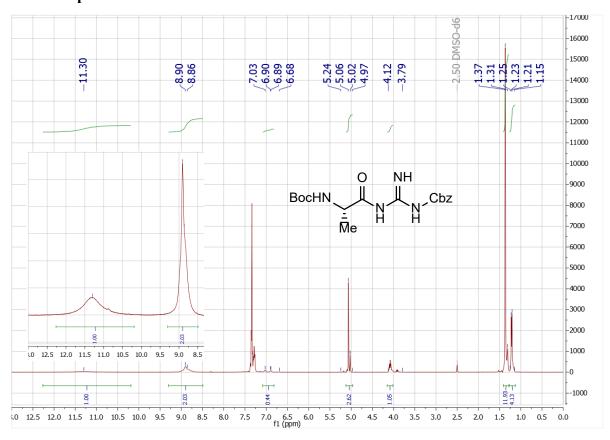


# Dibenzyl ((2S,2'S)-((iminomethylene)bis(azanediyl))bis(1-oxo-3-phenylpropane-1,2-diyl))dicarbamate 8d.





### <sup>1</sup>H NMR spectrum of 7b in DMSO-d6.



### <sup>1</sup>H NMR spectrum of 7f in DMSO-d6.

